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## Q5 Critical review of EPS production, synthesis and composition 2 for sludge flocculation

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### ABSTRACT

Extracellular polymeric substances (EPS) produced by microorganisms represent biological 16  
 macromolecules with unfathomable potentials and they are required to be explored further 17  
 for their potential application as a bioflocculant in various wastewater sludge treatment. 18  
 Although several studies already exist on biosynthetic pathways of different classical 19  
 biopolymers like alginate and xanthan, no dedicated studies are available for EPS in sludge. 20  
 This review highlights the EPS composition, functionality, and biodegradability for its 21  
 potential use as a carbon source for production of other metabolites. Furthermore, the effect 22  
 of various extraction methods (physical and chemical) on compositional, structural, 23  
 physical and functional properties of microbial EPS has been addressed. The vital 24  
 knowledge of the effect of extraction method on various important attributes of EPS can 25  
 help to choose the suitable extraction method depending upon the intended use of EPS. The 26  
 possible use of different molecular biological techniques for enhanced production of 27  
 desired EPS was summarized. 28

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### 42 Contents

43	Introduction . . . . .	0
44	1. Composition of EPS . . . . .	0
45	1.1. Polysaccharides (carbohydrates) . . . . .	0
46	1.2. Protein . . . . .	0
47	1.3. DNA and humic substances . . . . .	0
48	2. EPS biosynthesis . . . . .	0
49	2.1. Synthesis of precursor substrate . . . . .	0
50	2.2. Polymerization and cytoplasmic membrane transfer . . . . .	0
51	2.3. Export through the outer membrane . . . . .	0

Abbreviations: C/N, carbon to nitrogen molar ratio; CER, cation exchange resin; EDTA, Ethylene di amine tetra acetate group; EPS, exopolysaccharide or extra cellular polymeric substances; FTIR, Fourier Transform Infrared spectroscopy technique; GDP, Guanosine di phosphate; GT, Glucosyltransferase; MBR, membrane bioreactor; SEC, size exclusion chromatography; SS, suspended solids; UDP, uridine diphosphate; VSS, volatile suspended solids; WWTP, waste water treatment plant or wastewater treatment process.

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52	3. Engineering strategies for bacterial polysaccharides biosynthesis . . . . .	0
53	4. EPS production methodology . . . . .	0
54	4.1. EPS extracted directly from sludge . . . . .	0
55	4.2. EPS production using pure carbon sources . . . . .	0
56	4.3. Use of sludge as nutrient source for EPS production . . . . .	0
57	5. EPS extraction . . . . .	0
58	5.1. Physical methods . . . . .	0
59	5.2. Chemical methods . . . . .	0
60	5.3. Chemical methods vs physical methods . . . . .	0
61	5.4. Combination of different methods . . . . .	0
62	5.5. Effect of extraction methods on functional group and molecular weight of EPS . . . . .	0
63	6. Potential applications of EPS . . . . .	0
64	6.1. EPS as adsorbent . . . . .	0
65	6.2. EPS as carbon source . . . . .	0
66	6.3. Effect of functional group and molecular weight on flocculation activity . . . . .	0
67	7. Conclusion and recommendations . . . . .	0
68	References . . . . .	0

69

## 70 Introduction

72 In general, sludge settling is improved by the addition of  
 73 synthetic polymers, but they are known to be expensive and  
 74 may further pollute the environment (Deng et al., 2003). To  
 75 minimize the use of synthetic flocculants in sludge settling  
 76 applications, a novel alternative approach will be to use eco-  
 77 friendly bio coagulants/bioflocculants. The role of extracellular  
 78 polymeric substances (EPSs) produced by sludge microorgan-  
 79 isms during the wastewater treatment process have been  
 80 extensively studied (Hay et al., 2010; More et al., 2014;  
 81 Subramanian et al., 2010). Recently, a demand of biopolymers  
 82 for various industrial, biotechnological and environmental  
 83 applications like flocculation, settling, dewatering of sludge,  
 84 dyes and metal removal from wastewater has rekindled the  
 85 interest in EPS production (Nontembiso et al., 2011; Zhang et al.,  
 86 2012).

87 The main characteristic of EPS is to enhance aggregation of  
 88 bacterial cells and suspended solids (SS). Adhesion and cohesion  
 89 occur between EPS and the biomass along with suspended  
 90 solids by complex interactions such as London forces, electro-  
 91 statics interactions and hydrogen bonding, which leads to the  
 92 formation of flocs. These EPS properties make them suitable  
 93 for many applications such as sludge flocculation, settling,  
 94 dewatering, metal binding and removal of toxic organic  
 95 compounds (Chien et al., 2013; Jia et al., 2011; Nouha et al.  
 96 2016; Solís et al., 2012).

97 Microbial EPS biosynthesis promotes the attachment of the  
 98 cells to a solid support. It helps in the establishment and  
 99 continuation of microbial colonies to a mature biofilm structure  
 100 and protects from environmental stress. Rehm (2010) published  
 101 a review on critical EPS biosynthesis and metabolic pathways.  
 102 EPS biosynthesis pathway depends on the type of EPS being  
 103 produced i.e., homopolysaccharides or heteropolysaccharides.  
 104 Three major steps involved in EPS synthesis are (i) assimilation  
 105 of a carbon substrate, (ii) intracellular synthesis of the polysac-  
 106 charides and (iii) EPS exudation out of the cell (Vandamme et al.,  
 107 2002). However, these EPS production steps depend on multiple  
 108 factors like the microbial species (genes involved in EPS

synthesis), media composition (carbon and nitrogen source, 109  
 C/N ratio), and operating conditions (pH, temperature, dissolved 110  
 oxygen). 111

Many EPS extraction methods have been used to extract 112  
 EPS produced by pure microbial cultures (laboratory condi- 113  
 tions) and mixed culture (activated sludge) (Nguyen et al., 114  
 2016; Nouha et al., 2016a, 2016b). Chemical, physical and 115  
 combination of both methods were used for EPS extraction 116  
 (Comte et al., 2006a; Nguyen et al., 2016; Nouha et al., 2016a, 117  
 2016b). The efficiency of EPS extraction by different methods 118  
 have been compared (Comte et al., 2006a; Liu and Fang, 2002) 119  
 based on the quantity and the composition of extracted EPS. 120  
 EPS is mainly composed of carbohydrates and proteins. 121  
 Carbohydrate was mainly observed in EPS produced from 122  
 pure cultures, whereas proteins were found in higher quan- 123  
 tities in the sludge-EPS of many wastewater treatment plants 124  
 (Liu and Fang, 2002). However, the EPS chemical structure 125  
 (functional group), molecular weight (MW) and its effect on 126  
 bioflocculant activity were greatly limited by extraction 127  
 methods, which were never reviewed. 128

Scientific findings on general metabolism required for EPS 129  
 precursor biosynthesis and different metabolic engineering 130  
 strategies for EPS overproduction in some bacterial strains are 131  
 reported in this review. Secondly, the significant recent 132  
 developments concerning the impact of extraction methods 133  
 on EPS composition, chemical structure and molecular weight 134  
 was critically reviewed and discussed in the ambit of sludge 135  
 flocculation. 136

## 1. Composition of EPS

137

The chemical structure of polymeric substances secreted by 139  
 the microbial cells depends on the different environmental 140  
 conditions they grew, which are highly diversified. The most 141  
 investigated components of EPS are polysaccharides and 142  
 proteins (More et al., 2012; Nouha et al. 2016; Subramanian 143  
 et al., 2010). The presence of humic substances and nucleic 144  
 acids as part of EPS extracted from sludge were also reported 145

146 in some of the previous studies (Nguyen et al., 2016; Nouha  
147 et al. 2016; Sutherland, 2001).

### 148 1.1. Polysaccharides (carbohydrates)

149 Most EPS produced by microorganisms contains carbohydrate  
150 or polysaccharides. Microbial exopolysaccharides are com-  
151 prised of either homopolysaccharides or heteropolysaccharides  
152 (Monsan et al., 2001). Homopolysaccharides are composed  
153 of only simple sugars and heteropolysaccharides contain  
154 repeated units of various monosaccharides such as D-glucose,  
155 D-galactose, L-fructose, L-rhamnose, D-glucuronic acid,  
156 L-guluronic acid and D-mannuronic acid. For example, alginate  
157 is a heteropolysaccharide produced by *Pseudomonas aeruginosa*  
158 and *Azotobacter vinelandii*, which is composed of D-mannosyl  
159 and L-glucuronosyl residues. However, dextran, a homopoly-  
160 saccharide consisting only dextrose (glucose) units, is produced  
161 by *Leuconostoc* sp. and *Streptococcus* sp. (Rehm, 2010). The  
162 carbohydrate content of EPS can get affected by various factors  
163 during the production and extraction of EPS. The major factors  
164 that significantly affect the carbohydrate content of EPS are the  
165 microorganism, carbon substrate, nutrients (N, P) and the  
166 extraction method utilized for extraction.

167 The microbial species is also one of the main factors that  
168 define the composition of EPS produced based on their  
169 genetics and metabolic pathways, although, the same strain  
170 can also produce EPS with different concentrations and com-  
171 positions when fed with various carbon or nitrogen source  
172 in the cultivation media. It was reported that *Lactobacillus*  
173 *delbrueckii* produced 175 mg/L of EPS using glucose as carbon  
174 source whereas only 69 mg/L of EPS was obtained from  
175 fructose (Yuksekdag and Aslim, 2008).

176 The use of different carbon sources had a considerable  
177 change in EPS concentration and composition. Ye et al. (2011)  
178 reported that the polysaccharide content in loosely bound EPS  
179 (LB-EPS) produced in the activated sludge using acetate was  
180 lower than that of grown in starch or glucose. The possible  
181 cause of this phenomenon can be due to the different  
182 metabolic pathways employed by the microorganism to  
183 metabolize glucose and sodium acetate. Sodium acetate can  
184 enter the citric acid cycle directly, but glucose and starch have  
185 to be degraded to pyruvate and then oxidized to form acetyl-  
186 CoA before it enters the citric acid cycle (Ye et al., 2011).

187 Furthermore, the effect on the content of EPS components  
188 was evaluated by varying nitrogen and phosphorus ratio by Hoa  
189 et al. (2003). The content of total EPS produced in AS (activated  
190 sludge) media ranged from 24.4 to 89.9 mg/g SS with 16 to 94%  
191 carbohydrate component of the total EPS. It was reported that  
192 phosphorus had a more significant effect on the carbohydrate  
193 content of EPS than nitrogen (Hoa et al., 2003).

194 Shin et al. (2001), reported that maximum EPS concentration  
195 observed by physical extraction methods were 166 mg/g DW  
196 (dry weight) of EPS and 183 mg/g DW of EPS from sludge A and  
197 B, respectively (Comte et al., 2006b). However, a low content of  
198 24–53 mg polysaccharides/g EPS DW was observed when  
199 chemical extraction methods were used. Thus the carbohydrate  
200 content of extracted EPS varied widely as a function of sludge  
201 origin and the extraction conditions or the method used.

202 Therefore, the variation in carbohydrate content of EPS can  
203 be attributed to factors like media composition (carbon and

nitrogen source), extraction methods and growth conditions, 204  
which in turn can affect the EPS bioflocculant property. 205

### 1.2. Protein 206

Ton-That et al. (2004) stated that the protein was the principal 207  
component of the EPS matrix in the activated sludge and EPS 208  
(protein) production was not hugely affected by the type of 209  
substrates used for microbial growth. These results were in 210  
agreement with the observations of Frolund et al. (1995) and 211  
Liu et al. (2007) who also reported a consistent protein content 212  
(in activated sludge EPS), when microbe was supplied with 213  
different types of carbon sources (glucose, sodium acetate). 214  
Hoa et al. (2003) investigated the effect of nitrogen supple- 215  
mentation and reported that the protein content of EPS could 216  
be affected by nitrogen (NH<sub>4</sub>Cl) limiting situations, which 217  
result in an increase of protein content of EPS (1.25 to 8.56 mg 218  
protein/g SS). It was found that the protein content of EPS was 219  
inversely proportional to nitrogen content in the activated 220  
sludge, while it remains unaffected by phosphorus. 221

### 1.3. DNA and humic substances 222

DNA or nucleic acid is an intracellular component once 223  
released by cell lysis, which could be adsorbed to EPS matrix. 224  
Humic substances are components which are present natu- 225  
rally in activated sludge from hydrolysis of organic residues. 226  
The humic substances get adsorbed to EPS matrix (biofilm) by 227  
different functional groups like a carboxylic and phenolic 228  
group. A biofilm is defined as an aggregation of bacteria 229  
enclosed in a matrix consisting of a mixture of polymeric 230  
compounds (Vu et al., 2009). 231

Nucleic acids and humic substances have been reported to 232  
influence the rheological properties and stability of biofilms 233  
(Neu, 1996). The extracellular DNA (eDNA) is required for the 234  
initial establishment of biofilms by *P. aeruginosa*. The eDNA 235  
helps in bacterium-surface adhesion by modulating charge 236  
and hydrophobicity interactions between the microbe and the 237  
abiotic surface (Nguyen et al., 2016). Similarly, the biofilm is 238  
formed by many other bacteria that specifically release DNA 239  
in stress conditions or due to cell lysis (Marvasi et al., 2010). 240

As evident from the discussion above, EPS biochemical 241  
composition is affected by many factors like microbial species, 242  
carbon source, nutrient supplementation and the downstream 243  
extraction methods. The composition of the EPS molecule is 244  
very important as it determines ultimately the functional 245  
property of the molecule as bioflocculant. The chemical 246  
composition of the EPS thus produced can determine its 247  
suitability for various kinds of applications. Among the various 248  
novel applications that EPS can be used for, metal removal is the 249  
most prominent one. EPS as carbon substrate has drawn the 250  
significant attention of researchers and the subsequent section 251  
is dedicated to these two applications of EPS. 252

## 2. EPS biosynthesis 253

Extracellular polysaccharide synthesis by microorganisms is 255  
accomplished by a specific secreted enzyme (polymerization 256  
and precursor synthesis enzymes), and synthesis can occur 257

either outside the cell or within the cell wall (Roger, 2002). Table 1 present the several classes of polymers and their diverse characteristics.

The EPS biosynthesis pathway can be divided into three major steps: (i) synthesis of precursor substrate, (ii) polymerization and cytoplasmic membrane transfer and (iii) export through the outer membrane (Fig. 1). These three steps vary with carbon source used, from one microorganism to the other and specifically depends on polymers classes.

### 2.1. Synthesis of precursor substrate

These steps involved in the conversion of intermediate sugar metabolites into the EPS precursor, such as nucleoside diphosphate sugars (for example Guanosine diphosphate (GDP)-sugar) corresponding to substrate or carbon source assimilated. Sugar nucleosides (nucleoside diphosphate sugars) provide an active form of the monosaccharides and also provide the bacterial cell with a means of interconversion of various monosaccharides through epimerization, dehydrogenation and decarboxylation reactions.

Polymer-specific enzymes are required for biosynthesis of the active polymer precursor, which is the first committed step and has been targeted by metabolic engineers to enhance polymer production and to allow the synthesis of tailor-made polysaccharides. In this context, for each type of polymers (dextran, xanthan, and alginate) specific precursors and specific enzymes were involved in their biosynthesis (Lin and Hassid, 1966). For example, uridine diphosphate (UDP)-glucose is the direct precursor of cellulose synthesis by *Acetobacter xylinum* and pullulan production by *Aureobasidium pullulans*, using uridine diphosphate glucose (UDPG) pyrophosphorylase and glucosyltransferase activity, respectively (Duan et al., 2008;

Yoshinaga et al., 1997). Similarly, every polymer has a dedicated precursor and enzymes which vary from organism to organism.

### 2.2. Polymerization and cytoplasmic membrane transfer

The second step in EPS biosynthesis involves the transfer of the precursor nucleoside diphosphate polymerization of the monomers to polymer. Monosaccharides activation by the formation of sugar nucleotides complex is followed by sequential addition of the sugars on to an isoprenoid lipid and simultaneous addition of acyl groups. Highly specific sugar transferase enzymes facilitate a transfer of the monosaccharides and acyl groups to isoprenoid lipid acceptors (bactoprenol, C55-isoprenoid lipid) located in the cytoplasmic membrane. The oligosaccharide repeating units with acetyl, pyruvyl and other acyl adornments are then polymerized. After polymerization of the repeating units, the polysaccharide is excreted through the outer cytoplasmic membrane, which might be coordinated via the formation of a multi-protein complex involving cytoplasmic and outer membrane proteins as well as periplasmic proteins.

The biosynthetic pathway of xanthan (Fig. 1 as an example of polymer biosynthesis) has been explored by (Rosalam and England, 2006). The synthesis of Xanthan starts with the assembly of repeating pentose units (GDP-mannose and UDP-glucuronate). These units are then polymerized by GumE, which is the catalytic subunit of the xanthan polymerase, localized in the cytoplasmic membrane and then produces the macromolecule xanthan (Fig. 1). Once xanthan is synthesized, it is exuded into the extracellular environment.

In the case of Alginate, this step requires the transfer of the cytosolic precursor GDP-mannuronic acid across the cell membrane and the polymerization of the monomers to

**Table 1 – Metabolic characteristic of Bacterial polymers and their fermentation conditions.**

EPS	Polymer localization	Precursors	Polymerization enzyme	Fermentation conditions	Microorganisms	EPS (g/L)	Reference
Cellulose	Extracellular	UDP-d-glucose	Cellulose synthase (BcsA)	Glucose/fructose pH = 4-5; 30°C; 40 hr	<i>Acetobacter xylinum</i>	7-23.9	Hwang et al. (1999)
Dextran	Extracellular	Saccharose	Dextranucrase (DsrS)	Sucrose pH = 5.5; 35°C; 100 kPa; 5 days	<i>Leucomostoc</i> sp.	8-17	Santos et al. (2000)
Xanthan	Extracellular	UDP-glucose, GDP-mannose and UDP-glucuronate	Xanthan polymerase (GumE)	Molasse pH = 7; 28°C; 100 kPa; 24 hr	<i>Xanthomonas campestris</i>	50	Kalogiannis et al. (2003)
Alginate	Extracellular	GDP-mannuronic acid	Glycosyl-transferase (Alg8)	Glycerol + ethanol pH = 5.8-6.5; 28°C; 150 r/min; 48 hr	<i>Pseudomonas</i> sp.	15.2	Hay et al. (2010)
Pullulan	Extracellular	UDP-d-glucose	-	Sucrose pH = 4-4.5; 30°C; 100 hr	<i>Aureobasidium pullulans</i>	1.3-5 2.5	Jiang (2010)
Curdlan	Extracellular	UDP-glucose	Curdlan synthase (Crds)	Glucose/sucrose pH = 5.5; 22-26°C; 3-4 days	<i>Rhizobium</i> spp.	1-5	Pavlova et al. (2005)
Others EPS				Glycérol/glucose 6-18 pH = 7; 30°C, 4 days	<i>Enterobacter</i> sp.	6-18	Alves et al. (2010)
				Sucrose/maltose pH = 6.8-9.8; 54-87°C	<i>Geobacillus</i> sp.	0.1-14	Kambourova et al. (2009)
				Sucrose/glucose pH = 7; 32-37°C	<i>Halomonas</i> sp.	1.6-4.5	Béjar et al. (1998), Poli et al. (2009)

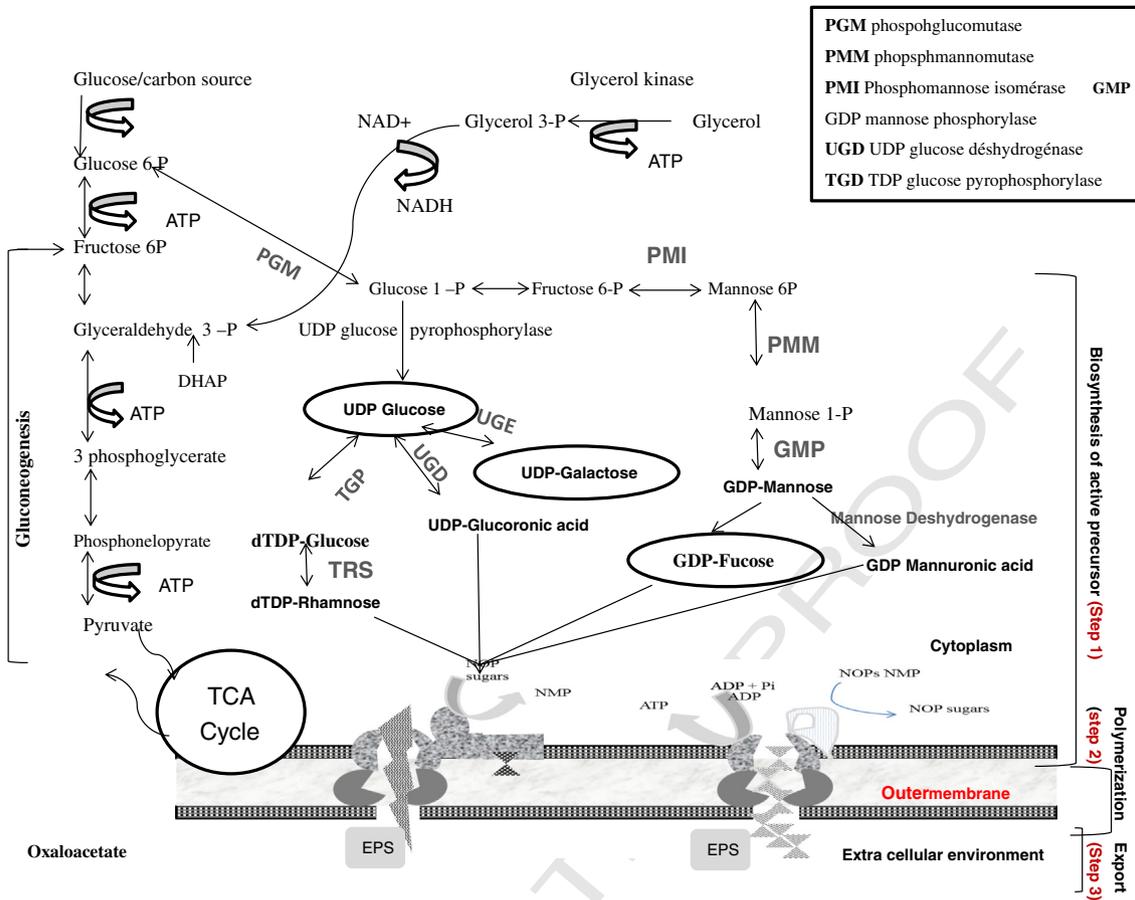


Fig. 1 – Biosynthesis pathway steps of bacterial polymers.

320 polymannuronate using Glycosyl-transferase (Alg8) (Figs. 1  
321 and 2) (Rehm, 2010).

### 322 2.3. Export through the outer membrane

323 The final stage of EPS is the secretion from the cytoplasmic  
324 membrane. It involves passage across the periplasm, outer  
325 membrane and finally excretion into the extracellular environ-  
326 ment. AlgE is the gene which produces the enzyme involved for  
327 Alginate export and GumJ is the gene producing the enzyme  
328 responsible for Xanthan excretion (Fig. 2).

329 In EPS synthesis, lipid transporters provide an anchor to the  
330 extracellular membrane, which facilitates the precise and  
331 orderly formation of the carbohydrate chain proceeded by the  
332 transport of the chain through the cell membrane. Polysaccha-  
333 rides are polymerized on the inner side of the cytoplasmic  
334 membrane and then directly exported through the intermediary  
335 of a lipid transporter. These transporters are long-chain  
336 phosphate esters and isoprenoid alcohols, identical to those  
337 described in the biosynthesis of lipopolysaccharides and pepti-  
338 doglycans (Sutherland, 1999). They play an important role in  
339 heteropolysaccharide synthesis, which is combined with the  
340 EPS excretion. After excretion, the intervention of an enzyme  
341 specific to the EPS may liberate the polymer. Table 1 shows  
342 further categorization dividing the polysaccharides into repeat-  
343 ed unit polymers and non-repeating polymers, presenting their  
344 main compounds, precursors and polymerizing enzymes.

In conclusion, the three dedicated steps of EPS synthesis 345  
requires an array of dedicated genes working in a much- 346  
regulated manner. These genes are translated to yield the 347  
proteins, which eventually perform the tortuous task of EPS 348  
synthesis. Molecular biologist and genetic engineers have 349  
targeted these genes and proteins in order to engineer the 350  
strains to have EPS of desired quality and quantity. It will be 351  
interesting to understand and overlay different molecular 352  
engineering approaches in the ambit of overproduction of EPS. 353

### 3. Engineering strategies for bacterial polysaccharides biosynthesis

354  
356

Several studies were performed to genetically engineer the 357  
EPS-producing microbes to produce novel polymer variants 358  
while improving the production. As presented in the previous 359  
section (Section 2), various enzymes are involved in all three 360  
stages of the metabolic pathway of EPS biosynthesis. Gene- 361  
encoded enzymes regulate the formation of nucleotide sugar 362  
metabolite, chain length determination, repeat unit assembly, 363  
polymerization and export of polymers (Figs. 1 and 2) (Broadbent 364  
et al., 2003). 365

Recently intensive researches have been performed by 366  
focusing on the underlying mechanisms behind bacterial 367  
exopolysaccharide biosynthesis pathways, on different operons, 368  
promoters and the expression of regulatory gene segments. The 369

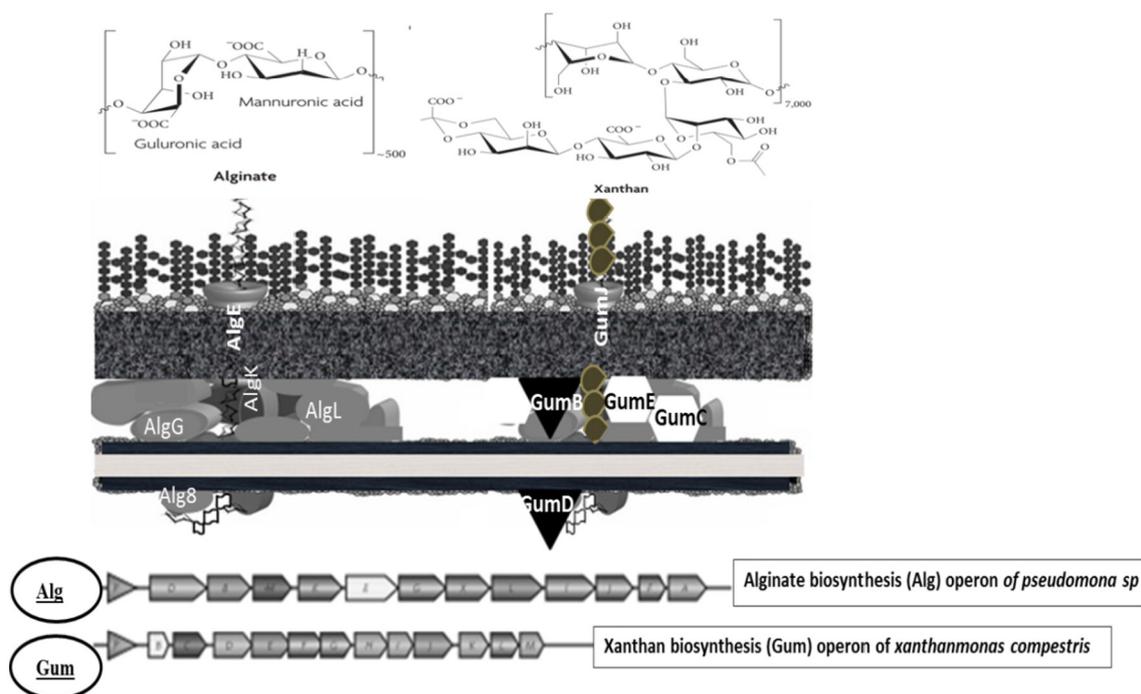


Fig. 2 – Model of alginate and xanthan biosynthesis and secretion mechanisms.

variability of sugar precursors, protein structure analysis, and new bioinformatics tools provide new avenues to enhance the EPS biosynthesis and understand the principal engineering strategies of EPS formation.

One of the primary goals of EPS production engineering is to increase the volumetric productivity of EPS in a very cost effective manner. In this context, we focus on the recent advances in potential engineering strategies for better EPS production. Vorhölter et al. (2008) attempted to increase the pool of sugar nucleotides (EPS precursors) to enhance the carbon flux toward the final polymer yield. Guo et al. (2014) studied xanA gene producing phosphoglucomutase (PGM) and phosphomannomutase (PMM) enzyme, which is involved in the conversion of glucose-6-phosphate to glucose-6-phosphate. They found that xanA is a regulator gene and it has a key role in precursor metabolite overexpression.

Researchers (Huang et al., 2013; Schatschneider et al., 2013; Wu et al., 2014) studied *Xanthanomonas campestris* EPS production in detail and found that it possess a series of 12 genes embedded in tandem. This operon includes seven genes needed for monosaccharide transfer and acylation of lipid intermediate to form the completely acylated repeating unit (Figs. 1 and 2). It has been suggested that alteration in promoters related to this operon can yield higher precursor metabolites (Galindo et al., 2007).

Vojnov et al. (1998) studied the gum-protein operon containing gumBCDEFGHIJKLM (Fig. 2) gene fragment. They tested a simple idea of whether inclusion of an additional promoter as upstream gumC may improve xanthan biosynthesis. It was found that promoter insertion to upstream of gumC resulted in enhanced yields and overexpression of the transcription of an operon and eventually increased the xanthan biosynthesis from 66 mg cell mass to 119 mg/g cell mass (Vojnov et al., 1998).

Schatschneider et al. (2013) studied most sensitive gene segments that significantly affect EPS synthesis. The study demonstrated that the overexpression of gumD is the key enzyme involved in the EPS assembly construction and precursor conversion (Fig. 2). Thus, it was suggested to clone the entire gumD gene cluster of a 16 kb chromosomal fragment with high copy number in *X. campestris*. The results indicated elevated expression of all biosynthetic eps gene, which could be achieved by cloning them on a high copy number plasmid (Janczarek et al., 2009).

In another strategy, the idea was to engineer the EPS molecules at the molecular level for having the desired behavior and material characteristics of the final polymer while improving the property as biofloculant. For example, this molecular alteration can be deleting substituents or monomeric sugar residue from the side chain.

Deactivation of the GT GumI gene resulted in a truncated tetramer xanthan version, as a consequence of deletion of the terminal mannose, it was found to have much lower viscosity. Similarly, inactivation of GumK (a GT) causes the removal of glucuronic acid side chains which resulted in an enhanced viscosity of the EPS as compared to wild type EPS (Hassler and Doherty, 1990). The gene deactivation is performed by homologous insertion of foreign genes within the operon segment at active gene locus (GT GumI and GT GumK). They have reported that transgenes can suppress their expression and that of endogenous homologous genes. This phenomenon has been called co-suppression (Hassler and Doherty, 1990).

Many efforts were done in engineering the degree of acetylation and pyruvylation of various polymers (alginate and xanthan), to control their rheological properties (Donati and Paoletti, 2009). The level of O-acetylation and pyruvylation can be controlled reasonably well using specific strains/mutants or altering the growth media and controlling cultivation conditions

438 such as aeration, pH and temperature (Gaytán et al., 2012; Peña  
439 et al., 2006).

440 Interesting insights were given by Rehm (2010) and Galván  
441 et al. (2013) regarding the general structure and functional  
442 relationships. The extent of acetylation and pyruvylation has  
443 antagonistic effects on viscosity. The GumL enzyme incorpo-  
444 rates pyruvyl residues to the external  $\beta$ -mannose residues,  
445 while the acetyl residues are incorporated into internal  
446  $\alpha$ -mannose units by GumF, and external  $\beta$ -mannose by GumG  
447 (Becker et al., 1998). A high degree of pyruvylation by increasing  
448 the transcription of gumL (cloning an additional promoter  
449 upstream gumL) resulted in higher viscosity, whereas more  
450 acetyl group decreased the viscosity of the resulting EPS (Gaytán  
451 et al., 2012; Rehm, 2015).

452 Taxonomically different microbes can produce the same  
453 types of extracellular compounds with different concentra-  
454 tions. *P. aeruginosa* and *A. vinelandii*, despite having most of the  
455 genes involved in the biosynthesis, their organization, regu-  
456 lations and genetic switch clusters identical have differences  
457 at transcriptional and functional level expression of EPS. In  
458 *P. aeruginosa* and *A. vinelandii* a cluster of 13 structural and five  
459 regulatory genes (Hay et al., 2010) involved in EPS biosynthesis  
460 (Fig. 2). In *P. aeruginosa* the transcriptome is regulated by algR,  
461 algB, algC and algD gene segment that are algT dependent  
462 whereas in *A. vinelandii*, these genes were independent of  
463 algT. The regulator gene algT encodes for the regulatory  
464 expressions of sigma factor ( $\sigma^{22}$ ) which could explain the  
465 variation in alginate concentration (Ahmed, 2007).

466 Though literature exists for genomic and proteomic level  
467 engineering to overproduce classical biopolymers, recombi-  
468 nant molecular engineering techniques are near to inexistent  
469 for EPS over production. Molecular techniques can be applied  
470 for transcriptional overexpression of RNAs involved in bio-  
471 synthetic pathways, translational overexpression of the  
472 proteome involved in EPS biosynthesis. The most generic  
473 approach is an in-frame insertion of a strong promoter  
474 upstream of EPS operon to have more than basal level of  
475 constitutive or inductive expression of genome and proteome  
476 level. Overexpression of EPS can be induced when the carbon  
477 flux channeling is streamed favorably toward the generation  
478 of precursor molecules like nucleotide diphosphate-Glucose  
479 conjugates by gene silencing, which divert carbon flux away  
480 from EPS synthesis without compromising the survival of the  
481 organism. Similarly, overexpressing enzymes involved in  
482 irreversible synthesis is a key mechanism and can be used  
483 as control points to induce diversion of this excess carbon  
484 pool toward EPS biosynthesis.

485 Bacteria produce a wide range of exopolysaccharides, which  
486 are synthesized via different biosynthesis pathways. A better  
487 understanding of basic biochemistry and genetics involved in  
488 exopolysaccharides biosynthesis and the regulatory mecha-  
489 nisms is critical for protein engineering approaches to produce  
490 novel polymers. At large scale production process having highly  
491 efficient downstream extraction is as important as enhancing  
492 the upstream process. Choice of downstream extraction meth-  
493 od should be made diligently to obtain the maximum product  
494 yield without hindering its natural properties. In the following  
495 sections, the impact of various operational process parameters  
496 and downstream extraction process on EPS quantity and  
497 quality was discussed.

## 4. EPS production methodology 498

### 4.1. EPS extracted directly from sludge 500

501 EPS has been successfully extracted directly from sludge by  
502 Urbain et al. (1993). They indicated that the EPS proteins  
503 extracted directly by ultra-sonication from municipal waste-  
504 water sludge were higher (97.8 mg/gVSS) as compared to  
505 15.6 mg/g VSS of carbohydrate content. Liu and Fang (2002)  
506 efficiently performed EPS extraction from acidogenic and  
507 methanogenic sludge. The results revealed that carbohydrate  
508 was a major component in acidogenic sludge (62% w/w of  
509 EPS), while protein was a powerful component in methano-  
510 genic sludge (41% w/w of EPS). 510

511 Further, researchers (Peng et al., 2012; Sponza, 2003)  
512 examined the EPS extracted directly from activated sludge  
513 treating various types of wastewaters. They found that the  
514 protein content of EPS is higher in the sludge treating winery  
515 and municipal wastewater than that of sludge produced in  
516 treating pulp-paper, textile and petrochemical wastewaters.  
517 Thus the EPS concentration varied accordingly to the type of  
518 wastewater treated in WWTPs. Simon et al. (2009), reported  
519 that the nucleic acid content of EPS could be affected by the  
520 type of sludge from different municipal WWTP. High nucleic  
521 acid (7 mg/g EPS DW) concentration was observed in Eerbeek  
522 municipal WWTP sludge as compared to 4 and 1 mg/g EPS DW  
523 in Emmtec and Revico WWTP, respectively. 523

### 4.2. EPS production using pure carbon sources 524

525 Although some researchers are convinced that the nature of  
526 the substrate cannot influence the composition of the EPS  
527 produced, however, many others have stated that medium  
528 composition either carbon source or nitrogen sources are  
529 important parameters in EPS biosynthesis and production  
530 (Simon et al., 2009). 530

531 Li and Yang (2007) reported that the activated sludge fed  
532 with glucose exhibited more EPS concentration than that fed  
533 with acetate. On the contrary, Ye et al. (2011) revealed that  
534 sludge fed with acetate was more favorable for high EPS  
535 production than that fed with glucose. Their results were also  
536 studied recently by (Geyik et al., 2016), which explained how  
537 the type of organic substrate in a wastewater affects the  
538 production and composition of EPS. The activated sludge  
539 reactors were operated with three different feeds composed of  
540 various organic compounds, first of peptone, glucose, and  
541 acetate; then the second feed was only using only peptone  
542 and third feed with only glucose. They proved that the type of  
543 substrate affected the relative proportion of protein and  
544 polysaccharide content of EPS. 544

545 The effect of substrate and its suitability depends on upon  
546 the organisms. Different bacteria may produce different  
547 bioflocculants in composition and structure. For example,  
548 *Bacillus licheniformis* (Shih et al., 2001) and *Nocardia amarae*  
549 (Takeda et al., 1992) produce protein bioflocculants whereas  
550 *Bacillus subtilis* (Yokoi et al., 1995) produce polysaccharide  
551 bioflocculant, and glycoproteins were produced by *Arcaudendron*  
552 sp. and *Arathrobacter* sp. (Wang et al., 1995). Van Geel-Schutten  
553 et al. (1998) reported that the *Lactobacillus* strains produce 553

554 varying amounts of EPS with different sugar compositions  
 555 when they are grown on sucrose, raffinose and lactose as a  
 556 carbon source. The EPS produced by *Lactobacillus casei* with  
 557 glucose as carbon source was reported to be different from that  
 558 produced on lactose (Pham et al., 2000). Vijayendra et al. (2003)  
 559 investigated the effect of different hexose sugars (glucose,  
 560 sucrose and lactose) on EPS production by *Agrobacterium*,  
 561 *Alcaligenes*, *Pseudomonas* and *Xanthomonas*. Lactose was found  
 562 to be the best carbon source for EPS production by *Pseudomonas*  
 563 sp. whereas other bacterial strains favored sucrose.

564 Yuksekdag and Aslim (2008) studied the impact of various  
 565 carbon sources on EPS production by *L. delbrueckii*, *Lactobacillus*  
 566 *bulgaricus* and *Streptococcus thermophiles*. For all the strains,  
 567 glucose was the most efficient carbon source and EPS  
 568 concentration of 211 mg/L produced by *L. delbrueckii sub sp.*  
 569 was the highest concentration. The effect of carbon source  
 570 on EPS synthesis by the marine bacterium *Saccharophagus*  
 571 *degradans* was also studied by González-García et al. (2015).  
 572 *S. degradans* was able to grow in the mineral medium while  
 573 producing EPS concentration depending on the carbon source:  
 574 with glucose or starch, EPS 1.5 g/L; with galactose, sucrose, or  
 575 xylose, EPS 0.7 g/L and with fructose, EPS 0.3 g/L. The results  
 576 were in agreement with the recent studies (Qin et al., 2015;  
 Q11 Mane and Hamde, 2015). Lactose gave the maximum EPS  
 578 concentration of 6.9 g/L as compared to 0.9 g/L when glucose  
 579 was used (Qin et al., 2015). According to Mane and Hamde  
 580 (2015), glucose gave maximum EPS concentration of 750 mg/L  
 581 as compared to other substrates (lactose 390 mg/L; sucrose  
 582 670 mg/L) used.

583 Modification in feed media affects the sugar composition  
 584 of EPS produced. The EPS composition characterized by 61.7%  
 585 of galactose and 33% of glucose and 5.3% of mannose was  
 586 produced by *L. bulgaricus* using milk as media. However, 39.7%  
 587 of galactose, 57.9% of glucose and 1.8% of mannose of total  
 588 carbohydrate composition was found when the glucose was  
 589 used as a carbon source (Petry et al., 2000).

590 In another case, the biosynthesis of extracellular polysac-  
 591 charides in *A. xylinum* was improved by using galactose and  
 592 xylose as carbon source compared to fructose, sucrose and  
 593 starch in the medium (Sutherland, 2001). Recently, EPS from  
 594 *Pseudomonas fluorescens* was produced using different concen-  
 595 trations of sucrose and sugarcane molasses as the carbon  
 596 substrates (Razack et al., 2013). Maximum EPS concentration  
 597 of 2843 mg/L was obtained at 5% (w/v) sugarcane molasses  
 598 concentration in the media. The sugarcane molasses as  
 599 carbon source gave a higher EPS concentration than sucrose  
 600 with 1389 mg/L of EPS concentration (Razack et al., 2013).

601 It has been observed that selection of microbial source and  
 602 growth substrates type, concentration and composition have  
 603 a significant effect on EPS yield.

#### 604 4.3. Use of sludge as nutrient source for EPS production

605 Recently EPS production using sterilized sludge as nutrient  
 606 and carbon source using pure bacterial strain isolated from  
 607 wastewater sludge have been reported by Subramanian et al.  
 608 (2010) and More et al. (2012). They indicated that different  
 609 pre-treatments (heat, alkaline and acidic treatment) of sludge  
 610 could vary the EPS production of pure cultures. More et al.  
 611 (2014) also used a consortium of pure bacterial strains to

improve the EPS production. Nouha et al. (2016) studied the  
 612 sterilized sludge as a culture media using *Cloacibacterium*  
 613 *normanense* for EPS production, and the sterilized sludge  
 614 inoculated with pure culture also fed with crude glycerol as  
 615 extra carbon source. High EPS concentration up to 13.3 g/L  
 616 was recorded using only sludge as a growth medium, and  
 617 21.3 g/L of EPS was produced when the medium was  
 618 supplemented with 20 g/L of crude glycerol. 619

In addition to the carbon source, the extraction methods  
 620 can also influence the EPS yield, chemical structure, molecu-  
 621 lar weight and their role as bioflocculant in wastewater  
 622 treatment. Furthermore, the EPS produced by varying differ-  
 623 ent optimized operating conditions has a different composi-  
 624 tion. Consequently, appropriate methods of extraction should  
 625 be chosen to obtain desired EPS properties, which are required  
 626 for specific applications. For this reason, different character-  
 627 istics of EPS must be considered, including identification of  
 628 the component monosaccharides and their relative propor-  
 629 tions and the physicochemical properties of the final EPS. 630

## 5. EPS extraction 632

Several methods (centrifugation, sonication, heating, EDTA and  
 633 CER) have been applied in different studies to extract EPS from  
 634 pure cultures or undefined mixed cultures, mainly related to  
 635 activated sludge and biofilms. The EPS extraction methods  
 636 include various physical, chemical methods or their combina-  
 637 tions. The extraction procedure must be selected considering  
 638 the efficiency of the method to extract EPS in high yield. The  
 639 best extraction method should not disturb the interactions that  
 640 keep the EPS components together in the matrix. 641

### 5.1. Physical methods 642

Many physical methods have been tested to evaluate their  
 643 extraction efficiencies and compare them to select the best  
 644 technique of extraction. Comte et al. (2006b) studied three  
 645 extraction techniques (centrifugation, sonication and heating).  
 646 EPS was extracted with very high efficiencies by using heating  
 647 method (82 mg/g VSS). High extraction of protein and polysac-  
 648 charides content was obtained with a heating method as  
 649 compared to others physical methods used (centrifugation  
 650 and sonication). These results were in agreement with Tapia  
 651 et al. (2009) studies. The concentration and composition of EPS  
 652 extracted by heating (813 mg/g DW) were higher to that  
 653 obtained by centrifugation (735 mg/g DW) method. Pan et al.  
 654 (2010) compared two physical extraction methods (centrifuga-  
 655 tion and ultra-sonication). They also observed that protein  
 656 content was low in EPS samples prepared by centrifugation, as  
 657 compared to the protein content in EPS sample extracted by  
 658 ultra-sonication. 659

In this context, each researcher explored the reason for the  
 660 variation in extraction efficiency while employing different  
 661 EPS extraction methods, such as the physical methods. Comte  
 662 et al. (2006b) proposed a hypothesis to explain these variations  
 663 in results, that the proteins and polysaccharide moieties of EPS  
 664 could be hydrolyzed during the extraction procedure by  
 665 heating. According to Tapia et al. (2009) the heating extraction  
 666 procedure allowed to extract the strongly bound EPS to flocs. 667

668 However, some studies (Frolund et al., 1995; Liu and Fang,  
669 2002) proposed that the high EPS extraction efficiency by the  
670 heating method may be caused by significant cell lysis and  
671 disruption, which may reveal results into high protein content  
672 in EPS.

## 673 5.2. Chemical methods

674 As per literature, many methods have been proposed and  
675 applied to extract EPS from pure or undefined mixed cultures.  
676 The extraction efficiencies of chemical methods, such as cation  
677 exchange resin (CER), EDTA (Ethylene diamine tetra-acetic Acid)  
678 and NaOH methods have been studied (Frolund et al., 1995; Liu  
679 and Fang, 2002; Sheng et al., 2005).

680 Cation exchange resin along with a high concentration of  
681 NaCl has been used for the extraction of adhesive exopolymers  
682 from *Pseudomonas putida* and *P. fluorescens* (Christensen and  
683 Characklis, 1990). The CER method has become the most widely  
684 accepted EPS extraction method, largely because the resin used  
685 for selective extraction of EPS can be removed and recycled  
686 easily.

687 A recent study (Zuriaga-Agusti et al., 2010) used CER and  
688 Triton X-100 as efficient EPS extraction methods to extract EPS  
689 from two different municipal MBRs. The protein and carbo-  
690 hydrate content was determined. This study was performed  
691 to understand the problem of membrane fouling of the MBR  
692 process due to EPS. Although EPS has effect fouling, it is not  
693 clear which EPS fraction (SMP or EPS) or component (proteins  
694 or carbohydrates) exert the most important contribution to  
695 membrane fouling. To elucidate the solution, activated sludge  
696 samples from two municipal MBRs were processed for  
697 extraction protocols comparison. They demonstrated that  
698 the proteins and carbohydrate content using Triton X-100  
699 extraction method was higher ( $81.64 \pm 12.98$  mg BSA/g VSS  
700 and  $10.30 \pm 1.42$  mg Glucose/g VSS, respectively) than cation  
701 exchange resin method ( $16.49 \pm 9.37$  mg/g VSS and  $3.93 \pm$   
702  $2.47$  mg/g VSS, respectively). They observed that CER protocol  
703 achieved lower extraction efficiency for the EPS than Triton  
704 X-100. The different values obtained between CER and Triton  
705 X-100 could be attributed to the floc composition. In fact,  
706 some researchers (Liu et al., 2010; Yu et al., 2009) asserted that  
707 the presence of two types of polymers in sludge flocs. One  
708 type of EPS is tightly bound within micro colonies of cells, and  
709 another is loosely bound in the floc matrix or at outer  
710 peripheries of the flocs. In this way, results may point that  
711 CER protocol withdraws mostly loosely bound EPS, whereas  
712 Triton X-100 extracts both types of EPS. The study of Meng  
713 et al. (2010) found that the EPS extracted by CER protocol was  
714 low comparing to Triton X-100.

715 Comparative study of chemical extraction methods has  
716 been exhaustively investigated. Tapia et al. (2009) conducted a  
717 comparative study of chemical and physical methods to  
718 understand the principal role of the different extraction  
719 methods on EPS composition and concentration. The concen-  
720 tration of EPS from bacterial culture of *Acidiphilium* sp. was  
721 higher when extracted by centrifugation (600 mg carbohy-  
722 drate/g DW, 220 mg protein/g DW, 820 mg EPS/g DW) than  
723 extracted with NaOH (430 mg carbohydrate/g DW, 170 mg  
724 protein/g DW, 650 mg EPS/g DW). However, in the study of  
725 Sheng et al. (2005) the concentration of EPS extracted by NaOH

was higher 159.2 mg/g-DW in comparison to centrifugation 726  
and ion exchange resin (Table 2). These results have also been 727  
confirmed by using more complex microbial consortia such as 728  
activated sludge (Table 2). Liu and Fang (2002) reported 729  
extractions of EPS ranging between 25.7 mg EPS/g volatile 730  
suspended solids (VSS) by centrifugation and 164.9 mg EPS/ 731  
g-VSS by chemical methods (NaOH and formaldehyde). A 732  
significant difference (>60%) of EPS extracted was observed 733  
between the two methods when EPS was extracted from the 734  
same culture. However, it appears that the chemical extrac- 735  
tion becomes more effective when it is combined with 736  
physical methods. Liu and Fang (2002) investigated high 737  
proteins, carbohydrates, humic acid and DNA content ex- 738  
tracted by combined methods of formaldehyde and ultra- 739  
sonication compared to that obtained by only formaldehyde. 740  
Comte et al. (2006a) also proved that sonication and CER as 741  
combined EPS extraction method were more effective to 742  
extract high proteins, carbohydrate and nucleic acid content 743  
comparing to CER or sonication alone. 744

## 5.3. Chemical methods vs physical methods 745

The type of EPS extraction method further influences its 746  
composition, nucleic acid content, protein content, and various 747  
functional properties. Comte et al. (2006a) investigated the 748  
effect of extraction methods from two different sludge on EPS 749  
composition. This study demonstrated that the protein content 750  
was higher (343–337 mg proteins/g EPS DW) with physical 751  
method (sonication), whereas low protein content was ob- 752  
served, 73–107 mg proteins/g EPS DW, with chemical extraction 753  
methods (Glutaraldehyde, Formaldehyde and NaOH). 754

Liu and Fang (2002) and Comte et al. (2006a) indicated a 755  
constant nucleic acid content of EPS using different physical 756  
extraction methods (Cation-exchange resin, centrifugation, 757  
sonication and heating). This study indicated that the physical 758  
methods resulted in lower extent of cell lysis than chemical 759  
extraction methods. 760

Further, Simon et al., 2009 have found the humic sub- 761  
stances content in EPS was different using different extraction 762  
methods (CER, centrifugation and heating). The highest 763  
humic acid concentration was 224 mg/g DW when heating 764  
was used as extraction method in case of anaerobic granular 765  
sludge obtained from Eerbeek municipal WWTP. However, the 766  
lowest humic content was 5 mg/g DW in the case of anaerobic 767  
granular sludge from Revico municipal WWTP by applying 768  
centrifugation as an extraction method. 769

In case of metal absorption  $\text{Cd}^{2+}$  sorption capacity of EPS 770  
extracted using sonication was  $245 \pm 46$   $\mu\text{mol/g}$  of EPS for 771  
Chau's model and  $336 \pm 22$   $\mu\text{mol/g}$  of EPS for Rezić's model. 772  
Although the metal binding capacity of sludge EPS extracted 773  
by physical methods was consistently identical (except by 774  
heating); however, the metal complexation capacity was 775  
significantly improved when EPS was extracted by chemical 776  
methods (Comte et al., 2006b). 777

Moreover, the effect of EPS extraction method from 778  
activated sludge on metal binding ability was evaluated 779  
(Comte et al., 2006b), EPS extracted by physical and chemical 780  
methods showed a greater affinity for Pb ions than Cd ions. 781  
The EPS extracted by a physical method (sonication) had a 782  
 $\text{Pb}^{2+}$  adsorption capacity of  $2135 \pm 55$   $\mu\text{mol/g}$  of EPS and 783

**Table 2 – Comparison of total EPS (carbohydrate and protein) extracted from various cultures and extraction methods.**

	Pure culture	Extraction methods	EPS composition (mg/g DW)				EPS concentration (mg/g- DW)	Prot./carb. ratio	References
			Prot.	Carbo.	HA	DNA			
t2.6	<i>Rhodopseudomonas acidophilap</i>		4.1	6.2	–	–	12.9	1.5	Sheng et al. (2005)
t2.7			6.5	58.5	–	–	70.3	9.0	
t2.8		Centrifugation	7.7	126.5	–	–	159.2	16.4	
t2.9		EDTA	10.3	37.3	–00	–	71.6	3.7	
t2.10		NaOH							
t2.11	Heating								
t2.12	<i>Acidiphilium</i> sp.		600	220	–	–	820	0.3	Tapia et al. (2009)
t2.13			550	200	–	–	750	0.3	
t2.14			430	170	–	–	600	0.4	
t2.15			570	200	–	–	770	0.4	
t2.16	Mixte culture							Liu and Fang (2002)	
t2.17	Aerobic		54.6	40.5	50.4	0.3	165		0.2
t2.18			17.7	12.7	16.4	0.1	109		0.1
t2.19			22.9	12.4	59.2	0.4	146		0.1
t2.20	Formal + NaOH		20.4	28.8	18.9	0.1	78		0.1
t2.21	CER								
t2.22	EDTA								
t2.23	Acidogenic	Formal + ultrasound	110.9	25.8	0.1	15.1	179	4.3	
t2.24			38.7	6.2	0.1	3.0	58	6.2	
t2.25			41.7	6.5	0.2	15.9	105	4.6	
t2.26			71.6	10.8	5.0	0.05	100	6.6	
t2.27	Mutagenic sludge								
t2.28			19.1	42.1	23.3	0.19	102	2.2	
t2.29			7.9	10.6	5.5	0.05	30	1.3	
t2.30			6.8	12.0	24.3	1.20	73	1.7	
t2.31			12.0	13.1	5.6	0.04	30	1.1	
t2.32	Activated sludge								Zuriaga-Agusti et al. (2010) Comte et al. (2006a)
t2.33		Sonication	140	343	62	46	200	2.4	
t2.34		Heating	166	378	126	17	369	2.3	
t2.35		CER	16	4	–	–	24	4.2	
t2.36		Triton X 100	81	10	–	–	100	7.9	
t2.37	Biofilm	Centrifugation	57.0	–	–	–	57	–	Pan et al. (2010)
t2.38		Ultrasonication	22.3	56.6	–	–	79	0.4	
t2.39		EDTA	1.7	3.2	–	–	45	0.5	
t2.40		Formaldehyde	8.0	25.2	–	–	33	0.3	
t2.41		Formal + NaOH	17.0	13.3	–	–	30	1.3	

Prot: Protein; Carb: Carbohydrate.

784 2184 ± 27 µmol/g of EPS as analyzed using two adsorption  
785 models, Chau and Rezig, respectively.

786 Recently, many other factors such as variation of pH,  
787 temperature and mixing speed have been reported to affect  
788 the structure of EPS, and its metal removal efficiency  
789 (AjayKumar et al., 2009; Ruan et al., 2013).

#### 790 5.4. Combination of different methods

791 According to D'Abzac et al. (2010), a combination of formalde-  
792 hyde and heating leads to the highest EPS quantity extracted.  
793 Humic-like substances and the nucleic acids are more readily  
794 extracted using formaldehyde method than the heating  
795 method alone. By using a combination of sonication and CER  
796 methods the protein and polysaccharide contents were found  
797 to be higher than obtained by only sonication or CER.

798 As discussed earlier, principally physical methods are  
799 simply mechanical which can explain the low extraction yield  
800 and it has been a common observation that the chemical

methods were always having higher yields than physical  
801 methods. Only handful of methods has been thoroughly  
802 evaluated and optimized to obtain high extraction efficiencies  
803 without cell lysis and reagent contamination. Most of the  
804 chemical extraction methods cause various problems either in  
805 the extraction or EPS analysis. For instance, in the case of alkali  
806 extraction methods, an addition of NaOH causes anionic  
807 groups, such as carboxylic groups in proteins and polysaccha-  
808 rides to lose their protons. The deprotonation causes a strong  
809 repulsion between EPS molecules within the EPS gel and  
810 provides a higher water solubility of the compounds. Similarly  
811 the EDTA method has high extraction efficiency, however,  
812 causes a high degree of cell lysis and possibly also contamina-  
813 tion with cellular macromolecules interfering in the protein  
814 determination (Comte et al., 2006a).  
815

816 Each and every method has their advantages and disadvan-  
817 tages (Table 3). Although the physical methods cause less cell  
818 lysis, it also has low EPS extraction efficiency. The chemical  
819 methods generated high protein content, and these proteins  
819

**Table 3 – Advantages and disadvantages of different extraction method used.**

Extraction methods	Mechanism and Extraction conditions	Limitation	Advantages	References	
Physical	Centrifugation	EPS separates from cell surface and dissolve to solution under the centrifugal force • 20,000/600 g in 20 min	(-) Low extraction efficiency (-) Bound EPS cannot be extracted.	(+) Simple method (+) No cell lysis	Liu and Fang (2002)
	Heating	Accelerates the EPS solubilization • 80°C, 1 hr	(-) Depend of shear rate (-) Cells disruption	(+) No chemical addition (+) Extraction of bound EPS	Comte et al. (2006a) Frolund et al. (1995)
	Ultra-sonication	Act of applying sound energy or mechanic pressure to agitate particles in a sample • 40 W, 0–30 min	(-) Denaturation of EPS (-) High cell lysis in less time	(+) Dissociation of aggregates	Comte et al. (2006a, 2006b)
Chemical	CER	Removes the divalent cations, causing the EPS to fall apart • 70 g resin/g MVS, 600 r/min, 1 hr	(-) Extract only the proteins and carbohydrates coupled to cations	(+) Avoid EPS pollution by chemical reagent	Liu and Fang (2002)
	EDTA	Remove the multivalent cations forming the bond between the charged compounds of EPS. 150 mL of 2% EDTA added for 3 hr at 4°C	(-) EPS contamination by intracellular compounds (-) Interfere the proteins analysis (-) Cost of chemical	(+) No modification of EPS structure	Liu and Fang (2002) Comte et al. (2006a)
	NaOH	Ionization of carboxylic group • 1N of NaOH, 3 hr at 4°C	(-) Severe disruption in EPS composition (-) high damage of cell (-) Many charged groups results repulsion between the EPS (-) Alkaline hydrolysis of many polymers may take place	(+) Break the covalent disulfide binding in glycoproteins	Liu and Fang (2002) Comte et al. (2006a)
	Formaldehyde	Fix the cell and denature the EPS, linking the proteins and carbohydrate • 36%, 1 hr, 4°C	(-) Modify EPS characteristics (-) Interferences in carbohydrates content (-) Cost of chemical reagents	(+) Prevent the cell lysis	Liu and Fang (2002) Comte et al. (2006a)
Combined chemical and physical methods	NaOH-Heating CER-centrifugation Formaldehyde-sonication	-	(-) Extraction time (-) EPS need purification (-)Not economical	(+) High Extraction efficiency	D'Abzac et al. (2010)

820 can be originated from extracellular enzymes and or intracel-  
821 lular materials contaminations. A combination of two methods  
822 could affect the production cost and the efficiency of EPS  
823 extraction during an industrial application (Domínguez et al.,  
824 2010). There is no simple and single method that exists to  
825 extract 100% of total or complete EPS components from the  
826 microbial cell or activated sludge flocs. Each technique extrac-  
827 tion efficiency depends on many factors mainly the origin of  
828 EPS. It is recommended that extraction is only performed after  
829 running a comparative study of various methods to select the  
830 best one for desired application. Furthermore, an extraction  
831 technique must be chosen and optimized for each case, taking  
832 into account many parameters (such as extraction time, cost  
833 and dosage of chemical used and evaluation of cell lysis), which  
834 could affect the cost and the properties of EPS.

### 5.5. Effect of extraction methods on functional group and molecular weight of EPS

835  
836  
837 The complex composition of EPS makes it difficult to analyze  
838 the conformation, chemical structure (their functional groups)  
839 and distribution of EPS. However, progress in analytical  
840 chemistry has led to the development of new instruments and  
841 techniques for characterization of EPS, which has generated a  
842 significant amount of information on the structural and  
843 functional properties of EPS as well as their molecular weight.  
844 Chromatography, mass spectrometry and their combination  
845 have been used to qualitatively and quantitatively analyze the  
846 EPS composition (Dignac et al., 1998).

847 Many researchers (Sheng and Yu, 2006; Tapia et al., 2009)  
848 proved the effect of the extraction methods applied on structural

**Table 4 – Effect of extraction method on functional groups of EPS.**

Extraction methods	Wave number (cm <sup>-1</sup> )	Transmittance (%)	Vibration type	Functional type	Molecular weight range	References
EDTA	1550–1600	50	Stretching vibration of C=O Proteins	Proteins (peptide bond)	8 kDa	Comte et al. (2006a)
	1300	55	C–N stretching		150–200 kDa	Simon et al. (2009)
NaOH + Formaldehyde	2450	60	Specific band corresponding to a product of a formaldehyde and EPS	Carboxylic acids	–	Comte et al. (2006a)
	1750	40	Stretching vibration of C=O			
	1400	10	Stretching vibration of C=O			
	800	30	Several bands visible	Carboxylates Phosphorus/Sulfur functional group		
Gluraldehyde	1950	75	Specific band corresponding to a product of a glutaraldehyde and EPS reaction	–	–	Comte et al. (2006a)
	1500	70				
Centrifugation	3200–3420	80	Stretching vibration of OH	OH into polymeric compounds	Low molecular weight	Comte et al. (2006a)
	2935	85	Stretching vibration of C=O and C–N (Amide I)		0.16–0.3 kDa	Simon et al. (2009)
	1630–1660	73	Stretching vibration of C–N and deformation vibration of N–H (Amide II)	Proteins (peptide bond)	0.7–2.7 kDa	
	1550–1560	83	Deformation vibration of CH <sub>2</sub>		4.6–6 kDa	
			Stretching vibration of C=O			
			Stretching vibration of OH, of polysaccharide			
	1450–1460	82	Several bands visible	Carboxylates	High molecular weight	
	1400–1410	85		Polysaccharides	16–190 kDa	
	1060–1100	75			270–275 kDa	
	<1000			Phosphor/Sulfur functional group		
Sonication	3200–3420	60	Stretching vibration of OH	OH into polymeric compounds	Low molecular weight	Comte et al. (2006a)
	2935	70	Stretching vibration of C=O and C–N (Amide I)		0.16–0.3 kDa	Simon et al. (2009)
	1630–1660	52	Stretching vibration of C–N and deformation vibration of N–H (Amide II)	Proteins (peptide bond)	0.7–2.7 kDa	
	1550–1560	60	Deformation vibration of CH <sub>2</sub>		4.6–6 kDa	
			Stretching vibration of C=O			
			Stretching vibration of OH, of polysaccharide			
	1450–1460	71	Several bands visible	Carboxylates	High molecular weight	
	1400–1410	65		Polysaccharides	16–190 kDa	
	1060–1100	63			270–275 kDa	
	<1000			Phosphorus or Sulfur functional group		
Sonication + resin	3200–3420	60	Stretching vibration of OH	OH into polymeric compounds	Low molecular weight	Comte et al. (2006a)
	2855	70	Stretching vibration of C=O and C–N (Amide I)	Proteins	0.16–0.3 kDa	Simon et al. (2009)
	1630–1660	52	Stretching vibration of C–N and deformation vibration of N–H (Amide II)	(peptide bond)	0.7–2.7 kDa	
	1550–1560	60	Deformation vibration of CH <sub>2</sub>	Carboxylates	4.6–6 kDa	
	1450–1460	71	Stretching vibration of C=O	Polysaccharides	High molecular weight	
					16–190 kDa	
					270–275 kDa	

Table 4 (continued)							
Extraction methods	Wave number (cm <sup>-1</sup> )	Transmittance (%)	Vibration type	Functional type	Molecular weight range	References	
t4.34	1400-1410	65	Stretching vibration of OH, of polysaccharide	Phosphorus/sulfur functional group			
t4.35	1060-1100	63	Several bands visible				
t4.36	<1000						
t4.37	3200-3420	60	Stretching vibration of OH	OH into polymeric compounds	Protein molecular weight	Comte et al. (2006a)	
t4.38	1630-1660	70	Stretching vibration of C=O and C-N (Amide I)	Proteins (peptide bond)	45 to 670 kDa	Simon et al. (2009)	
t4.39	1550-1560	52	Stretching vibration of C-N and deformation vibration of N-H (Amide II)				
t4.40	1450-1460	60	Deformation vibration of CH <sub>2</sub>				
t4.41	1400-1410	71	Stretching vibration of C=O	Carboxylates	Polysaccharides		
t4.42	1060-1100	65	Stretching vibration of OH, of polysaccharide	Polysaccharides	<1 kDa		
t4.43	<1000	63	Several bands visible	Phosphorus/Sulfur functional group			
t4.44	3200-3420	30	Stretching vibration of OH	OH into polymeric compounds	Low molecular weight	Comte et al. (2006a)	
t4.45	2935	40	Stretching vibration of C=O and C-N (Amide I)		0.16-0.3 kDa	Simon et al. (2009)	
t4.46	1630-1660	20	Stretching vibration of C-N and deformation vibration of N-H (Amide II)	Proteins (peptide bond)	4.6-6 kDa		
t4.47	1550-1560	25	Deformation vibration of CH <sub>2</sub>				
t4.48	1450-1460	45	Stretching vibration of C=O	Carboxylates	High molecular weight		
t4.49	1400-1410	33	Stretching vibration of OH, of polysaccharide		16-190 kDa		
t4.50	1060-1100	41	Several bands visible	Polysaccharides	270-275 kDa		
t4.51	<1000			Phosphorus or sulfur functional group			
t4.53	EPS: Extracellular polymeric substance; EDTA: Ethylene diamine tetra acetate group; CER: cation exchange resin.						

properties of EPS and their molecular weight (Table 4). Fourier transform infrared spectroscopy (FTIR) was mostly used to identify the functional group of extracted EPS (Omoike and Chorover, 2004; Sheng and Yu, 2006). However, quantitatively this aspect has never been reviewed. Tapia et al. (2009) compared two FTIR spectra of EPS obtained from EDTA and centrifugation method. Significant peaks were visible in both the spectra corresponding to hydroxyl, carbonyl and peptide group bonding. However, the spectrum of EPS extracted with EDTA shows specific bands in the fingerprint region, especially the thick band at 1717 cm<sup>-1</sup>. This band corresponds to the C=O asymmetric stretching vibration of carboxylic acids of EDTA. It has been a general observation by many studies that during extraction with a chemical method the final EPS gets contaminated by the chemical reagent (Comte et al., 2006a). Similar contaminations were observed in other methods like NaOH-formaldehyde (Comte et al., 2006a; Pervaiz and Sain, 2012; Sheng et al., 2005).

In another analysis done by Lee et al. (2013), it was demonstrated that the FT-IR results of CER extracted EPS contained lower content of protein to carbohydrate, fewer acidic functional groups (i.e., COOH or OH groups) as compared to formaldehyde-NaOH technique. The same results were recorded for the EPS extracted from anaerobic granular sludge with physical methods which displayed very similar IR spectra

(D'Abzac et al., 2010). Humic-like substances were hardly identifiable in physical extraction techniques and adsorption bands were intensively present at 2930 and 1650 cm<sup>-1</sup>.

The molecular weight range of EPS varies from 10<sup>3</sup> to 2.5 × 10<sup>6</sup> kDa (Yokoi et al., 1995). The molecular weight of the EPS reported in different studies has been presented in Table 5. To determine molecular weight, size exclusion chromatography (SEC) was frequently used in many studies (Simon et al., 2009; Comte et al., 2006a). These researchers investigated differences appeared in the peak corresponding to the biggest and lowest molecules.

**Table 5 – Metal-binding potential of bacterial EPS and wastewater sludge EPS.**

EPS producer	EPS concentration mg/L	Metal biosorbed	References
<i>Methylobacterium organophilum</i>	184.2	Pb(II)	Kim et al. (1996)
<i>Pseudomonas aeruginosa cur</i>	200.3	Cu(II)	Kazy et al. (2008)
<i>Rhizobium etli M4</i>	320	Cu(II)	Pulsawat et al. (2003)
	67	Mn(II)	

883 Large additional peaks appeared in the chromatograms  
884 recorded for EPS extracted from sludge by cation exchange  
885 resin (CER), heat treatment and centrifugation methods. The  
886 EPS extracted by CER contains more polysaccharides and  
887 uronic acid 210 nm, which indicated that better carbohydrate  
888 content could be extracted by this method. For the EPS  
889 extracted by centrifugation, the highest peak observed had a  
890 lower absorbance at 210 nm representing a lower polysaccharide and uronic acid content.

892 [Simon et al. \(2009\)](#) studied EPS extraction from anaerobic  
893 granular sludge with different methods (heating and centri-  
894 fugation) and proved that the extracted molecules of EPS were  
895 insignificantly affected. According to SEC analysis, two kinds  
896 of differences could be observed on the EPS fingerprints:  
897 variation of the number of detected peaks and significant  
898 evolutions of peak absorbance which corresponds to the high  
899 or low molecular weight.

900 [Domínguez et al. \(2010\)](#) conducted a comparative study  
901 between EPS extracted by physical and chemical methods to  
902 compare their MW distribution using high-pressure size exclu-  
903 sion chromatography (HPSEC). The EPS extracted using chem-  
904 ical methods did not have any effect on the MW distribution  
905 (fingerprints) of EPS or their average MW. Nevertheless,  
906 different physical extractions showed different behavior of EPS  
907 fingerprints. These results were in agreement with the results  
908 of [Alasonati and Slaveykova \(2012\)](#). The study revealed the  
909 effect of the extraction methods (centrifugation, EDTA and  
910 formaldehyde-NaOH) on the size distribution of EPS. According  
911 to [Lee et al. \(2013\)](#), the EPS obtained from aerobic sludge using  
912 CER method were made up of more aromatic and compact  
913 structures possessing higher molecular weight than those  
914 extracted using formaldehyde-NaOH extraction method.

915 Despite extensive efforts to analyze qualitatively the EPS  
916 chemical structure and size distribution, little is known about  
917 the effect of these parameters on EPS properties, functions and  
918 structure, which are essential for understanding the role of EPS  
919 in biofilms and floc formations. In this review, we performed  
920 compositional analyses of the EPS obtained by different extrac-  
921 tion techniques. We also highlighted the effect of extraction  
922 methods on EPS molecular weight. In the next section a  
923 discussion on how the functional group and molecular weight  
924 could interfere and vary their properties as biofloculant, has  
925 been presented.

## 926 6. Potential applications of EPS

### 928 6.1. EPS as adsorbent

929 Heavy metal removal from a polluted environment is a major  
930 challenge in bioremediation, which has been studied extensively  
931 using microbes. EPS produced by many microorganisms are of  
932 particular significance to the bioremediation process because of  
933 their metal ions binding ability from solutions ([Kachlany et al., 2001](#)). The use of purified biopolymers in biosorption phenomenon is a cost-effective and useful approach than using methods like precipitation, coagulation, ion exchange, electrochemical and membrane processes used for metal removal ([Gutnick and Bach, 2000](#)). EPS and other biopolymers exhibit excellent metal-binding properties with varying degrees of specificity and

942 affinity ([Gutnick and Bach, 2000](#)). The effect of and EPS bridging  
943 occur by electrostatic interactions with negatively charged  
944 functional groups such as uronic acids and phosphoryl groups  
945 of carbohydrates or carboxylic groups of amino acids in protein  
946 moiety. Besides, there may also be anionic binding by positively  
947 charged polymers or coordination with hydroxyl groups. EPS  
948 were able to chelate some metals (like  $\text{Th}^{4+}$  and  $\text{Al}^{3+}$ ) and bind  
949 them to the cell surface ([Santamaría et al., 2003](#)).

950 Polysaccharides and protein moieties of EPS, rich in  
951 negatively charged amino acids, including aspartic and  
952 glutamic acid, contribute to their anionic properties, which  
953 play a major role in complexation of metal ions ([Mejare and Bülow, 2001](#)). DNAs are anionic in nature due to the presence  
954 of phosphate groups available in sugar-phosphate backbone  
955 of the molecule. The uronic acids, acidic amino acids and  
956 phosphate-containing nucleotides, act as negatively charged  
957 components of EPS, which are known to bind with multiva-  
958 lent cations by electrostatic interactions ([Beech and Sunner, 2004](#)). Therefore a change in EPS composition will affect the  
959 availability of the functional groups which are responsible for  
960 metal binding and consequently may result in to decrease in  
961 the metal binding efficiency.

962 Numerous bacteria have been shown to produce exo-  
963 polymeric substances. Several studies ([Forster and Clarke, 1983](#); [Prado Acosta et al., 2005](#)) have compared and evaluated  
964 the metal binding potential of microbial biofilms obtained from  
965 activated sludge and purified exopolysaccharides (Table 5).  
966 Over 90% of metal adsorption was achieved at an EPS con-  
967 centration (67 and 160 mg/L) using different bacterial strain  
968 such as *Rhizobium etli* M4 and *Paenibacillus polymyxa* (P13),  
969 respectively. The cells and EPS showed a strong ability to bind  
970 Mn, Pb and Cu ions ([Forster and Clarke, 1983](#); [Nouha et al. 2016](#);  
971 [Prado Acosta et al., 2005](#); [Salehizadeh and Shojaosadati, 2003](#)).

972 Sludge EPS exhibited greater metal complexation, which  
973 suggests that the carboxylic and phosphoric groups in carbo-  
974 hydrate moiety of EPS might have played a major role in the  
975 complexation of metals ([Singh et al., 2000](#)). Few researchers  
976 ([AjayKumar et al., 2009](#); [Comte et al., 2006b](#); [Ruan et al., 2013](#);  
977 [Yuncu et al., 2006](#)) investigated different factors affecting the  
978 metal binding ability to EPS. The metal sorption capacity of the  
979 activated sludge was dependent upon the C/N ratio of sludge.  
980 An increase in C/N ratio (by supplying a carbon source such as  
981 glucose) resulted in an increase in  $\text{Cd}^{2+}$  but decrease in  $\text{Cu}^{2+}$   
982 sorption capacity. The sorption capacity could be explained by  
983 the variation of EPS concentration and composition by using  
984 different C/N ratios. However, the adsorptive capacity of Zn and  
985 Ni was independent of C/N ratio.

986 EPS hydrophobicity is another significant factor, which  
987 favors the sludge flocculation and settling. EPS hydrophobic-  
988 ity can be rendered by EPS-proteins produced by the microbial  
989 communities. According to [Geyik et al. \(2016\)](#), higher protein  
990 content or protein to carbohydrate (P/C) ratio gives higher  
991 EPS hydrophobicity which is correlated with the substrate  
992 provided to the microbial communities. EPS hydrophobicity is  
993 significantly affected by the functional groups in its protein  
994 fraction. The hydrophobicity is an important factor when EPS  
995 is intended to use in organic pollutant removal ([Flemming  
996 et al., 2000](#)). A strong correlation was demonstrated by [Zita  
997 and Hermansson \(1997\)](#) between sludge particle adhesion and  
998 EPS hydrophobicity.  
999

The pH of the surrounding environment affects the deprotonation state of the side residues of the protein fractions present in EPS. At lower pH, the acidic residues are protonated to have higher hydrophobicity. Thus this allows dense intramolecular hydrogen bonding between flocculating particles and further improve sludge compactness and settling efficiency (Wang et al., 2012).

## 6.2. EPS as carbon source

Microorganisms often live and exposed to stressful conditions caused by natural environments, thus the production of EPS augments the survival capacity (Patel and Gerson, 1974). The bacterial EPS was found to be utilized either by self or neighboring microbes during carbon deficiency, using extracellular enzymes and this enzyme complex can be utilized for complete degradation of EPS (Patel and Gerson, 1974; Pirog et al., 1997; Wu and Ye, 2007; Zhang and Bishop, 2003). The study by Pirog et al. (1997) demonstrated a successful utilization of EPS as a carbon source by isolated *Acinetobacter* sp. from a soil sample.

The EPS biodegradability studies were performed by various authors (Wu and Ye, 2007). Zhang and Bishop (2003) observed that carbohydrate fraction of the supplemented EPS as carbon source was consumed more rapidly than the protein fraction. Pannard et al. (2016) investigated and confirmed the biodegradability of EPS by bacteria for growth under nitrogen (or phosphorus) and carbon limiting conditions.

The susceptibility of EPS toward degradation depends on the hydrolyzing agent, which leads to breaking the polymer chains, and also depends on the chemical nature of the polymer (Wingender et al., 1999). Many reports (Neyens et al., 2004; Watson et al., 1987; Watson et al., 2014) have been presented on the influence of sludge treatment in EPS degradation.

Watson et al. (1987) found that protein and carbohydrates, the main component of EPS in activated sludge are degraded or hydrolyzed by heat treatment. Neyens et al. (2004) revealed that heat treatment (120°C) alters the structure of the sludge in term of proteins and carbohydrate moieties of EPS and transforming some of EPS from less degradable to easily degradable. Acid-thermal and alkaline-thermal hydrolysis were also used by Neyens et al. (2004). They indicated an increase in the sludge filtration rate (capillary suction time, CST). Extreme pH (acidity or alkalinity) also causes EPS proteins to lose their natural shapes thus improved their degradability. Polysaccharides and the other components of EPS, are unstable in strong acids, which lead to acid hydrolysis of the glycosidic linkages (Fig. 3) (Wingender et al., 1999). The strong alkali may solubilize gels not only because of chemical degradation but also because of the ionization of the carboxylic groups, which leads to subsequent solubilization of EPS (Wingender et al., 1999). As the EPS is a complex molecule, different treatments could interfere with their chemical structure and even could form a novel compounds. Fig. 3 presents alginate as an example of polymers and their transformation due to several different treatment methods.

Apart from being an excellent bioflocculant, EPS can also act as an adsorbent for heavy metal removal application and source of carbon substrate for various biotechnical applications. To attain more scientific knowledge and potential industrial applications of EPS needs to be explored. It is

also important to have proper EPS production processes with higher yield and properties for desired applications. The first pre-requisite for any process improvement is to understand the underlying biosynthesis mechanism at molecular level. This information can significantly improve and enhance the EPS concentration and quality during a production process by using advanced techniques, which are discussed in the following sections.

## 6.3. Effect of functional group and molecular weight on flocculation activity

The flocculation ability of EPS has been one of the key properties for biopolymer application. Different studies are available which have investigated the important structure-function relationship between EPS functional composition and flocculation abilities. The flocculation activity has been modeled by various mechanisms, and the flocculants activity of high-molecular weight EPS has been explained by the bridge formation model. In the case of Patch model, flocculation of the bacteria with the negatively charged cell surface, is a result of binding of the positively charged macromolecules to the surface of particles Coulomb forces, resulting in neutralization of part of the surface charge (patch model). Reduced electrostatic repulsion leads to agglomeration of particulate matter and formation of flocs by bridges between negatively charged particles (Zhou and Franks, 2006). Fig. 4 illustrates the different mechanisms of flocculation.

No consensus exists on the role of (importance) carbohydrate and protein content of EPS for flocculation. Deng et al. (2005) concluded in his study that EPS containing 76.3% of sugar and 21.6% of protein gave high flocculating abilities of 98.1%.

Freitas et al. (2009) studied the monosaccharides in carbohydrates and concluded that 82.6% flocculation ability was achieved by EPS whose 70% (mol/mol) of carbohydrate was galactose and 23% (mol/mol) was mannose. When monosaccharide percentage decreased in carbohydrate fraction, the flocculation abilities were seen to be decreasing as observed by Kavita et al. (2014) only 40% flocculation ability was achieved when mannose (47.8%) and glucose (29.7%) were present in lower quantity. Li et al. (2008) emphasized the importance of acetyl groups on flocculation ability. The study showed that EPS with acetyl group shows a good flocculation (49.3%) comparing to deacetylate EPS (27.8%).

A more functional level analysis of flocculating abilities was performed by various studies (Deng et al., 2005; Kavita et al., 2014) to understand the importance of functional groups on flocculation ability. They investigated that cations stimulate flocculation by neutralization and stabilization of residual negative charges of the carboxyl group of a bioflocculant forming bridge that binds kaolin particles to each other. Further, the negatively charged carboxyl group (COO<sup>-</sup>) of the bioflocculant could bind with the positively charged site of the suspended kaolin particles.

Although flocculation ability of EPS seems to be sensitive to the carbohydrate content (Shin et al., 2001), a study conducted by Yu et al. (2009) concluded that protein fraction of EPS is the most important parameter for flocculation activity. Researchers advocating for protein suggest that negatively charged amino

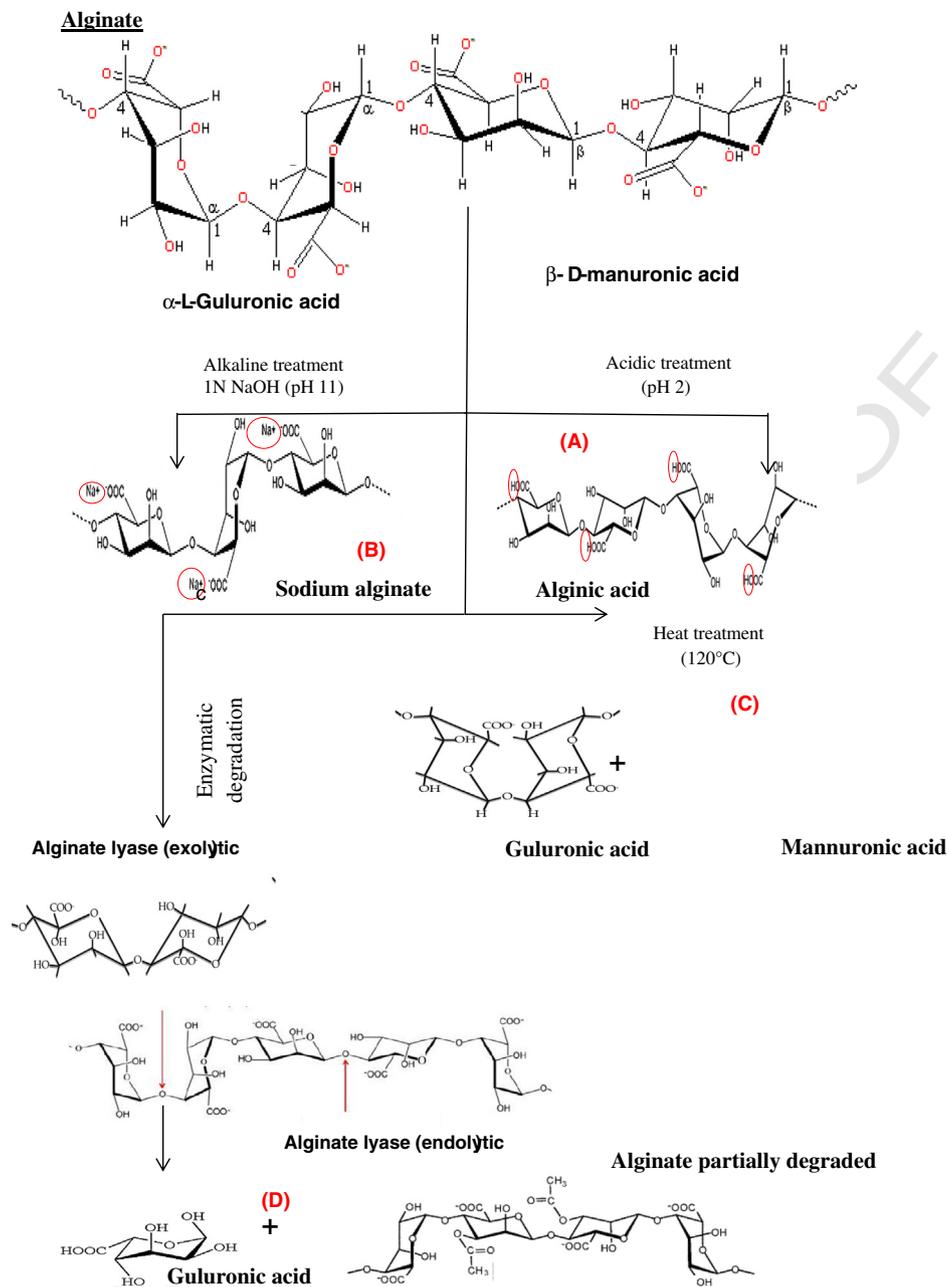


Fig. 3 – Alginate degradation: (A) acidic treatment; (B) alkaline treatment; (C) heat treatment and (D) enzymatic degradation.

1116 acid contribute to flocculation abilities. The hydrogen bonds are  
1117 present frequently in proteins and they could affect the capacity  
1118 of bioflocculant to agglomerate.

1119 The relationship between molecular weight and flocculation  
1120 activity of bioflocculant remains unclear until now.  
1121 Flocculation with high molecular weight bioflocculant involves  
1122 more adsorption points, stronger bridging, and higher  
1123 flocculating activity than the flocculation with a low-  
1124 molecular-weight bioflocculants. Larger molecular weight  
1125 flocculants usually has a sufficient number of free functional  
1126 groups, to form bridges to bring many suspended particles  
1127 together, and hence produce a larger floc size in the flocculating  
1128 reaction (Shih et al., 2001). These results were in agreement

with the findings of many researchers (Deng et al., 2005; He et al., 2002; Wu and Ye, 2007). 1129

High molecular weight ( $2.6 \times 10^6$  Da) bioflocculant produced by *Bacillus mucilaginosus* revealed high flocculation activity (99%) compared to low MW ( $10^5$  Da) obtained from *Corynebacterium glutamicum* (80% of flocculation activity) (Deng et al., 2003; He et al., 2002). Furthermore, the bioflocculant produced by *Bacillus subtilis* DYU500 ( $3.20 \times 10^6$  Da) in the study of Wu et al. (2014) seems to favor the performance of flocculation (97%) comparing to 90% obtained by *Gyrodinium impudicum* KG03 (MW  $1.58 \times 10^6$  Da) (Yim et al., 2007). 1139

Recently, Tang et al. (2014) discovered a new bioflocculant produced by *Enterobacter* sp. ETH-2. The MWs of ETH-2 ranges 1140 1141

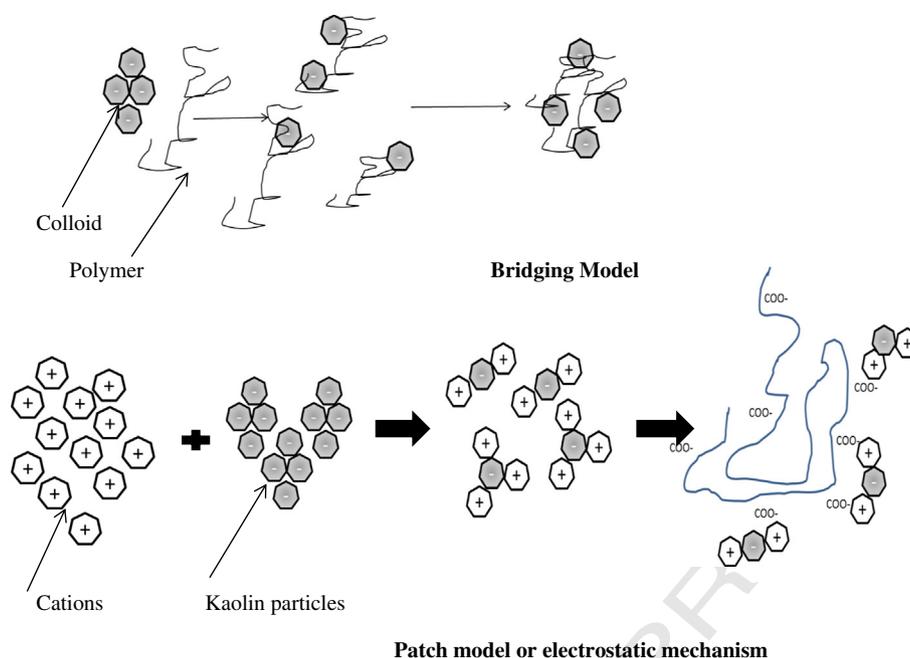


Fig. 4 – Flocculation mechanism of bioflocculant.

1142 from 603 to 1820 kDa, which is within the high MWs range with  
 1143 high flocculation ability of ETH-2 (94%) as compared to other  
 1144 strains. For example, *Bacillus megaterium* TF10 produced  
 1145  $2.5 \times 10^3$  kDa with 95.5% capacity to flocculate (Xiong et al.,  
 1146 2010) and 90% of bioflocculation was obtained by *B. licheniformis*  
 1147 with  $1.8 \times 10^3$  kDa molecular weight (Yuan et al., 2010).

1148 In summary, bioflocculant produced by the various isolated  
 1149 microorganism are very diverse corresponding to their chemi-  
 1150 cal structure (functional group) and their molecular weight as  
 1151 discussed earlier. These parameters affect their properties of  
 1152 EPS as a bioflocculant agent. High MW bioflocculant possess  
 1153 more adsorption points for bridging thus larger and stable flocs  
 1154 are obtained. However, the higher bioflocculant concentration  
 1155 were required to achieve these results. The concentration of EPS  
 1156 used to obtain the highest flocculation activity is important  
 1157 from an economic standpoint.

## 1158 7. Conclusion and recommendations

1160 As evident from the review that sludge EPS remains to be an  
 1161 unexplored field of study with a plethora of research opportu-  
 1162 nities for many industrial and eco-friendly applications. EPS is  
 1163 composed of mainly carbohydrates and proteins, and they play  
 1164 very significant role in determining their functionality. The  
 1165 presence of nucleic acid and humic acid substances as a  
 1166 resultant of cellular lysis during post-production processing  
 1167 can further contribute to enhance the functional properties of  
 1168 EPS. The microorganism, carbon substrate, and other growth  
 1169 conditions play very significant role in determining EPS  
 1170 composition. EPS has great potential to be used as metal  
 1171 removing agent in mining industries, as a flocculating agent in  
 1172 WWTP for sludge dewatering and as a carbon source for  
 1173 biotechnological production of other metabolites.

Progress in molecular level knowledge about EPS production, 1174  
 its genetics, and enzymology have been very limited. The limited 1175  
 knowledge has restricted the application of various engineering 1176  
 techniques to enhance biological production of EPS. A huge scope 1177  
 of research and development lies in developing mutant strains to 1178  
 have a higher titer of EPS with novel properties. There are no 1179  
 dedicated industrial production processes for EPS production. 1180  
 This technical development has been limited by the optimization 1181  
 of EPS production using mixed cultures and pure cultures in 1182  
 combination with various kinds of carbon substrate which 1183  
 significantly affect the functional properties. More dedicated 1184  
 studies are required toward optimization of EPS production 1185  
 processes using novel and cheaper carbon substrates like sludge. 1186  
 A dedicated process development and simulation are required to 1187  
 have high upstream production coupled with efficient down- 1188  
 stream extraction. Proper characterization and documentation 1189  
 of the effect of extraction processes on EPS concentration, 1190  
 composition and functional groups (functionality) are required 1191  
 to realize a large scale EPS production process. 1192

The research on sludge EPS still lacks the clarity on the 1193  
 mechanism of EPS production, proteome involved in EPS 1194  
 biosynthesis, mechanistic knowledge of role and effect of 1195  
 various components of EPS toward its functionality is still 1196  
 missing and very limited research is available. Currently 1197  
 many studies are going on EPS production, but there is a lot 1198  
 more to know about EPS than what has been known to make 1199  
 EPS a successful commercial product of application. 1200

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