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Project: Carbonate production by sequestration of industrial CO₂: revalorization of mine and industrial waste

Mineral Carbonation: what is it?

A Natural Process

The CO₂ from the atmosphere reacts with rocks and minerals
This reaction gives another solid: stable, inert, non-toxic
→ Perfect form to permanently and safely sequester CO₂

Example of Mg-bearing rock : serpentine
Magnesium carbonate: Magnesite

Example of Ca-bearing rock : anorthosite
Calcium carbonate: Calcite

Efficient but slow kinetics in nature

Solution?
→ Accelerating the reaction rate

How?
→ Various paths: dry carbonation, aqueous carbonation
→ Optimizing the reaction parameters
→ Working with various materials (wastes, rocks and minerals)
→ Enhancing the reactivity of the material

In a reactor

CO₂ bearing gas mixture + ground mine/industrial wastes →
Favorable conditions → ≈ 80% CO₂ capture → Carbonates

Waste concrete, Serpentine tailings, Anorthosite tailings, Grinding, Carbonation, Carbonates

The Mineral Carbonation Process: useful?

In industry...

The process use gaseous effluents and as such :

- no need of a pre-concentration or capture step
- no need of 100% CO₂ gas or high pressure
- possible with a mixture of CO₂ (up to 30%), H₂O, N₂, NO_x, SO_x...
- At least 80 % of the CO₂ emissions is stored at low cost.

The process would be applicable to large CO₂ emitters like in the steel, aluminum, copper, cement and petroleum industries...

And waste management

Mine waste	Industrial waste
Use of tailings:	Use of industrial residue:
→ Decrease size of dump	→ Decrease amount of discharge
→ Revalorization and added value to an otherwise waste material	→ Decrease sizes of disposal site
	→ Revalorization and added value
<u>Feed examples:</u> Chromite, ilmenite, talc, soap stone, nickel exploitations tailings, etc.	<u>Feed examples:</u> Waste concrete and cement, various bricks and granulates, etc.

Is it that easy?

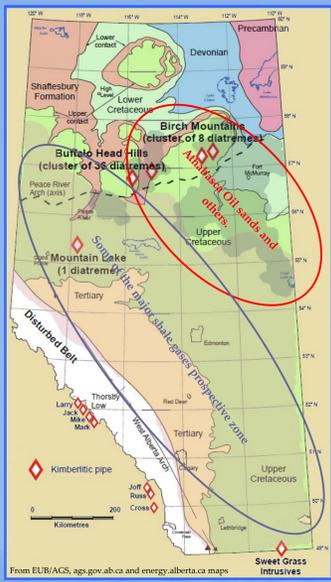
Challenges

- ⚠ Convincing the companies of the efficiency of the process.

This comes with:

- Scaling up of the process on site
- Installing the pilot plant on an industrial site
- ⚠ Having the material close to industrial sites/major GHG emitters:
 - Various materials tested : wider possibilities (sandstone, granite, basalt, peridotite, etc.)
- ⚠ Acceptance from the public

Applicability to Alberta



Materials available:	Industries
→ Kimberlite (See map)	→ Steel, tubing and alloy plants
→ Mafic sills and lava (Southern Alberta Rockies)	→ Cement, aggregate and ready mix plants
→ Mafic rocks (NE Alberta)	→ Oil refineries, petrochemical producers and plastic manufacturers...
→ Waste concrete and aggregates (Airdrie, Mountain View...)	
→ And possibly sandstone, granite...	
	<u>Localization:</u> Calgary, Edmonton, Red Deer, Sylvan Lake, Fort McMurray, Scotford...

Conclusion

Use of gaseous and solid wastes: CO₂ emitted + mine/industrial wastes → Lower CO₂ emissions + recycle wastes + value-added/ environmentally safe product : Carbonates