



# Carbon dynamics in highly heterotrophic subarctic thaw ponds

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**Abstract.** Global warming has accelerated the formation of permafrost thaw ponds in several subarctic and arctic regions. These ponds are net heterotrophic as evidenced by their greenhouse gas (GHG) supersaturation levels (CO<sub>2</sub> and CH<sub>4</sub>), and generally receive large terrestrial carbon inputs from the thawing and eroding permafrost. We measured seasonal and vertical variations in the concentration and type of dissolved organic matter (DOM) in five subarctic thaw (thermokarst) ponds in northern Quebec, and explored how environmental gradients influenced heterotrophic and phototrophic biomass and productivity. Late winter DOM had low aromaticity indicating reduced inputs of terrestrial carbon, while the high concentration of dissolved organic carbon (DOC) suggests that some production of non-chromophoric dissolved compounds by the microbial food web took place under the ice cover. Summer DOM had a strong terrestrial signature, but was also characterized with significant inputs of algal-derived carbon, especially at the pond surface. During late winter, bacterial production was low (maximum of 0.8 mg C m<sup>-3</sup> d<sup>-1</sup>) and was largely based on free-living bacterioplankton (58 %). Bacterial production in summer was high (up to 58 mg C m<sup>-3</sup> d<sup>-1</sup>), dominated by particle-attached bacteria (67 %), and strongly correlated with the amount of terrestrial carbon. Primary production was restricted to summer surface waters due to strong light limitation deeper in the water column or in winter. The phototrophic biomass was equal to the heterotrophic biomass, but as the algae were mostly composed of mixotrophic species, most probably they used bacteria rather than solar energy in such shaded ponds. Our results point to a strong heterotrophic energy pathway in these thaw pond ecosys-

tems, where bacterioplankton dominates the production of new carbon biomass in both summer and winter.

## 1 Introduction

Traditional view of inland waters as sinks of carbon has changed during the past decades, and, at present, lakes and ponds are considered net sources of carbon to the atmosphere (Tranvik et al., 2009). Increasing attention has been given to thaw ponds and lakes after recognizing the cumulative effect of their high abundance and greenhouse gas (GHG) emissions on global warming (Kling et al., 1992; Walter et al., 2006; Laurion et al., 2010), especially when they emit old carbon thus having the potential to act as a positive feedback mechanism on climate (Walter Anthony et al., 2014). Thaw ponds and lakes are dominant in continuous and discontinuous permafrost areas, for example in permafrost regions of Siberia where they represent 90 % of all lakes (Walter et al., 2006), but until recently little has been known on the limnological properties and microbial communities of this important freshwater ecosystem (Vonk et al., 2015).

Subarctic thaw ponds can be highly turbid and have relatively high nutrient concentrations (Breton et al., 2009; Rautio et al., 2011) compared to most high-latitude freshwater ecosystems (Pienitz et al., 1997; Hamilton et al., 2001; Medeiros et al., 2012). Moreover, high perimeter/volume ratio, the presence of palsas and bogs and the thawing of organic-rich permafrost in their watershed favor high inputs of dissolved organic matter (DOM) to the ponds. Chromophoric DOM (CDOM) can influence water column tem-

perature regimes through its absorption of sunlight (Caplanne and Laurion, 2008), and with suspended particles, they contribute to form a distinct and stable thermocline (Laurion et al., 2010). The rate of DOM input to aquatic systems has been documented to increase in boreal lakes over the last decades (Hudson et al., 2003), and while no such information is yet available for thaw ponds, the recent mobilization of terrestrial carbon stocks stored for thousands of years in the permafrost (Vonk et al., 2012) suggests a similar DOM increase is taking place also in the north.

High DOM concentration also strongly regulates the depth of the euphotic zone. Light is usually limited to the first meter in thaw ponds (Squire and Lesack, 2003; Watanabe et al., 2011), therefore restricting the water volume where photosynthesis can occur. This type of environment favors heterotrophic bacterial production (BP) that was shown to reach similar rates as in eutrophic lakes (Breton et al., 2009). The limnological characteristics of thaw ponds are likely also beneficial to mixotrophic algae and heterotrophic protozoans, as flagellated species are generally well adapted when lakes are rich in humic compounds (Jones, 2000). In low light conditions they can utilize bacteria as an energy source, and as they are motile they also benefit from the nutrient-rich bottom waters (Arvola et al., 1991). They have been shown to be a key link in the carbon transfer from low to high trophic levels in small boreal forest lakes (Salonen and Rosenberg, 2000).

Circumpolar thaw ponds are ice covered for a large fraction of the year, which further affects carbon cycling and generates large seasonal variations. The aim of this study was to demonstrate the difference in GHG accumulation, CDOM characteristics, and autotrophic and heterotrophic carbon pools between two seasons and depths. Comparisons were made between late winter and summer, and between surface and bottom waters in summer. We hypothesized that in the absence of light in winter or in the aphotic summer hypolimnion, autotrophic productivity stops while respiration continues, resulting in large summer and winter storage of GHG in the water column. We further hypothesized that the thermal structure in summer generates an accumulation of organic carbon in the hypolimnion where heterotrophy dominates. This study follows earlier ones looking at the limnological characteristics and bacterial communities of subarctic thaw ponds in summer (Breton et al., 2009; Negandhi et al., 2013; Rossi et al., 2013; Crevecoeur et al., 2015).

## 2 Methods

### 2.1 Sample collection and limnological analyses

The study was carried out in the discontinuous permafrost region near the village of Whapmagoostui-Kuujuarapik (55°20' N, 77°30' W), northern Quebec, Canada, along the River Kwakwatanikapistikw (local Cree name; KWK is used

to name the ponds hereafter) where a large number of thaw ponds can be found. At this site, ponds are thermokarstic and form over an impermeable clay–silt bed, on the center of thawed lithalsas (or inorganic palsas), preventing most hydrological interactions among ponds. They are surrounded by dense shrubs and sparse trees, and occasional areas of mosses and aquatic plants. More details on the study area are provided in Bouchard et al. (2011, 2014).

Five ponds were studied during late winter (6–11 April) and summer (8–14 August) in 2009. They are all small (averaged diameter  $17 \pm 1.3$  m) and shallow (average depth  $2.5 \pm 0.4$  m). Sampling was done with a Kemmerer water sampler (Wildco® USA) through a borehole drilled into the ice in winter, and from an inflatable boat in summer at the middle of the pond. During winter sampling, the water started pouring over the ice through the hole because of the pressure from the weight of snow and ice, and possibly resulting in a mixing of surface with deeper water. This was suggested as the factor explaining the similar GHG concentrations obtained in surface (just under the ice) and bottom waters (approximately 0.5 m above the sediments), despite the expected inverse thermal stratification at this period of the year (Laurion et al., 2010). All other samples were collected from about 1 m below the ice and considered representative of an integrated water column. In summer, surface and bottom waters were sampled from just below the surface and approximately 0.5 m above the bottom sediments.

Measurements of temperature, dissolved oxygen, pH and conductivity were done with a 600R YSI multiparametric probe (Yellow Springs Inc.). Further analyses were performed on water prefiltered through a 50 µm sieve. Total suspended solids (TSS) were obtained by filtering 500–1000 mL of water onto precombusted and preweighted GF/F filters (Advantec MFS Inc.). The 50 µm prefiltration reduced the TSS by an average of 5.8% in comparison to bulk water samples, both analyzed in summer (data not shown). Filters were stored at  $-20^{\circ}\text{C}$  until they were dried for 24 h at  $60^{\circ}\text{C}$ , and weighted to obtain sestonic dry weight or TSS. They were subsequently combusted at  $450^{\circ}\text{C}$  for 2 h to calculate the organic fraction of TSS. An aliquot of water was filtered through a pre-rinsed cellulose acetate filter (0.2 µm pore size; Advantec MFS Inc.) to analyze soluble reactive phosphorus (SRP; unpreserved samples) and dissolved iron (Fe; samples preserved with  $\text{HNO}_3$  at 0.15% final concentration). Total phosphorus (TP) and total nitrogen (TN) were analyzed from unfiltered water preserved with  $\text{H}_2\text{SO}_4$  (final concentration of 0.15%). All nutrient samples were stored in acid-washed glass bottles in a dark and cold ( $4^{\circ}\text{C}$ ) room until further analysis as in Breton et al. (2009).

### 2.2 Carbon characterization

Dissolved  $\text{CO}_2$  and  $\text{CH}_4$  were determined as in Laurion et al. (2010). In brief, 2 L of pond water was equilibrated into 20 mL of ambient air for 3 min, with the headspace sampled

in duplicated glass vials (Vacutainers<sup>®</sup>) previously flushed with helium and vacuumed. Gas samples were analyzed by gas chromatography (Varian 3800), and dissolved gas concentrations were calculated according to Henry's law.

The water filtered through pre-rinsed cellulose acetate filters as above was also used to analyze dissolved organic carbon concentration (DOC), and for optical analyses of CDOM. Samples were stored in amber glass bottles in a dark and cold (4 °C) room until the analysis. DOC was quantified using a carbon analyzer (Shimadzu TOC-5000A) calibrated with potassium biphthalate. The absorbance was measured from 200 to 800 nm with a spectrophotometer (Varian Cary 300), using 1 cm quartz cuvettes on dual-beam mode, at 1 nm intervals for winter samples, and 4 nm intervals for summer samples. Null-point adjustment was performed using the mean value from 750 to 800 nm, and the absorption coefficients ( $a_\lambda$ ) were calculated from absorbance measurements ( $A_\lambda$ ) at 254 and 320 nm using  $a_\lambda = 2.303 A_\lambda/L$ , where  $L$  is the length of the cuvette in meters (Mitchell et al., 2002). The absorption coefficient at 320 nm ( $a_{320}$ ) was used as an index of CDOM concentration. Specific UV-absorbance index (SUVA<sub>254</sub>) was determined from DOC normalized  $A_{254}$ , and used as an index of aromaticity (Weishaar et al., 2003).

CDOM absorption spectra were further analyzed to calculate the spectral slopes  $S_\lambda$  (Loiselle et al., 2009) between 250 and 450 nm (Galgani et al., 2011). Slopes were calculated over 20 nm intervals with a 1 nm step (i.e., 250–269, 251–270, etc.) for winter samples, and 4 nm steps for summer samples (i.e., 250–269, 254–273, etc.). The resulting set of spectral slopes was plotted by center wavelengths. Calculations were performed in open-source software package SciLab 4.15. The individual spectral slope  $S_{289}$  was used to evaluate the amount of fulvic and humic acids likely related to the autochthonous production (Loiselle et al., 2009). Algal-derived carbon has a maximum slope at 289 nm, while terrestrial carbon spectral slopes lack this peak and present a spectrally variable distribution with a steady increase from 260 to 390 nm.

### 2.3 Microbial abundance and productivity

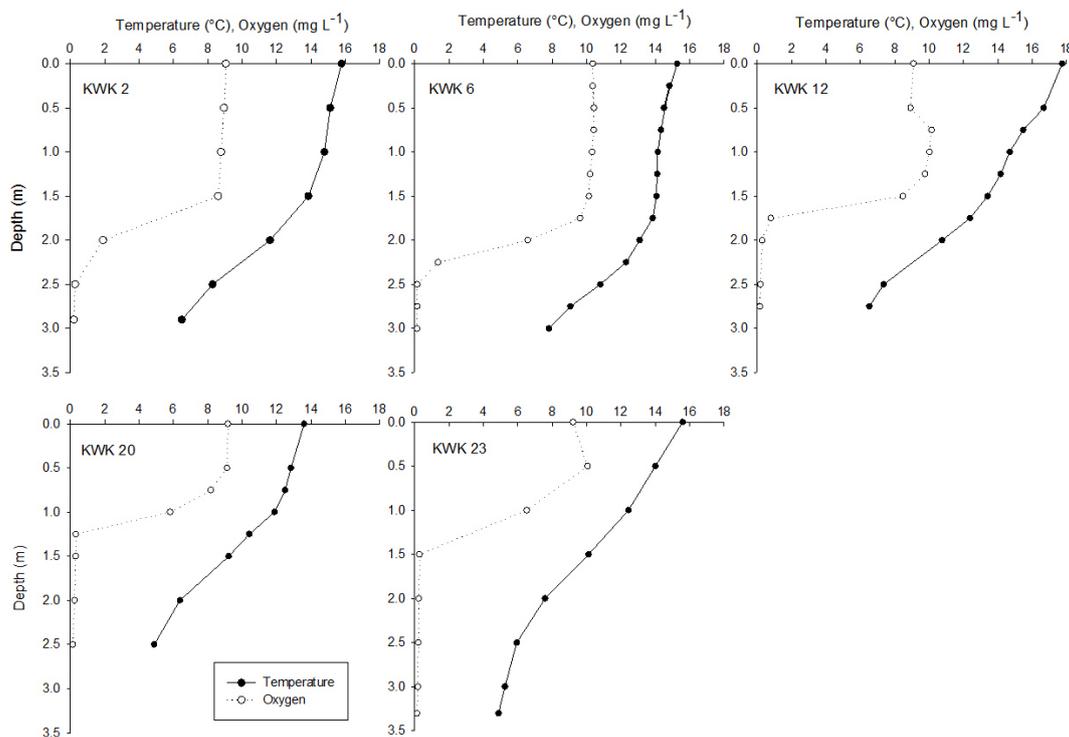
For the determination of chlorophyll *a* (Chl *a*) concentrations, 1–2 L of 50 µm sieved pond water was filtered onto GF/F filters. Samples were collected in duplicates and stored at –80 °C until pigment extraction, and fluorometric analysis were done in hot ethanol according to Nush (1980). Water samples for the enumeration of nanoflagellates (PNFs: phototrophic and mixotrophic cells; HNFs: heterotrophic cells) and phototrophic picoplankton (PPA) were preserved with 0.2 µm filtered glutaraldehyde solution at a final concentration of 1 %. Three replicate aliquots (3–5 mL) were filtered through 0.6 µm black polycarbonate filters (Nuclepore). Samples were stained with 4-,6-diamido-2-phenylindole (DAPI), mounted to slides and stored at –20 °C until counting under UV excitation with an epifluo-

rescence microscope (Zeiss Axiovert 200) at 1000× magnification. Discrimination between pigmented and heterotrophic nanoflagellates was done with green excitation light. At least 50 cells or a maximum of 30 fields were counted from each sample. Nanoflagellate cells were further divided into three size categories: <5, 5–10 and >10 µm. Volumes were converted to carbon biomass using the carbon content coefficient of 0.19 pg C µm<sup>–3</sup> for PNFs and HNFs (Putt and Stoecker, 1989). Phototrophic picoplankton carbon content was calculated using the carbon conversion equation (pg C = 0.433 × V<sup>0.866</sup>, where V is in µm<sup>3</sup>) for picoeukaryotic cells (Verity et al., 1992; Campbell et al., 1994).

Bacterial abundance (BA) was measured with a flow cytometer (FACSCalibur, Becton-Dickinson) as in Rossi et al. (2013) before and after sonication to obtain estimations of free-living and particle-attached bacterial abundance. Biomass calculation for bacterioplankton was based on biovolumes measured from digital images acquired from DAPI-stained microscope slides using the Cell C program (Selinummi et al., 2005). Average bacterial carbon content was first estimated with allometric conversion formula recommended for DAPI-stained cells (fgC = 218 × V<sup>0.86</sup>, where V is in µm<sup>3</sup>; Posch et al., 2001). The average cellular carbon content (18.1 fg C cell<sup>–1</sup>) was then multiplied by the bacterial abundance.

Bacterial production was measured from unfiltered and filtered (3 µm) water samples using tritiated leucine (<sup>3</sup>H-leucine) incorporation with centrifugation (Smith and Azam, 1992). A solution of <sup>3</sup>H-leucine (specific activity of 164 Ci mmol<sup>–1</sup>) was added to each vial to obtain a final leucine concentration of 30 nM, at which bacteria were saturated (experimentally tested). During the winter, the samples were incubated at ~0 °C in the dark for 2 h. During the summer, samples were incubated at in situ temperatures (5–18 °C), depending on ponds. BP was stopped by adding trichloroacetic acid (5 % final concentration), and samples were then stored at –20 °C before the centrifugation step. Winter samples were radioassayed with a Perkin Elmer Tri-Carb 2800, and summer samples with a Beckman LS 6500. Leucine incorporation was converted to carbon biomass using coefficients from Simon and Azam (1989). Bacterial respiration (BR) was estimated from BP according to Del Giorgio and Cole (1998) BR model II.

Primary production (PP) was measured from the mixed water column during winter and from the surface in summer. An incubation system generating a gradient of photosynthetically active radiation (PAR), introduced by Rae and Vincent (1998), was used to obtain photosynthesis versus irradiance curves from complete darkness to full sunlight ( $E_{\max}$ ). The value for  $E_{\max}$  was calculated from in situ PAR measurements available from a weather station located in Whapmagoostui-Kuujuarapik, and assuming the air–water surface in these humic waters to cause a 50 % decrease in radiation levels (Doxaran et al., 2004). A water volume of 20 mL was inoculated with a working solution



**Figure 1.** Summer temperature and oxygen profiles from sampled ponds.

of  $[^{14}\text{C}]\text{-HCO}_3^-$  ( $80\ \mu\text{Ci mL}^{-1}$ ) to a final concentration of  $0.2\ \mu\text{Ci mL}^{-1}$ . In winter, the incubations were done outside ( $\sim 0^\circ\text{C}$ ) under natural light conditions in a water bath for 45–60 min, while in summer a circulating water bath was used to keep samples at in situ temperature ( $14\text{--}18^\circ\text{C}$ ) and incubations were run for 120 min. The incubations were terminated by filtrating samples onto GF/F filters, the method only capturing  $^{14}\text{C}$  fixed in particulate organic matter and not the carbon allocated to exudates. The filters were subsequently frozen before they were radioassayed with a Perkin Elmer Tri-Carb 2800 liquid scintillation counter. Chl *a* normalized carbon fixation rates were fitted to the equation of Platt et al. (1980) or Jassby and Platt (1976) depending on the presence or absence of photoinhibition, respectively. Data fit to the equations had on average a  $r^2$  value of 0.92, and always higher than 0.81. The iterative non-linear regression of SigmaPlot 11.0 was used to obtain the maximum photosynthetic rate ( $P_{\text{max}}$ ). PP at the pond bottom was estimated from the photosynthesis-irradiance curves and PAR estimated from in situ surface measurements, and diffuse attenuation coefficients were calculated from a linear regression using TSS and DOC (Watanabe et al., 2011). Without light under the ice and snow cover (measured with a Li-Cor Li-192 submersible PAR quantum sensor), PP was established at zero.

## 2.4 Data analysis

Differences in environmental and microbial variables among waters were tested using a one-way ANOVA (fixed factor; JMP software, version 11.0.0, SAS Institute Inc. 2013). Conductivity, BP, PPA, SRP, and PNF biomass were log-transformed, while TP, TN, and BB were square-root transformed to achieve normality and homogeneity of variance in residuals (visually checked with predicted values). When differences were detected, a Tukey HSD multiple comparison test was performed. Environmental variables are illustrated with a principal component analysis (PCA; PAST version 2.17c), where biological variables were entered as vectors which length and direction were determined by Spearman's correlation analysis between biological variables and PCA axis 1 and 2. Correlations between organisms (abundance, biomass and productivity measurements) and environmental variables were tested with non-parametric Spearman's rank correlation coefficient analysis (IBM SPSS statistic 20.0). A significant level of  $\alpha = 0.05$  was used for all statistical tests.

## 3 Results

### 3.1 Physicochemical properties of thaw ponds

In late winter, the ponds were covered with 50 to 70 cm of snow and approximately 30 to 60 cm of ice, which absorbed all light. Water temperature was  $2.0^\circ\text{C}$  under the ice,

**Table 1.** Limnological properties of the subarctic ponds sampled in late winter and during summer at the surface and bottom of the water column, including pH, conductivity (Cond.), total phosphorus (TP), total nitrogen (TN), soluble reactive phosphorus (SRP), total suspended solids (TSS), particulate organic carbon (POC) and dissolved iron (Fe).

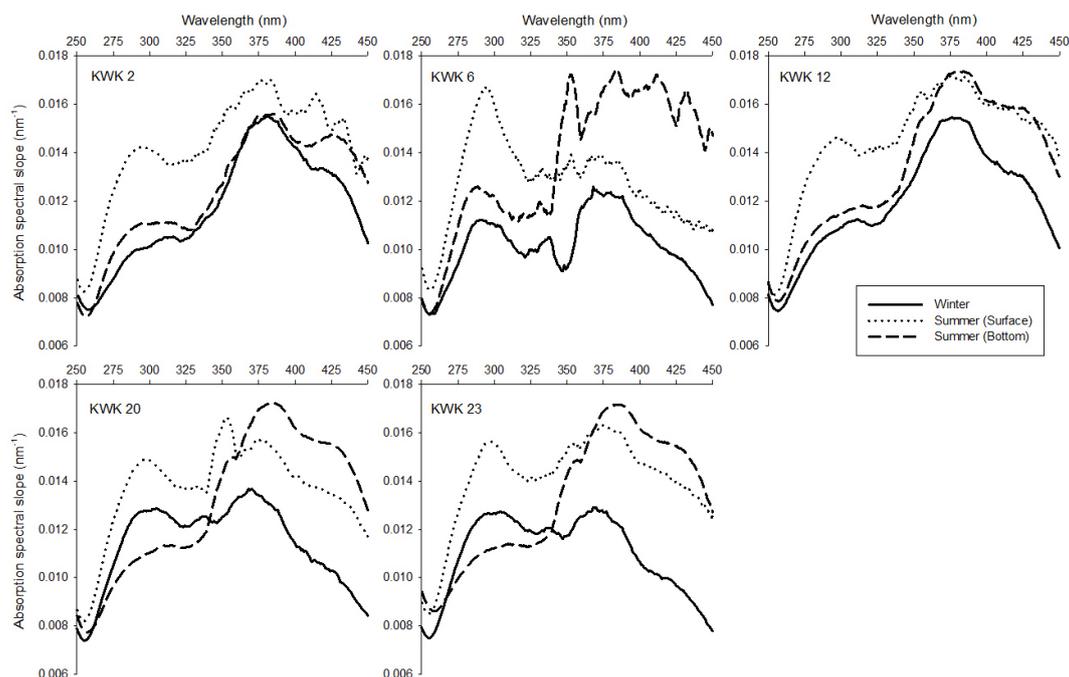
Site	pH	Cond. ( $\mu\text{S cm}^{-1}$ )	TP ( $\mu\text{g L}^{-1}$ )	TN ( $\mu\text{g L}^{-1}$ )	SRP ( $\mu\text{g L}^{-1}$ )	TSS ( $\text{mg L}^{-1}$ )	POC ( $\text{mg L}^{-1}$ )	Fe ( $\text{mg L}^{-1}$ )
Winter								
KWK 2	6.6	60	61.9	957.4	20.8	2.7	0.8	na
KWK 6	6.8	49	50.6	687.8	7.7	3.7	1.1	na
KWK 12	6.6	46	73.2	1038.4	18.4	4.1	1.2	na
KWK 20	6.7	52	180.7	1301.6	47.1	53.9	16.6	na
KWK 23	6.4	42	119.3	802.7	33.9	23.5	6.8	na
Summer surface								
KWK 2	7.4	38	34.9	289.3	1.2	3.4	0.9	0.3
KWK 6	6.9	61	43.9	227.9	0.9	7.2	2.3	0.2
KWK 12	7.3	36	23.9	311.9	0.7	2.6	0.9	0.3
KWK 20	7.5	47	92.6	263.4	7.1	14.2	1.8	0.5
KWK 23	7.1	48	78.0	227.9	5.7	16.1	2.1	0.4
Summer bottom								
KWK 2	6.2	200	341.5	496.1	1.5	27.7	4.4	2.5
KWK 6	6.2	265	197.7	389.4	na	13.6	2.3	1.1
KWK 12	6.1	247	207.1	447.6	1.0	37.0	8.8	2.9
KWK 20	6.0	155	377.0	289.3	73.8	85.8	5.7	5.1
KWK 23	6.2	102	431.8	266.7	45.7	126.8	7.8	4.1

and the ponds were likely  $\text{O}_2$  depleted at the bottom because of the smell of hydrogen sulfide, while  $\text{O}_2$  concentration was  $4.5 \pm 3.5 \text{ mg L}^{-1}$  in the integrated water column sample. In summer, despite their shallow depth, ponds are strongly stratified with nearly  $10^\circ\text{C}$  difference in temperature and  $10 \text{ mg L}^{-1}$  difference in  $\text{O}_2$  between surface and bottom (Fig. 1). Some ponds (KWK 2, 6 and 20) had a thermally homogenous epilimnion until 1 to 1.5 m depth, while in KWK 12 and 23 the temperature declined nearly linearly towards the bottom with a rate of  $1\text{--}2^\circ\text{C}$  every 50 cm. The pH varied between 6.0 and 7.5, with lowest and highest values respectively measured in summer bottom and surface waters. There was no difference ( $p = 0.9034$ ) in conductivity between winter ( $50 \pm 7 \mu\text{S cm}^{-1}$ ) and summer at the surface ( $46 \pm 10 \mu\text{S cm}^{-1}$ ), but bottom waters had significantly ( $p < 0.009$ ) higher conductivity ( $194 \pm 67 \mu\text{S cm}^{-1}$ ) than winter and summer surface waters. Nutrients were relatively high in all samples. TN concentrations were elevated in winter ( $958 \pm 235 \mu\text{g N L}^{-1}$ ) compared to the summer values ( $264 \pm 118 \mu\text{g N L}^{-1}$ ;  $p = 0.0001$ ). TP was relatively high in winter ( $97 \pm 43 \mu\text{g P L}^{-1}$ ), and in summer the concentrations were clearly lower at the surface ( $55 \pm 29 \mu\text{g P L}^{-1}$ ) than at the bottom ( $3110 \pm 1043 \mu\text{g P L}^{-1}$ ;  $p = 0.0001$ ). SRP was available during the whole year, with higher concentrations in late winter ( $26 \pm 15 \mu\text{g P L}^{-1}$ ) and summer bottom waters ( $31 \pm 36 \mu\text{g P L}^{-1}$ ) than in summer surface where concentrations were relatively low ( $3.1 \pm 3.0 \mu\text{g P L}^{-1}$ ) but not statistically different from the other values ( $p = 0.064$ ). All ponds had higher TSS in bottom waters ( $85 \pm 43 \text{ mg L}^{-1}$ ),

while in winter and summer surface waters the concentrations were generally an order of magnitude lower. In most ponds, organic fraction of TSS followed the same trend as TSS and made on average 24 % of TSS. Fe was lower at the surface ( $0.3 \pm 0.1 \text{ mg L}^{-1}$ ) than at the bottom of ponds ( $3.1 \pm 1.6 \text{ mg L}^{-1}$ ) in summer. No Fe measurements were done during late winter. Limnological properties of the ponds are summarized in Table 1.

### 3.2 Carbon characterization

$\text{CO}_2$  and  $\text{CH}_4$  concentrations were high in winter (up to 357 and  $14.1 \mu\text{M}$  respectively), and particularly high at the pond bottom in summer (up to 815 and 312 respectively; Table 2). Surface waters, in contact with the atmosphere, had 1 order of magnitude lower concentrations, but values remained supersaturated. The inverse relationship ( $r = -0.9228$ ;  $p < 0.0001$ ) between  $\text{O}_2$  and  $\text{CO}_2$  suggests that the  $\text{CO}_2$  originated from  $\text{O}_2$  respiration and not from external sources such as weathering, runoff and groundwater inputs that are unlikely in the studied ponds. DOC ranged from 4.1 to  $10.5 \text{ mg L}^{-1}$ , with the highest concentrations obtained in winter. Despite the higher DOC observed in winter, CDOM was lowest at this period, as indicated by  $a_{320}$  values ( $7.1 \pm 2.5 \text{ m}^{-1}$  compared to  $27 \pm 10$  and  $70 \pm 37 \text{ m}^{-1}$  at the surface and bottom in summer, respectively).  $\text{SUVA}_{254}$  followed the same trend as  $a_{320}$ , with lower values in winter ( $0.8 \pm 0.2 \text{ L mg C}^{-1} \text{ m}^{-1}$ ) compared to summer surface ( $4.8 \pm 1.0 \text{ L mg C}^{-1} \text{ m}^{-1}$ ) and bottom waters ( $7.7 \pm 2.4 \text{ L mg C}^{-1} \text{ m}^{-1}$ ), indicating the pres-



**Figure 2.** Absorption spectral slope curve ( $S_{\lambda}$ ) of dissolved organic matter as an indicator of the presence of algal-derived carbon (peak at  $S_{289}$ ) and terrestrial carbon (steady increase in the slopes from 260 to 390 nm).

ence of more aromatic compounds, and possibly DOM-Fe complexes in summer. The shape of absorption spectral slopes showed a clear maximum at 382 nm, indicating the presence of terrestrial fulvic and humic acids. Consistent with the  $SUVA_{254}$  index,  $S_{382}$  values were relatively low in winter ( $0.0137 \pm 0.0016 \text{ nm}^{-1}$ ), with similar values in summer surface waters ( $0.0158 \pm 0.0014 \text{ nm}^{-1}$ ), but they show significant difference with bottom waters ( $0.0169 \pm 0.0007 \text{ nm}^{-1}$ ;  $p = 0.0061$ ). The individual spectral slope  $S_{289}$  indicates that higher amounts of algal-derived carbon were present in summer surface waters ( $0.0148 \pm 0.0009 \text{ nm}^{-1}$ ), especially in the most transparent pond KWK6 ( $0.0162 \text{ nm}^{-1}$ ), while their role was limited in the dark habitats under the ice ( $0.0113 \pm 0.0011 \text{ nm}^{-1}$ ) and in bottom waters ( $0.0112 \pm 0.0007 \text{ nm}^{-1}$ ). Carbon quantity and CDOM quality properties are summarized in Table 2, and the seasonal differences in absorption spectral slopes are shown in Fig. 2.

### 3.3 Microbial abundance and productivity

The abundance of all organisms examined was highest in the summer bottom waters, but the magnitude and seasonal distribution differed among taxa and ponds (Table 3). Especially, the abundant motile PNFs accumulated in the nutrient-rich bottom waters, where their abundance was in most ponds an order of magnitude higher than at the surface ( $110 \pm 89$  vs.  $27 \pm 8 \times 10^6 \text{ L}^{-1}$ ). The less abundant HNFs also favored the bottom waters, with an order of magnitude dif-

ference in the abundance between bottom and surface waters ( $9.2 \pm 11.6$  vs.  $0.8 \pm 0.7 \times 10^6 \text{ L}^{-1}$ ). The smaller phototrophic picoplankton (PPA) and bacteria were more uniformly distributed, although the abundances were significantly higher ( $p < 0.05$ ) at the bottom. Abundance was also quantified for free-living and particle-attached bacteria. During late winter,  $74 \pm 9\%$  of bacterioplankton was attached to particles, and  $46 \pm 21$  and  $70 \pm 17\%$  in summer surface and bottom waters (Table A1 in the Appendix).

When converted to biomass, the seasonal and vertical distribution pattern was even more accentuated (Table 3, Fig. 3). Overall, lower biomasses were encountered in winter, and higher at the bottom of the water column in summer. Total phototrophic biomass as Chl *a* concentration ( $5.6 \pm 2.5 \mu\text{g L}^{-1}$ ), as well as the PNF biomass ( $152 \pm 68 \mu\text{g CL}^{-1}$ ) under the ice, was relatively high. Summer surface phototrophic biomass was similar to winter waters ( $p = 0.1213$ ), and it was different ( $p < 0.05$ ) and highest in the dark bottom waters, likely reflecting cell sedimentation. The biomass of strictly phototrophic PPA was very low in winter ( $5.8 \pm 4.4 \mu\text{g CL}^{-1}$ ) and an order of magnitude higher in summer. The bacterial biomass was always high when compared to literature, with the lowest biomass occurring in winter ( $109 \pm 67 \mu\text{g CL}^{-1}$ ), 2 times higher biomass in summer at the surface ( $223 \pm 36 \mu\text{g CL}^{-1}$ ), and 5 times higher at the bottom ( $520 \pm 146 \mu\text{g CL}^{-1}$ ). The HNFs contributed little to the total microbial biomass, especially in the illuminated summer surface waters where they averaged  $3.2 \pm 2.9 \mu\text{g CL}^{-1}$ . Their biomass was higher

**Table 2.** Concentrations of carbon dioxide (CO<sub>2</sub>), methane (CH<sub>4</sub>), and dissolved organic carbon (DOC), and dissolved organic matter (DOM) optical properties, including absorption coefficient of DOM at 320 nm (*a*<sub>320</sub>), specific UV-absorbance index (SUVA<sub>254</sub>), and absorption spectral slopes at 289 nm (*S*<sub>289</sub>) and 382 nm (*S*<sub>382</sub>), of subarctic pond water sampled in late winter and during summer at the surface and bottom of the water column.

Site	CO <sub>2</sub> (μM)	CH <sub>4</sub> (μM)	DOC (mg L <sup>-1</sup> )	<i>a</i> <sub>320</sub> (m <sup>-1</sup> )	SUVA <sub>254</sub> (L mg C <sup>-1</sup> m <sup>-1</sup> )	<i>S</i> <sub>289</sub> (nm <sup>-1</sup> )	<i>S</i> <sub>382</sub> (nm <sup>-1</sup> )
Winter							
KWK 2	316	14.1	7.7	8.5	0.9	0.0099	0.0155
KWK 6	354	5.1	5.2	3.0	0.5	0.0112	0.0122
KWK 12	215	2.4	10.5	9.5	0.8	0.0105	0.0154
KWK 20	na	na	10.3	8.1	0.8	0.0125	0.0130
KWK 23	357	2.5	7.9	6.6	0.8	0.0124	0.0123
Summer surface							
KWK 2	72	0.6	5.6	26.4	4.7	0.0141	0.0169
KWK 6	27	0.5	4.1	11.8	3.4	0.0162	0.0135
KWK 12	55	0.3	6.0	26.5	4.5	0.0142	0.0170
KWK 20	71	0.3	7.1	38.9	6.1	0.0145	0.0156
KWK 23	53	0.3	6.3	30.7	5.4	0.0150	0.0161
Summer bottom							
KWK 2	na	na	6.1	51.5	6.7	0.0109	0.0156
KWK 6	422	145.2	4.2	21.4	4.3	0.0125	0.0173
KWK 12	761	259.0	7.4	67.9	7.4	0.0111	0.0173
KWK 20	815	311.9	9.3	115.6	9.8	0.0107	0.0172
KWK 23	570	131.6	7.5	94.5	10.2	0.0109	0.0171

**Table 3.** Phototrophic and heterotrophic properties of subarctic pond water sampled in late winter and during summer at the surface and bottom of the water column, including chlorophyll *a* concentration (Chl *a*), maximum photosynthesis parameter (*P*<sub>max</sub>), in situ primary production (PP), pigmented nanoflagellate abundance (PNF), phototrophic picoplankton abundance (PPA), bacterial production (BP), bacterial respiration (BR), bacterial abundance (BA) and heterotrophic nanoflagellate abundance (HNF).

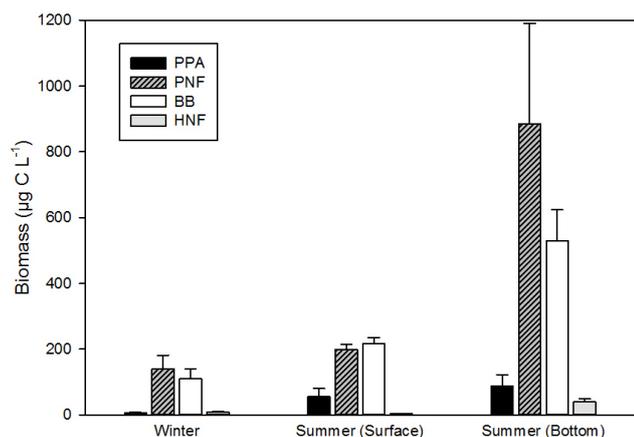
Site	Chl <i>a</i> (μg L <sup>-1</sup> )	<i>P</i> <sub>max</sub> (mg C m <sup>-3</sup> d <sup>-1</sup> )	PP (mg C m <sup>-3</sup> d <sup>-1</sup> )	PNF (×10 <sup>6</sup> L <sup>-1</sup> )	PPA (×10 <sup>5</sup> mL <sup>-1</sup> )	BP (mg C m <sup>-3</sup> d <sup>-1</sup> )	BR (mg C m <sup>-3</sup> d <sup>-1</sup> )	BA (×10 <sup>6</sup> mL <sup>-1</sup> )	HNF (×10 <sup>6</sup> L <sup>-1</sup> )
Winter									
KWK 2	2.8	1.2	0	33.0	0.2	0.2	1.3	4.2	2.0
KWK 6	9.9	0.8	0	51.0	0.9	0.5	2.2	11.2	3.3
KWK 12	11.5	5.8	0	29.1	0.7	0.3	1.6	2.5	2.4
KWK 20	2.8	0.8	0	6.3	0.2	0.8	3.0	8.7 <sup>a</sup>	2.4
KWK 23	0.8	1.5	0	10.0	0.1	0.4	2.0	3.7	0.3
Summer surface									
KWK 2	5.4	4.5	2.6	16.3	3.3	27.2	25.7	13.3	0.0
KWK 6	13.4	27.9	15.3	20.0	4.2	37.4	31.2	11.3	0.8
KWK 12	2.2	5.9	3.2	28.8	0.9	31.2	27.9	9.6	0.7
KWK 20	14.3	59.0	26.4	33.1	17.3	15.2	18.0	14.8	1.9
KWK 23	12.3	55.6	15.6	35.0	3.8	45.9	35.3	12.7	0.4
Summer bottom									
KWK 2	180.2	na	0.8	149.6	9.4	57.6	40.5	35.7	26.2
KWK 6	87.2	na	4.2	201.6	3.0	39.0	32.0	14.3	6.3
KWK 12	158.9	na	0.3	72.5	26.8	48.8	36.6	38.0	3.7
KWK 20	203.4	na	0.3	na	4.0	39.1	32.0	27.2	na
KWK 23	37.1	na	0.0	4.5	6.2	23.0	23.2	24.3	0.6

<sup>a</sup> BA was calculated from DAPI-stained microscope slide.

at the bottom ( $39.2 \pm 49.4 \mu\text{g C L}^{-1}$ ) and in winter samples ( $8.9 \pm 4.7 \mu\text{g C L}^{-1}$ ).

Snow and ice cover prevented light penetration under the ice, and therefore we assumed that no PP occurred in situ in late winter. However, the photosynthetic capacity of late winter phytoplankton community was not null, and represents the potential at the ice-out, which took place 2–3

weeks later. The maximum photosynthetic rate (*P*<sub>max</sub>) of late winter phytoplankton community was for most ponds an order of magnitude lower than *P*<sub>max</sub> of the summer community ( $2.0 \pm 2.1$  vs.  $30.6 \pm 26.1 \text{ mg C m}^{-3} \text{ d}^{-1}$ ). BP rates were very low in winter ( $0.4 \pm 0.2 \text{ mg C m}^{-3} \text{ d}^{-1}$ ) and different ( $p = 0.0001$ ) compared to the summer surface ( $31 \pm 14 \text{ mg C m}^{-3} \text{ d}^{-1}$ ) and bottom



**Figure 3.** Seasonal changes in the biomass of phototrophic picoplankton (PPA), pigmented nanoflagellates (PNFs), bacteria (BB), and heterotrophic nanoflagellates (HNFs). Error bars represent standard errors.

( $42 \pm 19 \text{ mg C m}^{-3} \text{ d}^{-1}$ ) BP rates. BR rates calculated from BP were low in winter ( $2.0 \pm 0.6 \text{ mg C m}^{-3} \text{ d}^{-1}$ ) and equally high during summer in surface ( $28 \pm 6 \text{ mg C m}^{-3} \text{ d}^{-1}$ ) and in bottom ( $33 \pm 6 \text{ mg C m}^{-3} \text{ d}^{-1}$ ). Particle-attached bacterioplankton played an important role in pond production at both sampling seasons (late winter:  $42 \pm 20\%$ ; summer surface:  $66 \pm 20\%$ ; bottom:  $68 \pm 44\%$ ; Table A1).

Winter, surface and bottom environmental characteristics formed three distinct groups on the PCA (Fig. 4a), with the first two axes explaining 72.9% of the variation in the environmental variables (PC1 46.0 and PC2 26.9%). Axis PC1 has strong positive correlations with TP ( $r = 0.91$ ),  $a_{320}$  ( $r = 0.80$ ), TSS ( $r = 0.83$ ) and GHG ( $\text{CO}_2$   $r = 0.92$  and  $\text{CH}_4$   $r = 0.90$ ), and strong negative correlations with pH ( $r = -0.81$ ) and  $S_{289}$  ( $r = -0.65$ ), whereas PC2 has positive correlation with temperature ( $r = 0.81$ ),  $S_{382}$  ( $r = 0.69$ ) and  $\text{SUVA}_{254}$  ( $r = 0.76$ ), and negative correlation with TN ( $r = -0.93$ ) and DOC ( $r = -0.67$ ). Heterotrophs (BBs and HNFs) were correlated with PC1 while phototrophs (PP, PNF and PPA) were more strongly correlated with PC2 (Fig. 4b). Further, BP and BB were best correlated with carbon quality indices  $a_{320}$ ,  $\text{SUVA}_{254}$ , and  $S_{289}$ , while HNFs were best correlated with conductivity (Table 4). PP, PNF and PPA were best correlated with a combination of carbon quality indices, DOC and nutrients (Table 4).

## 4 Discussion

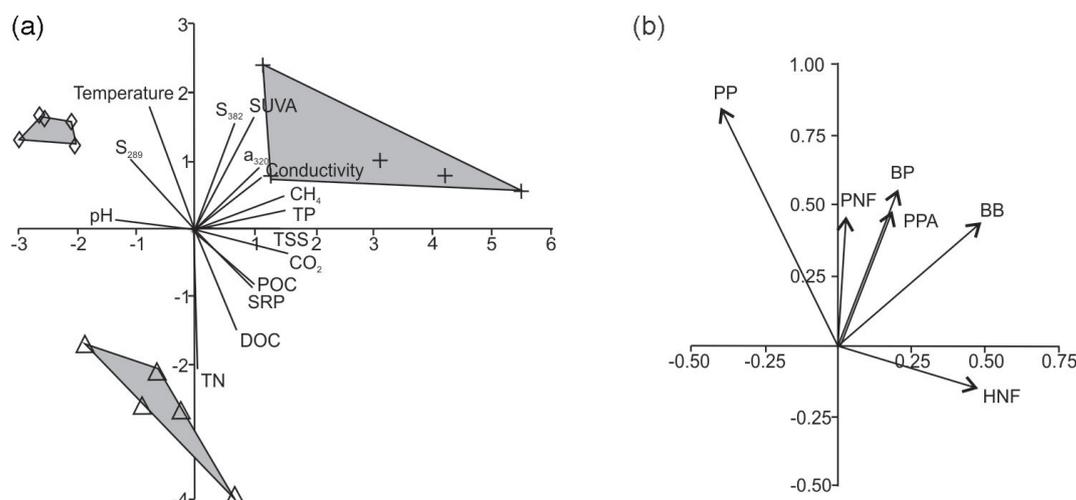
The large variations in physicochemical characteristics of thaw ponds between late winter and summer and between surface and bottom waters were reflected in their ecological properties. The cold and nutrient-rich winter waters had the lowest microbial abundance and productivity. The warm and illuminated summer surface waters favored primary pro-

duction, while the colder and darker bottom waters accumulated nutrients, carbon compounds and microorganisms. The overall supersaturation in GHG and the clear dominance of heterotrophy over phototrophy, as expressed in the biomass and activity of the microbial food web components, indicate that the majority of the energy was flowing through the heterotrophic food web.

### 4.1 Spatial and seasonal variations in GHG

Greenhouse gas storage in pond waters occurred in winter under the ice cover and in summer under the steep thermocline, due to the isolation of these water masses from atmospheric venting. Partial pressures measured were especially large in summer hypolimnetic waters (in average  $13\,674 \mu\text{atm}$  of  $\text{CO}_2$  and  $4783 \mu\text{atm}$  of  $\text{CH}_4$ ) compared to values at the surface (in average  $1206 \mu\text{atm}$  of  $\text{CO}_2$  and  $9 \mu\text{atm}$  of  $\text{CH}_4$ ). Such heterogeneity in the vertical distribution of GHG is typical for seasonally stratified lakes where bottom values are often many folds higher than at the surface (Eller et al., 2005; Guérin et al., 2006; Bastviken et al., 2008). Partial pressures of  $\text{CO}_2$  at the surface of thaw ponds ( $598\text{--}1545 \mu\text{atm}$ ) were comparable to values obtained in other regions, for example in European boreal lakes ( $990 \mu\text{atm}$ ; Kortelainen et al., 2006), Canadian boreal streams and rivers ( $1850 \mu\text{atm}$ ; Teodoru et al., 2009), western Siberian lakes and ponds ( $1935 \mu\text{atm}$ ; Shirokova et al., 2013), and Canadian subarctic and arctic ponds ( $1896 \mu\text{atm}$ ; Breton et al., 2009). Surface water  $\text{CH}_4$  partial pressures ( $6\text{--}13 \mu\text{atm}$ ) were also in the same range as values reported in lakes and ponds from North America ( $11\text{--}59 \mu\text{atm}$ ; Bastviken et al., 2008), Finland ( $23 \mu\text{atm}$ ; Juutinen et al., 2009), Canadian subarctic and arctic ( $22 \mu\text{atm}$ ; Breton et al., 2009), and western Siberia ( $30 \mu\text{atm}$ ; Shirokova et al., 2012).

The dominance of BR over PP, combined with the isolation of water masses during the long winter, followed by a very short spring mixing period (Laurion et al., 2010) and the steep summer stratification of these ecosystems controlled the accumulation of GHG in the hypolimnion. When normalized by the duration of the known isolation period (Laurion et al., 2010), GHG accumulated more rapidly in summer at the pond bottom ( $6.8 \pm 2.2 \mu\text{M CO}_2 \text{ d}^{-1}$  and  $1.9 \pm 0.9 \mu\text{M CH}_4 \text{ d}^{-1}$ ) than in winter under the ice cover ( $1.5 \pm 0.3 \mu\text{M CO}_2 \text{ d}^{-1}$  and  $0.030 \pm 0.027 \mu\text{M CH}_4 \text{ d}^{-1}$ ), assuming a constant accumulation rate during the two sampling periods and ignoring freezing exclusion effects. This may be partly due to the more abundant microbial community in summer, although most GHG produced likely came from benthic respiration, especially in such shallow systems (Kortelainen et al., 2006). Summer also provides fresh organic matter that may stimulate GHG producers that indeed presented higher respiration rates. The smaller  $\text{CH}_4$  accumulation rates in winter might also result from more optimal conditions for methanotrophy under the ice cover where bubbling  $\text{CH}_4$  is trapped and dissolves in water. The amount



**Figure 4.** (a) Seasonality in pond environmental variables illustrated with principal component analysis during late winter (triangles), summer surface (diamonds) and summer bottom (crosses). (b) Spearman correlation coefficients between biological variables and principal components 1 ( $x$  axis) and 2 ( $y$  axis). BP – bacterial production, BB – bacterial biomass, HNF – heterotrophic nanoflagellate biomass, PP – primary production, PNF – pigmented nanoflagellate biomass, and PPA – phototrophic picoplankton biomass.

of CH<sub>4</sub> in a lake was shown to depend on the size of the anoxic layer (Fendinger et al., 1992; Bastviken et al., 2004), and in the present study this layer represents a particularly large proportion of the water volume in summer. During the autumnal turnover period (lasting about 2 months), the whole water column becomes oxygenated, and in the first part of the winter surface layer of the pond may contain sufficient oxygen (Deshpande et al., 2015) for efficient CH<sub>4</sub> consumption. Anoxic methanotrophy has also been reported (Sivan et al., 2011) and may occur at our study sites. Nevertheless, these GHG storage periods indicate that subarctic thaw ponds will release large amounts of CH<sub>4</sub> during the autumnal mixing period, and also during spring turnover if it lasts long enough (Walter et al., 2006; Laurion et al., 2010).

#### 4.2 Seasonality in CDOM properties

We used  $a_{320}$  as an indicator of the quantity of CDOM, and SUVA<sub>254</sub>,  $S_{289}$  and  $S_{382}$  as indicators of the aromatic content of CDOM and the relative proportions in allochthonous (terrestrial) versus autochthonous (algal) carbon sources. The high  $a_{320}$  and SUVA<sub>254</sub> values indicate a large contribution by terrestrial humic substances in DOM at the surface of thaw ponds in summer, with values similar to a few years earlier in the same ponds (Breton et al., 2009; Watanabe et al., 2011). CDOM values were clearly smaller in winter (late winter  $a_{320}$ ; Table 2) and less aromatic (SUVA<sub>254</sub>) than in summer, suggesting that carbohydrates or protein-like compounds from bacteria-induced CDOM-degradation made up a larger fraction of DOM in winter. Laurion et al. (unpublished data) tested the role of bacterial CDOM degradation in arctic thaw ponds and found that its importance can almost match that of photodegradation in certain cases, with-

out significant losses in DOC, potentially indicating a bacterial transformation into less colored compounds. The higher DOC values observed in winter (average of 8.3 mg L<sup>-1</sup>) compared to summer at the surface (5.8 mg L<sup>-1</sup> in) thus likely resulted from a combination of a generation of dissolved compounds from the slower but uninterrupted microbial activity, and out-freezing inputs associated with the formation of ice. Although BP in late winter was low, the long subarctic winters allow accumulation effects that could lead to a decrease in CDOM. The labile part of DOC is known to be consumed within days by bacterioplankton, leaving the recalcitrant fraction of DOC to dominate in environments where fresh inputs are limited (Del Giorgio and Davis, 2002; Roehm et al., 2009; Guillemette and del Giorgio, 2011). However this is not clearly observed in the present study thaw ponds, where less aromatic and non-chromophoric fractions increased along winter, likely as the result of DOM recycling by a rich microbial food web, together with a lack of new carbon inputs from the frozen catchment.

The high  $a_{320}$  and SUVA<sub>254</sub> values found in bottom waters are very likely caused by the elevated dissolved iron concentrations (up to 5.1 mg L<sup>-1</sup>;  $r^2 = 0.947$  between SUVA<sub>254</sub> and Fe in summer samples,  $n = 10$ ). High concentrations of iron (> 2 mg L<sup>-1</sup>) were shown to form complexes with humic substances and increase DOM absorbance (Maloney et al., 2005; Xiao et al., 2013). During the autumnal turnover, CDOM-Fe complexes from bottom waters are exposed to photochemical reactions at the surface, potentially generating hydroxyl radicals that dissociate carboxyl groups from humic substances. These reactions can lead to lower molecular weight carbon components at the same time as increasing the water clarity (Brinkmann et al., 2003). The spectral slope values indicate that algal-derived proteins ( $S_{289}$ ; Gal-

**Table 4.** Spearman correlations between biological and environmental variables. Only statistically significant correlations are given.

Biological variable	Environmental variable	<i>n</i>	<i>r</i>	<i>p</i>
Bacteria production (BP)	Temperature	15	0.600	0.018
	TN	15	−0.524	0.045
	<i>a</i> <sub>320</sub>	15	0.704	0.003
	SUVA <sub>254</sub>	15	0.699	0.004
	S <sub>382</sub>	15	0.633	0.011
Bacteria biomass (BB)	O <sub>2</sub>	15	0.555	0.049
	Conductivity	15	0.668	0.007
	TP	15	0.600	0.018
	TSS	15	0.521	0.046
	Fe	10	0.879	0.001
	CH <sub>4</sub>	13	0.588	0.035
	<i>a</i> <sub>320</sub>	15	0.825	<0.001
	SUVA <sub>254</sub>	15	0.857	<0.001
	S <sub>382</sub>	15	0.716	0.003
HNF	Conductivity	15	0.657	0.011
	CH <sub>4</sub>	15	0.636	0.026
Primary production (PP)	Temperature	10	0.896	<0.001
	SRP	10	−0.662	0.007
	TN	10	−0.827	<0.001
	DOC	10	−0.610	0.016
	SUVA <sub>254</sub>	10	0.533	0.041
	S <sub>289</sub>	10	0.732	0.002
PNF	SRP	14	−0.732	0.003
	DOC	14	−0.647	0.012
PPA	TN	15	−0.665	0.007
	<i>a</i> <sub>320</sub>	15	0.825	<0.001
	SUVA <sub>254</sub>	15	0.717	0.003
	S <sub>382</sub>	15	0.564	0.029

gani et al., 2011) were more abundant in summer surface waters when primary production and Chl *a* were highest, and at the same time fulvic and humic acids (*S*<sub>382</sub>) also seemed more abundant than in winter. Algal proteins are generally highly available to heterotrophic organisms and were likely supporting the higher summer BP (Kritzberg et al., 2004). Nevertheless, Xiao et al. (2013) showed how high Fe concentrations, such as found in bottom waters, can significantly lower DOM absorption slopes; thus interpretations of bottom water *S*<sub>λ</sub> should be done cautiously.

### 4.3 Thaw pond microbial food web

Primary production rates in summer at the surface of thaw ponds were in the same range as in other arctic and subarctic lakes (Lizotte, 2008), while in late winter snow and ice cover prevented in situ PP. However, when exposed to light, the production of carbon by the winter phytoplankton community was in the same range as the heterotrophic production, suggesting that the high phytoplankton biomass encountered during late winter was made of mixotrophic species that switched from hetero- to autotrophic produc-

tion at ice-out. Similarly, the high Chl *a* and PNF biomass in the light-limited bottom waters in summer were most likely composed of mixotrophs. Many species of Chrysophyceae, Cryptophyceae and Dinophyceae are known to be 99 % mixotrophic (Olrik, 1998, and references therein). Chrysophyceae were dominating the phytoplankton community in KWK2 and KWK20 (Dupont, 2008, L. Forsström, unpublished data). Many Chrysophyceae species have flagella, and thus can thrive in sheltered humic ponds that have been shown to favor diurnal migrations of phytoplankton (Jones, 1991). This is especially advantageous in steeply stratified and shallow ponds where phytoplankton can migrate within a few hours (Arvola et al., 1991), and are able to use nutrient-rich hypolimnetic waters (Salonen et al., 1984). The high nutrient concentrations in bottom waters likely reflect the high biomass of PNF observed in these waters. Anoxic bottom waters were also devoid of *Daphnia* sp. (M. Wauthy et al., unpublished data) providing to phytoplankton a refuge from grazing. Absence of light, hypoxia and low temperatures further prevent photo-oxidation of algal pigments, preserving them for extended periods (Vallentyne, 1960). Thus some of the Chl *a* and PNF encountered at the bottom may have been from dead yet still not degraded cells. Another factor that could have caused bias in very high bottom water Chl *a* values is the occurrence of chlorophyll *b* containing green-sulfur bacteria that are known to colonize the anoxic layer of the studied ponds (Rossi et al., 2013), and that may interfere with the spectrophotometric analyses of Chl *a*. However, green-sulfur bacteria were not found in all ponds with high values of Chl *a*. In winter, the lack of PP and allochthonous inputs of carbon, and a DOM pool composed by a large fraction of molecules with low aromaticity (low SUVA<sub>254</sub> and *S*<sub>289</sub>) suggest that intensively recycled DOM formed the basal organic carbon pool in thaw ponds.

BP and BB were strongly linked to CDOM originating from a terrestrial source (high SUVA<sub>254</sub> and *a*<sub>320</sub>), indicating a positive link between bacterioplankton and terrestrial carbon (Tranvik, 1988; Crump et al., 2003). These results are in accordance with the experiment from Breton et al. (2009), which showed that bacterioplankton in thaw ponds was more carbon-limited than nutrient-limited (glucose was used as a labile carbon source). In contrast to several previous studies (Granéli et al., 2004; Säwström et al., 2007; Roiha et al., 2012), the amount of DOC did not seem to control the BP in our study. The studies that reported a positive correlation between DOC and BP were conducted in summer, when BP and DOC concentrations were highest, but without necessarily taking into account the variability in DOC type, which had a dominant role in our study.

The presence of a mixture of autochthonous and allochthonous carbon may have helped in making the terrestrial carbon available to bacteria. Phytoplankton and macrophytes apparently produced significant amounts of dissolved carbon in surface waters during summer as indicated by the larger *S*<sub>289</sub> peak. This carbon is considered more labile

(lower molecular weight) and with a higher nutritional value to heterotrophs (Kritzberg et al., 2004; Brett et al., 2009). The availability of high-quality phytoplankton carbon during summer possibly enhanced the bacterial use of more recalcitrant terrestrial carbon through the priming effect (Bianchi, 2011), which is a small supplement of nutritionally rich algal cells increasing bacterioplankton terrestrial carbon assimilation.

Correlation between TSS and TP ( $r = 0.903$   $p < 0.001$ ,  $n = 15$ ) indicates that phosphorus was binding to particles, which may be more accessible to attached bacterioplankton than to phytoplankton in summer surface waters where SRP concentrations were low. The high proportion of particle-attached bacterial abundance ( $62 \pm 30\%$ ) and productivity ( $59 \pm 30\%$ ) in thaw ponds suggests that particle attachment is beneficial. Although studies have shown that BP is often nutrient-limited (Kritzberg et al., 2004; Vrede, 2005), there were no correlation between BP and nutrients in thaw ponds.

One possible additional explanation for the difference between winter and summer BP is the difference in temperature that is known to have a large impact on production (Adams et al., 2010). On the other hand, bacterial communities are known to adapt at low temperatures, and some communities can have multiple temperature maxima (Adams et al., 2010). In our study, the highest BP was found in relatively cool bottom waters. Moreover, bacterial biomass (BB) was not correlated with temperature, although several studies found temperature as a main driver of bacterial biomass (Ochs et al., 1995; Rae and Vincent, 1998). Trophic interactions certainly contributed to the observed patterns in microbial abundance. Rotifers are the dominant zooplankton group in the studied thaw ponds, while the cladoceran community is composed of a few *Daphnia* sp. observed only at the surface (Bégin, 2014). These two zooplankton groups were only found in summer, when they could exert a grazing pressure on bacteria and phytoplankton, but experiments would be needed to confirm the extent of grazing. We tentatively estimated the top-down control of nanoflagellates on bacteria, using the average nanoflagellate ingestion rate of  $0.66 \text{ pg C cell}^{-1} \text{ day}^{-1}$  measured for *Dinobryon* sp. in Svalbard (Laybourn-Parry and Marshall, 2003). From these calculations, it seems that nanoflagellate grazing on bacteria could have been efficient enough in late winter when the grazing potential exceeded 100% of the BP, while summer removal estimations corresponded to only 1.7% at the surface and 9.3% at the bottom of the BP, possibly contributing to explaining the accumulation of BB in summer.

## 5 Conclusions

Low light intensity, strong stratification and high concentrations of CDOM in subarctic thaw ponds are the main drivers creating an environment highly beneficial to heterotrophs. We have also shown some important differences in their ecosystem dynamics between winter and summer. Although a high biomass of bacterioplankton is found under the ice, the ponds are producing relatively small amounts of carbon in late winter, possibly caused by the lack of fresh carbon inputs and its associated priming effect. Nevertheless, the ponds are accumulating GHG during winter. The relatively high phototrophic biomass and primary production potential encountered under the ice are at the same level as the heterotrophic biomass and production, indicating the importance of mixotrophy in winter. In summer, fresh primary production and terrestrial carbon inputs enhance the heterotrophic production. Although a higher phototrophic (mixotrophic) biomass is encountered in summer, primary production is restricted by light and space and is significant only at the pond surface; thus the heterotrophic carbon production remains significantly higher. Consequently, GHGs continue to accumulate in hypolimnetic waters during this strongly stratified summer period, and are most likely released during the autumnal turnover period, in addition to the constant emissions of GHGs throughout the summer from the highly supersaturated surface waters.

## Appendix A

**Table A1.** Particle attachment of bacterioplankton in winter, summer at the surface and bottom of ponds, including sonicated bacterial abundance ( $BA_T$ ) representing the total abundance, bacterial abundance before sonication ( $BA_{FL}$ ) representing free-living bacteria, proportion of particle-attached bacteria ( $BA_P$ ), total bacterial production (BP; as in Table 3), filtered ( $3\ \mu\text{m}$ ) bacterial production ( $BP_{FL}$ ) and proportion of particle attached bacterial production ( $BP_P$ ).

Site	$BA_T$ ( $\times 10^6\ \text{mL}^{-1}$ )	$BA_{FL}$ ( $\times 10^6\ \text{mL}^{-1}$ )	$BA_P$ (%)	BP ( $\text{mg C m}^{-3}\ \text{d}^{-1}$ )	$BP_{FL}$ ( $\text{mg C m}^{-3}\ \text{d}^{-1}$ )	$BP_P$ (%)
Winter						
KWK 2	4.2	0.8	81	0.2	0.1	54
KWK 6	11.2	3.2	72	0.5	0.2	54
KWK 12	2.5	0.5	79	0.3	0.2	48
KWK 20	na	na	na	0.8	0.4	47
KWK 23	3.7	1.4	62	0.4	0.4	7
Summer surface						
KWK 2	13.3	5.8	57	27.2	4.7	83
KWK 6	11.3	8.5	24	37.4	3.3	91
KWK 12	9.6	6.0	37	31.2	17.4	56
KWK 20	14.8	4.1	73	15.2	7.0	54
KWK 23	12.7	7.7	40	45.9	18.6	60
Summer bottom						
KWK 2	35.7	11.3	68	57.6	1.7	97
KWK 6	14.3	8.3	42	39.0	1.5	96
KWK 12	38.0	11.2	70	48.8	26.0	47
KWK 20	27.2	3.7	86	39.1	1.0	97
KWK 23	24.3	4.2	83	23.0	26.6	0

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