



Using water stable isotopes for tracing surface and groundwater flow systems in the Barlow-Ojibway Clay Belt, Quebec, Canada

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Complete List of Authors:	Rey, Nathalie; Universite du Quebec en Abitibi Temiscamingue, Institut de recherche en mines et environnement Rosa, Éric; Universite du Quebec en Abitibi Temiscamingue, Groundwater Research Group Cloutier, Vincent; Université du Québec en Abitibi-Témiscamingue, Institut de recherche en mines et en environnement; Lefebvre, René; Institut national de la recherche scientifique, Centre Eau Terre Environnement
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1 **Using water stable isotopes for tracing surface and groundwater flow**
2 **systems in the Barlow-Ojibway Clay Belt, Quebec, Canada**

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4 Nathalie Rey^{1*}, Eric Rosa¹, Vincent Cloutier¹ and René Lefebvre²

5
6 ¹ Groupe de Recherche sur l'Eau Souterraine, Institut de Recherche en Mines et en
7 Environnement, Université du Québec en Abitibi-Témiscamingue, Amos, Québec,
8 Canada, J9T 2L8.

9
10 * Corresponding author: nathalie.rey@uqat.ca ((819) 732-8809 #8033)

11
12 ² Centre Eau Terre Environnement, Institut national de la recherche scientifique,
13 Québec, Canada, G1K 9A9.

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19 **Abstract**

20 This study aims to improve the understanding of surface and groundwater flow systems
21 based on water stable isotopes data in a 19,549 km² region of the Barlow-Ojibway Clay
22 Belt, in western Quebec, Canada. The available geochemical database contains
23 645 samples including precipitation, snow cores, surface waters, groundwater and
24 springs. All samples were analyzed for water stable isotopes ($\delta^2\text{H}$ - $\delta^{18}\text{O}$) and
25 complementary tritium analyses were conducted on 98 groundwater and spring samples.
26 Precipitations depict a clear temperature-dependent seasonal pattern and define a local
27 meteoric water line (LMWL) without a latitudinal trend in $\delta^2\text{H}$ - $\delta^{18}\text{O}$. Samples collected
28 from the snowpack plot on the LMWL, suggesting that the bulk snowpack preserves the
29 isotopic composition of precipitation throughout the frozen period, prior to the spring
30 snowmelt. Surface water samples define a local evaporation line (LEL) and evaporation
31 over inflow (E/I) ratios range between 0% and 36%. Groundwater and springs samples
32 are evenly distributed around the LMWL, suggesting that evaporation processes are
33 limited prior to infiltration and that surface waters do not significantly contribute to
34 groundwater recharge. Shallow unconfined aquifers present a greater variability in $\delta^2\text{H}$ -
35 $\delta^{18}\text{O}$ compared to confined aquifers located further down gradient, suggesting the mixing
36 of varied recharge waters along the regional groundwater flow system. A three-
37 component mixing model based on isotopic and specific electrical conductivity data
38 allows the quantification of such mixing processes. The interpretation of isotopic data
39 constrains a regional-scale conceptual model of groundwater flow systems and describes
40 processes related to the timing of recharge, evaporation, mixing and discharge.

41

42 **Keywords:**

43 Water stable isotopes; Groundwater; Surface waters; Precipitation; Boreal region

44

45 **Résumé**

46 Cette étude vise à mieux comprendre les systèmes d'écoulement dans une région de
47 19 549 km² au sein de la ceinture argileuse Barlow-Ojibway (ouest du Québec, Canada).
48 La base de données contient 645 échantillons analysés pour les isotopes stables de l'eau
49 ($\delta^2\text{H}$ - $\delta^{18}\text{O}$) et inclue les précipitations, la neige au sol, les eaux de surface, les eaux
50 souterraines et les sources alors que 98 échantillons d'eau souterraine et de sources ont
51 été analysés pour l'activité tritium. La composition isotopique des précipitations présente
52 un cycle saisonnier dépendant de la température et définissant une droite locale des eaux
53 météoriques (DLEM) sans tendance $\delta^2\text{H}$ - $\delta^{18}\text{O}$ reliée à la latitude. Les données de neige
54 au sol tombent sur la DLEM, ce qui suggère que le couvert nival préserve sa composition
55 isotopique en hiver, avant la fonte des neiges printanières. Les échantillons d'eau de
56 surface définissent une droite évaporatoire locale (DEL) et les rapports de l'évaporation
57 sur les flux intrants variant entre 0% et 36% à l'échelle régionale. Les données d'eau
58 souterraine et de sources suggèrent que les processus d'évaporation sont limités avant
59 l'infiltration et que les eaux de surface contribuent peu à la recharge. Les aquifères à
60 nappe libre peu profonds présentent une plus grande variabilité isotopique par rapport aux
61 aquifères à nappe captive situés en aval le long des lignes d'écoulement, ce qui suggère
62 une atténuation de la variabilité isotopique issue de la recharge le long des systèmes
63 régionaux. Un bilan géochimique appuyé sur les données isotopiques et de conductivité
64 électrique de l'eau permet de quantifier ces mélanges. L'interprétation des données
65 isotopiques permet de contraindre des modèles conceptuels représentant les systèmes
66 d'écoulement régionaux et de documenter les processus de recharge, d'évaporation, de
67 mélange et de résurgence.

68 Introduction

69 Groundwater plays key functions for several ecosystems and represents an essential source of
70 potable water for approximately half of the world's population (Taylor et al. 2013). Nevertheless,
71 groundwater resources have been heavily used for human water supply and agriculture for many
72 years on a global scale (WWAP 2009) and climate change will most likely affect the hydrological
73 cycle and aquifers in the near future (Green et al. 2011). Groundwater resources are expected to be
74 increasingly stressed in response to demographic growth and growing water demand (Vörösmarty
75 et al. 2000). It is estimated that by 2025, nearly 50% of the world's population will be living in
76 water-stressed regions (WHO 2017). Such considerations stress the critical need to better
77 understand hydrogeological processes at various scales to support concrete actions aimed at
78 protecting groundwater resources, to maintain ecosystems and meet water supply needs of future
79 generations. Regionally and locally, groundwater protection must rely on land management
80 strategies that are based on an appropriate understanding of the hydrogeological system. This is
81 especially important in areas where most of the population relies on groundwater for potable water
82 supply, as is the case for the Abitibi-Temiscamingue region in western Quebec. This region is
83 renowned for its large eskers and moraines and for the remarkable quality of groundwater found
84 within the aquifers associated with these formations (Veillette et al. 2004; Collini et al. 2007).
85 However, these shallow unconfined aquifers are highly vulnerable to contamination and face
86 increasing human pressures such as forest operations, sand and gravel extraction, former in-trench
87 disposal sites, private and municipal wells and commercial water bottling (MRNF 2006; Nadeau et
88 al. 2015; Cloutier et al. 2016). The impacts on the groundwater resources associated with eskers
89 might also affect aquifers located down-gradient, further highlighting the need to better understand
90 regional hydrogeological processes. In this context, providing regional stakeholders with a coupled
91 understanding of physical and geochemical conditions is mandatory to ensure that land-use
92 decisions related to groundwater protection are based on reliable information.

93 Geochemical tracers are included in a wide spectrum of hydrological studies. They are increasingly
94 used to delineate flowpaths and quantify the processes affecting surface and groundwater resources
95 at various scales and in various regions of the world (e.g. Jeelani et al. 2013; Kortelainen and Karhu
96 2004; Wassenaar et al. 2011). Among the wide range of geochemical tracers that have been
97 exploited, water stable isotopes (^2H - ^{18}O) data have provided unique insights into hydrological

98 processes (Araguás-Araguás et al. 2000; Gibson et al. 2005; Birks and Gibson 2009). The stable
99 isotopes of water are partitioned within a hydrological system, mainly in response to systematic
100 fractionation mechanisms occurring within the water cycle, including phase-change (solid-liquid-
101 vapour) processes, diffusion and rainout (Craig 1961; Rozanski et al. 1993; Gat 1996; Clark and
102 Fritz 1997). The isotopic composition of water is widely used in theoretical and applied
103 hydrological studies to document sources of atmospheric moisture and precipitation patterns
104 (Dansgaard 1964; Fritz et al. 1987), snow and ice-melting (Siegel and Mandle 1984), evaporation
105 processes in cold regions (Gat et al. 1994) and drylands (Kebede et al. 2009), sources mixing
106 (Ferguson et al. 2007; Wolfe et al. 2007) and groundwater recharge (Praamsma et al. 2009), among
107 other applications.

108 Cloutier et al. (2013; 2015; 2016) provided a comprehensive assessment of groundwater resources
109 in the Abitibi-Témiscamingue region, within the framework of the *Programme d'Acquisition de*
110 *Connaissances sur les Eaux Souterraines* (PACES) supported by the Quebec Ministry of the
111 Environment. A regional scale conceptual hydrogeological model representing regional flowpaths
112 from unconfined aquifers, located in recharge areas, to confined aquifers, found beneath the clay
113 belt, was proposed through the course of these studies. Nevertheless, the proposed conceptual
114 model is mainly constructed on the basis of physical hydrogeological data, whereas geochemical
115 data have not been yet fully exploited, which leads to important limitations in the understanding of
116 regional hydrogeological dynamics. Fitting into the pre-established framework, this study aims to
117 improve the understanding of regional surface and groundwater flow systems in the southern
118 portion of the Barlow-Ojibway Clay Belt (Abitibi-Témiscamingue, Quebec, Canada) on the basis
119 of geochemical data. The present study focuses on the stable (^2H - ^{18}O) and radioactive (^3H) isotopes
120 of the water molecule. The proposed approach involves a regional-scale comprehensive assessment
121 of isotopic data within all hydrological components: precipitations, winter snowpack, surface
122 waters, groundwater and springs. The specific objectives of the present study are (1) to document
123 the atmospheric processes driving spatiotemporal variations in precipitation isotopic composition,
124 (2) to assess the potential interactions and mixing processes between surface waters and aquifers,
125 and (3) to decipher the mechanisms controlling the water isotopic variability in regional aquifers.
126 Ultimately, the data are interpreted to support an improved conceptual model of regional
127 groundwater flow systems representative of observed hydrogeological and geochemical conditions.

128

129 Study Area

130 The study area (Figure 1) covers 19,549 km² and corresponds to the municipalized territory of the
131 Abitibi-Témiscamingue region in Western Quebec, Canada. It encompasses the City of Rouyn-
132 Noranda and the Regional County Municipalities (RCM) of *La Vallée-de-l'Or*, *Abitibi*, *Abitibi-*
133 *Ouest* and *Témiscamingue*, for a total population of approximately 148,000 inhabitants (MAMROT
134 2016). It is estimated that more than 70% of this population relies on groundwater as a source of
135 domestic drinking water. The characteristics of the study area are briefly summarized below.
136 Further details about the hydrogeological framework can be found in Nadeau et al. (2015), Cloutier
137 et al. (2016), Nadeau et al. (2017) and within the geological maps of the Geological Survey of
138 Canada (Veillette 1986; 1987a; 1987b; Thibaudeau and Veillette 2005; Paradis 2005; 2007).

139

140 **Figure 1. Study area with the location of sampling sites**

141

142 *Physiography and hydrography*

143 The landscape of the study area is relatively flat, with elevations generally ranging
144 between 280 m and 320 m. The main positive reliefs correspond to heterogeneously
145 distributed bedrock outcrops, while the main depressions are associated with large lakes
146 (Abitibi, Témiscamingue and Simard; Figure 1). Despite its featureless landscape, the
147 region is crossed by the continental water divide separating surface waters of the
148 St. Lawrence River Basin (southern sector of the study area, Ottawa River watershed)
149 and those of the James Bay Basin (northern sector of the study area, Moose, Harricana
150 and Nottaway rivers watersheds).

151 *Climate and air masses*

152 The study region is characterized by a cold and humid continental climate with damp
153 summers, rather cold and long winters and a marked seasonality. The average monthly air
154 temperature remains below the freezing point from November to March (based on
155 Quebec Ministry of the Environment climate data; 1981-2010). The region is

156 characterized by a significant latitudinal thermal gradient and a difference of 2.1 °C
157 between the average annual air temperature in Temiscamingue (south) and Abitibi (north)
158 was reported by Asselin (1995). This difference is likely imparted by latitude and by the
159 influence of the continental water divide that creates a discontinuity between the northern
160 and southern sectors of the study area. The presence of numerous lakes on a territory
161 characterized by a gentle relief likely favors the development of buffered microclimates.
162 Such a phenomenon is documented in the vicinity of lakes Abitibi and Témiscamingue
163 (Asselin 1995). Cloutier et al. (2015) provided daily air temperatures and vertical inflows
164 based on interpolated data provided by *Centre d'Expertise Hydrique du Québec* (CEHQ)
165 for a period ranging from 1900 to 2010 (Poirier et al. 2012). Vertical inflows correspond
166 to the fluxes of water from liquid precipitations and from the melting of snow, and
167 therefore to the fluxes of water that are available for infiltration or surface runoff. If the
168 temperature is below the freezing point, vertical inflows can be null even if precipitation
169 events are observed. In the absence of snow on the ground and when precipitations fall as
170 liquid water, vertical inflows correspond to precipitation. Yearly average vertical inflows
171 range between 790.5 and 856.5 mm at the regional scale. The vertical inflows associated
172 with snow falling during the frozen period will mainly be observed during spring
173 snowmelt. The greatest vertical inflows are recorded in April for most of the study area,
174 except for the northernmost sector where maximum values are observed in May,
175 probably due to later thaw (Cloutier et al. 2015). Minimum vertical inflows are recorded
176 in January over the entire region due to temperatures that generally remain below the
177 freezing point. Regional-scale monthly averages of daily maximum and minimum
178 temperatures reveal that the warmest month is July (with daily average minimum and
179 maximum temperatures of 11.2 °C and 23.7 °C, respectively) and the coldest month is
180 January (with daily average minimum and maximum temperatures of -23.3 °C and -10.3
181 °C, respectively). The mean annual precipitation ranges from 875 mm for *La Vallée-de-*
182 *l'Or* RCM and 989 mm in the city of Rouyn-Noranda (Quebec Ministry of the
183 Environment climate data).

184 The main air masses affecting western Quebec are (1) the cold and dry Arctic, (2) the
185 moist and warm Maritime-tropical air that originates from the Caribbean, subtropical
186 Atlantic and Gulf of Mexico and migrates northwards following the Mississippi and

187 Missouri valleys before being deflected eastwards across the Great Lakes basin and (3)
188 the northern Pacific westerlies (Bryson 1966; Bryson and Hare 1974; Hare and Hay
189 1974). The Pacific westerlies are dominant in the continental climate for much of the year
190 (Hare and Hay 1974). Summers in the study area are warm and humid, reflecting the
191 dominance of Maritime-tropical air punctuated by intrusion of mild and dry westerlies.
192 Winters are influenced by Arctic air masses that lead to cold and dry conditions (Bryson
193 and Hare 1974). The Abitibi-Temiscamingue region is characterized by weather
194 variability and instability reflecting the contribution of the different air masses. The
195 dominant winds generally originate from the northwest during winter and southwest
196 during summer. The polar jet stream, corresponding to the limit between the Arctic and
197 Maritime-tropical air masses, crosses the study area along a sinuous west-east pattern at
198 latitudes roughly corresponding to the position of the continental water divide.

199 *Hydrogeological framework*

200 The regional geological framework, as conceptually represented in Figure 2, is described
201 in details in numerous previous studies (Nadeau et al. 2015; Cloutier et al. 2016; Nadeau
202 et al. 2017; Veillette 1986; 1987a; 1987b; Veillette et al. 2004; Thibaudeau and Veillette 2005;
203 Paradis 2005; 2007) and it is thus only briefly presented here. Table 1 provides a summary
204 of the hydrological domains (atmosphere, surface, subsurface), components (precipitation
205 (snow, rain), surface water, groundwater, springs) and aquifers (granular (unconfined &
206 confined), fractured rock (unconfined & confined)) that are discussed in this study, in
207 relation with the main geological units of the region. The bedrock of the study area
208 (geological unit A in Figure 2) is composed of a wide variety of igneous, metamorphic
209 and metasedimentary rocks of the Abitibi and Pontiac sub-provinces, both components of
210 the Superior Geological Province. Groundwater flow within the bedrock is controlled by
211 the architecture of structural discontinuities. Cloutier et al. (2016) estimated hydraulic
212 conductivities ranging between 10^{-9} and 10^{-4} m/s for this unit. However, there is currently
213 no sufficient data to propose a regional-scale interpretation of bedrock hydrogeological
214 characteristics based on structural interpretations. Nevertheless, Rouleau et al. (1999)
215 proposed that groundwater flow within the bedrock of the study area is most likely
216 restricted to depths shallower than 75 m where sub-horizontal fractures are sufficiently

217 interconnected to allow significant groundwater flow. Most of the domestic wells of the
218 region are withdrawing groundwater from this unit using wells that are cased through
219 surficial sediments and open across the bedrock.

220

221 **Table 1. Summarized characteristics of the regional hydrogeological framework**

222

223 The oldest unconsolidated unit of the region corresponds to the till associated with the
224 last glaciation (geological unit B in Figure 2). It is recognized as a compact heterometric
225 geological unit characterized by a matrix composed of sand, silt and clay (Cloutier et al.
226 2015). The hydrogeological characteristics of this unit are poorly documented because it
227 is rarely exploited for groundwater supply, although it is not considered as a regional
228 aquitard (Cloutier et al., 2013; 2015). This till unit is overlain in places by glaciofluvial
229 sediments. The coarse grained glaciofluvial sediments (eskers and moraines; geological
230 unit C1 in Figure 2) of the region were mainly deposited at the emergence of subglacial
231 meltwater within proglacial Lake Barlow-Ojibway (Nadeau et al. 2015). These
232 formations are characterized by relatively high hydraulic conductivities, 10^{-6} to 10^{-1} m/s
233 according to Cloutier et al. (2016), and constitute the most productive unconfined
234 aquifers of the region (Nadeau et al. 2015). Other glaciofluvial sediments, mainly
235 consisting in finer sand and gravel, are also widespread between the main eskers and
236 moraines of the region (geological unit C2 in Figure 2). This unit is generally
237 characterized by hydraulic conductivities that are slightly lower than that of the eskers
238 and moraines (Cloutier et al. 2013; 2015; 2016). The glaciofluvial sediments are covered
239 in places by fine grained deep-water glaciolacustrine sediments deposited within
240 proglacial Lake Barlow-Ojibway (identified as the Clay Belt in Figure 1; geological unit
241 D in Figure 2). Owing to its structure consisting in centimeter-scale alternating layers of
242 clay and silt (varves), this unit is considered as a regional aquitard (Cloutier et al. 2015),
243 although silt horizons may allow preferential sub-horizontal groundwater flow.
244 Sublittoral sands and beach or eolian sediments (geological unit E in Figure 2) overlying
245 the fine grained deep-water glaciolacustrine sediments are spatially associated with the

246 glaciofluvial deposits whereas organic deposits (geological unit F in Figure 2) are
247 widespread across the region, with large peatlands often developing on the flanks of
248 eskers and moraines.

249 Given this geological framework, and based on previous work from Cloutier et al. (2016)
250 and Nadeau et al. (2017), four main types of aquifers are distinguished within the study
251 region (Figure 2) and used to regroup the isotopic data:

252 (1) **Unconfined granular aquifers**, mainly associated with glaciofluvial formations
253 (eskers and moraines; number 5 in Figure 2). These formations are characterized
254 by relatively high hydraulic conductivities and represent the most productive
255 aquifers of the region, providing a high quality resource to numerous municipal
256 and private wells.

257 (2) **Confined granular aquifers**, in sectors where till and glaciofluvial sediments are
258 covered by the fine grained deep-water glaciolacustrine sediments (number 6 in
259 Figure 2);

260 (3) **Unconfined fractured rock aquifers**, essentially restricted to sectors
261 characterized by bedrock and till outcrops (number 7 in Figure 2);

262 (4) **Confined fractured rock aquifers**, in sectors covered by fine grained deep-water
263 glaciolacustrine sediments (number 8 in Figure 2). Most residences in rural
264 sectors of the region withdraw water from private wells installed in this aquifer.

265 Cloutier et al. (2013; 2015; 2016) provided estimates of the average groundwater
266 recharge rates (*GRR*) by calculating the water balance for each parcel of the study region
267 on a monthly basis. The calculations were provided on a 100 m x 100 m grid over the
268 study region, using the following equation:

$$269 \quad GRR = VI - R - \Delta W_S - ET \quad \text{Equation 1}$$

270 These water balance calculations are based on the vertical inflows (*VI*) provided by
271 Poirier et al. (2012) and on hypotheses associated with the regional hydrogeological

272 framework for estimating runoff (R), water storage within the unsaturated soil (ΔW_S) and
273 evapotranspiration (ET). Runoff coefficients were estimated according to land use, soils
274 characteristics and surface slopes, among others. Potential evapotranspiration rates were
275 quantified using the Thornthwait (1944) equation with climate data from Poirier et
276 al. (2012). The water storage within the unsaturated zone and the available water for
277 transpiration were evaluated for the different soils based on estimates of water retention
278 capacities, wilting points and thickness of the root zone. The calculated recharge rates
279 range between less than 91 mm/yr in sectors where fine-grained deep-water
280 glaciolacustrine sediments are found and more than 300 mm/yr, mainly in sectors
281 corresponding to eskers / moraines and other unconfined granular aquifers. Overall,
282 maximum recharge rates are observed during the snowmelt period whereas the lowest
283 rates are recorded throughout the winter period, when temperatures are below the
284 freezing point.

285

286 **Figure 2. Regional hydrogeological conceptual model and water sampling scheme**
287 **used to characterize the isotopic signature of all hydrological system components**

288

289 **Methods**

290 *Sampling procedures and in situ measurements*

291 Overall, 645 water samples were analyzed for their isotopic composition ($\delta^2\text{H}$ - $\delta^{18}\text{O}$) as
292 part of this study, over a period ranging from 2006 to 2015 (Figure 1; Table 2). Table 2
293 provides a summary of the available data and associated sampling periods. From this
294 dataset, 98 water samples (only springs and groundwater) were also analyzed for their
295 tritium (^3H) content, over a period ranging from 2006 to 2011. The procedures associated
296 with the sampling of precipitations, the snowpack, surface waters, springs and
297 groundwater are presented separately below.

298

Table 2. Summary of available water stable isotope data

299 *Precipitation monitoring*

300 Monthly composite precipitation samples were collected at four monitoring stations
301 within the study area (Figure 1; Station 1 = Béarn; Station 2 = Montbeillard; Station 3 =
302 Sainte-Hélène-de-Mancebourg; Station 4 = Amos). Samples were collected monthly from
303 July 2013 to 2015 at stations 1-2-3 (Figure 1), three sites included in the Quebec Ministry
304 of the Environment Climate Monitoring Network. Rainfall samples were collected using
305 a 10.1 cm inside diameter *All Weather P-2000* standard rain gauges from *Productive*
306 *Alternatives*. A thin layer (approximately 1 cm) of paraffin oil (Mineral Oil, Light
307 (NF/FCC), *Fisher Chemical*) was added to the rain gages to prevent evaporation and
308 plastic screens were installed at the base of the rain gauge funnels to block debris and
309 insects. Snowfall samples were collected owing to the participation of local operators /
310 observers associated with the monitoring network. These observers are mandated to make
311 measurements of snow (calculated in terms of equivalent water) intercepted by shielded
312 *Nipher* snow gauges. Snow samples are collected daily at stations 1 and 3 (but not at
313 station 2 owing to logistical constraints), left to melt overnight and subsequently stored in
314 1 L high density polyethylene (HDPE) bottles. Daily samples are mixed in order to obtain
315 composite monthly samples. Paraffin oil (Mineral Oil, Light (NF/FCC), *Fisher*
316 *Chemical*) is added to the bottles in order to prevent evaporation. The Amos monitoring
317 station (Station 4), in operation from September 2009 to 2015, is located on the roof of
318 the university campus and is not associated with the MDDELCC climate network. At this
319 station, rain and snow samples are collected daily using a 30 cm ID bucket. Snow
320 samples were left to melt overnight at 4 °C and subsequently stored in HDPE bottles.
321 Daily samples were combined to produce a monthly composite sample. In all cases, the
322 volume of monthly composite samples was first evaluated in the laboratory and smaller
323 aliquots were subsequently stored in 60 ml HDPE bottles and kept at 4 °C until stable
324 isotopes ($\delta^2\text{H}$ - $\delta^{18}\text{O}$) analyses.

325 *Samples from the Snowpack*

326 A sampling campaign was carried out in March 2014 in order to retrieve cores from the
327 snowpack accumulated throughout the winter. Sampling sites were selected in the

328 western part of the study region in an effort to collect data in the vicinity of the grid
329 points used by the CEHQ for calculating vertical inflows (Poirier et al. 2012). At each
330 site, a straight trench varying in length from 5 m to 6 m was first dug in a direction
331 oriented perpendicular to the wind. Snow cores ($n \geq 5$) were subsequently sampled at
332 ≤ 1 m intervals, on the side of the trench facing the dominant wind. Snow cores were
333 collected using 4.2 cm ID x 1.6 m length HDPE tubes allowing the collection of the
334 snowpack over its entire height. Samples were weighed on site using a *Mettler Toledo*
335 digital scale with a precision of 0.0001 kg in order to calculate equivalent water contents.
336 For each site, the collected snow cores were transferred and combined within wide neck
337 HDPE bottles in order to obtain > 800 g composite samples. Samples were kept frozen
338 within the tightly sealed wide neck bottles during the field campaign and subsequently
339 left to melt overnight at room temperature. The melted samples were filtered using
340 $0.45 \mu\text{m}$ *Waterra FHT* cartridges to remove soil particles and/or windblown debris
341 embedded in the snow. The filtered samples were transferred into 30-60 ml HDPE
342 containers and stored at $4 \text{ }^\circ\text{C}$ until stable isotopes ($\delta^2\text{H}$ - $\delta^{18}\text{O}$) analyses.

343

344 *Surface waters*

345 Surface water samples were collected in July 2009 and from July to October, 2013.
346 Rivers and streams were sampled near the central part of the main channel, at depths of
347 0.5-1.5 m below surface. *In situ* parameters (pH, dissolved oxygen, conductivity, redox
348 potential and temperature) were measured using a *YSI 556 MPS* probe, a few centimeters
349 downstream of the sampling point. A 7.32 m *NASCO Swing Sampler* telescopic pole
350 equipped with a 1 L HDPE bottle was used to collect samples from the banks of small
351 rivers and streams. The perch and its container were rinsed on site at a point located a few
352 meters downstream of the sampling site. Whenever sampling from a bridge was possible,
353 the samples were collected using 1 L *Waterra ecobailers*. A nylon wire equipped with a
354 float was attached to the bailer to indicate the sampling depth. The bailers were
355 systematically rinsed on site, with water from rivers prior to sampling. Some of the
356 streams and rivers were sampled immediately at the outlet of large lakes (Table 3). These

357 samples are used to provide an instantaneous estimate of the isotopic composition of
358 water in the epilimnion at the outlet of the lake. Grab samples were also collected from 16
359 additional lakes, including 14 kettle lakes (Table 3). The samples from Lake
360 Témiscamingue were collected from a boat, at a depth of 1 m using a *NASCO Swing*
361 *Sampler* telescopic pole equipped with a 1 L HDPE bottle. The samples from Lake Tee
362 were collected from a boat, approximately 1 m below the surface, using a plastic tubing
363 equipped with an inertial valve. The samples from kettle lakes were collected from the
364 shoreline, approximately 1 m below surface, using a *NASCO Swing Sampler* telescopic
365 pole equipped with a 1 L HDPE bottle.

366 **Table 3. Summary of surface waters sampling conditions**

367 All surface water samples were filtered immediately after collection using 0.45 μm
368 *Waterra FHT*. The samples collected for stable isotopes analyses ($\delta^2\text{H}$ - $\delta^{18}\text{O}$) were stored
369 in 30-60 ml HDPE bottles and kept at 4 °C until analysis.

370 *Springs*

371 Groundwater springs were sampled between July 2009 and September 2011 through the
372 course of a previous study (Castelli 2012). These samples are associated with the
373 Vaudray-Joannès, Launay and Saint-Mathieu-Berry eskers and with the Harricana
374 Moraine. Other spring samples were collected during the summer of 2013. These
375 correspond to springs that are equipped for drinking water supply. All spring samples
376 were filtered immediately after collection using 0.45 μm *Waterra FHT* cartridges. The
377 *in situ* parameters (pH, dissolved oxygen, specific electrical conductivity (SEC), redox
378 potential and temperature) were measured on site using a multi-parameter probe. Further
379 details about the sampling methodology can be found in Castelli (2012). The samples
380 collected for stable isotopes analyses ($\delta^2\text{H}$ - $\delta^{18}\text{O}$) were stored in 30-60 ml HDPE bottles.
381 The samples collected for tritium (^3H) analyses were stored in 1 L HDPE bottles.
382 Sampling bottles for stable isotopes analyses ($\delta^2\text{H}$ - $\delta^{18}\text{O}$) and tritium (^3H) analyses were
383 filled to the top with no head space and kept at 4 °C until analysis.

384 *Groundwater*

385 Municipal, private and observation wells were sampled during this study. The targeted
386 wells were commonly identified using the Quebec Ministry of the Environment water
387 well databases, including the Hydrogeological Information System (HIS).
388 Complementary information about the characteristics of the sampled wells was obtained
389 from the private owners or from municipal employees. Well logs were used to document
390 the characteristics of the sampled observation wells. Two conditions had to be met prior
391 to groundwater samples collection: the infrastructures had to be purged of a minimal
392 volume of water and the monitored *in situ* parameters had to reach predefined stability
393 criteria. Water was therefore systematically purged from domestic and observation wells
394 before sample collection. This procedure allows draining the stagnant water from the
395 infrastructure to obtain a water sample that represents groundwater from the aquifer. For
396 private wells, it was assumed that water was withdrawn on a regular basis and that the
397 minimal purge could be limited to the volume of the pressurized reservoirs. The purging
398 was done from taps supplying untreated water. The minimum purging volume for
399 observation wells corresponded to three times the total volume of water stored within the
400 well and its gravel pack. In municipal wells, it was assumed that water is withdrawn at a
401 sufficient rate to prevent stagnation within infrastructures and that no minimal purge was
402 required. During the purge, *in situ* parameters (pH, dissolved oxygen, SEC, redox
403 potential and temperature,) were measured using a multi-parameter probe. The samples
404 were retrieved only after all of the *in situ* parameters had reached the stability criteria
405 reported in Table 4 for three consecutive measurements separated by five minutes
406 intervals. The samples collected for stable isotopes analyses ($\delta^2\text{H}$ - $\delta^{18}\text{O}$) were stored in
407 30-60 ml HDPE bottles and kept at 4 °C until analysis. The samples collected for tritium
408 (^3H) analyses were stored in 1 L HDPE bottles. Sampling bottles for tritium (^3H) analyses
409 were filled under water in a bucket and capped under water with no head space following
410 a filling procedure described by the USGS and kept at 4 °C until analysis.

411

412

Table 4. Stability criteria for groundwater sample collection

413

414 ***Analytical procedures***

415 Water samples were analyzed for $\delta^2\text{H}$ and $\delta^{18}\text{O}$ at the stable isotopes laboratory of the
416 Geotop-UQAM research center in Montreal, Canada. Measurements were made using a
417 dual inlet *Micromass Isoprime*TM isotope ratio mass spectrometer coupled to an
418 *Aquaprep*TM system. For oxygen and hydrogen isotopic analyses, 200 μl of water were
419 transferred into septum vials and equilibrated in a heated rack with a known volume of
420 CO_2 (H_2 in the case of hydrogen analyses, with a platinum catalyst). The isotopic
421 compositions of samples were corrected using internal reference waters ($\delta^{18}\text{O} = -6.71\text{‰}$,
422 -13.98‰ and -20.31‰ ; $\delta^2\text{H} = -51.0\text{‰}$, -99.0‰ and -155.4‰) calibrated on the VSMOW-
423 SLAP scale (Coplen 1996). Reference waters were systematically analyzed between
424 series of samples in order to check for instrumental stability. The analytical uncertainty is
425 $\leq 1\text{‰}$ for $\delta^2\text{H}$ and $\leq 0.05\text{‰}$ for $\delta^{18}\text{O}$ at the 1σ level. Values are reported in permil units
426 (‰) against the Vienna Standard Mean Ocean Water standard (VSMOW). Water samples
427 for tritium (^3H) analyses were measured by liquid scintillation counting (LSC) at the
428 Environmental Isotope Laboratory of the University of Waterloo, Canada. The detection
429 limit for enriched samples is 0.8 tritium units (TU).

430 ***Data organization and statistical analyzes***

431 In order to support the interpretation of results, the isotopic data are organized according
432 to the main hydrological components and aquifers that were presented previously
433 (Table 1; Figure 2). Overall, eight distinct groups are identified for segregating the
434 isotopic data. The latter include (with numbers corresponding to those reported in
435 Figure 2) samples corresponding to (1) precipitation, (2) the snowpack, (3) surface
436 waters, (4) springs, (5) unconfined granular aquifers, (6) confined granular aquifers, (7)
437 unconfined fractured rock aquifers and (8) confined fractured rock aquifers. The basic
438 statistical analyses conducted on the dataset and related plots were performed using the
439 *Statistica version 6* (StatSoft Inc., 2004) and *Microsoft Excel* software.

440

441 Results and discussion

442 The entire dataset associated with stable isotopes analyses is shown in Figure 3a, whereas
443 Figure 3b allows a better representation of surface and groundwater data. Figure 3c
444 presents the average values of groundwater and surface water groups in the same range as
445 Figure 3b. As a complement, Figure 4 presents boxplots showing the non-outlier range,
446 25th and 75th percentiles and median SEC, $\delta^{18}\text{O}$ and ^3H values for selected groups of data.
447 The non-outlier range is defined as maximum and minimum values within 1.5 times the
448 interquartile range. It was chosen not to show the data outside of the non-outlier range in
449 the boxplots (Figure 4) in order to allow a better representation of the data. Nevertheless,
450 the entire dataset is shown in Figure 3. Results associated with three of the main
451 components of the water cycle (namely precipitation, surface waters and groundwater)
452 are first discussed separately to address specific topics and jointly interpreted afterward to
453 propose a regional-scale conceptual model of groundwater flow.

454

455 **Figure 3. Isotopic composition of precipitation, surface waters, springs and**
456 **groundwater**

457

458 **Figure 4. $\delta^{18}\text{O}$ (a; b), ^3H (c) and SEC (d) boxplots for precipitation, surface waters,**
459 **springs and groundwater**

460

461 *The isotopic composition of precipitation: air masses and seasonal patterns*

462 The $\delta^2\text{H}$ and $\delta^{18}\text{O}$ values of precipitation ranged from -217.9‰ to -29.9‰ and -28.4‰ to
463 -4.8‰ respectively (Figure 3a). These data are used to define the Local Meteoric Water
464 Line, calculated using precipitation (n=164) and snowpack data (n=31) (LMWL; $\delta^2\text{H} =$
465 $7.84 \delta^{18}\text{O} + 9.20 \text{‰}$ (n = 195, $r^2 = 0.99$); Figure 3). The LMWL is relatively close to
466 Craig's global meteoric water line (GMWL; $\delta^2\text{H} = 8 \delta^{18}\text{O} + 10 \text{‰}$ SMOW; Craig 1961).
467 In terms of temporal variations, the monthly precipitation data revealed the expected
468 temperature-dependant isotopic cycle, with heavy isotopes depletion during the colder

469 period and enrichment during the warmer period (Figure 5). Such a pattern is typical of
470 boreal regions with summer rains enriched in heavy isotopes and winter precipitation
471 depleted in heavy isotopes in response to strong seasonal variations in temperature (Clark
472 and Fritz 1997). Overall, the temporal variations in the isotopic composition of
473 precipitation reflect the marked seasonality of the study area.

474 The weighted mean annual isotopic compositions of precipitation at stations 1, 3 and 4
475 are shown in Table 5. The data do not show a clear relationship between precipitation
476 $\delta^2\text{H}$ - $\delta^{18}\text{O}$ and latitude. The absence of a latitudinal gradient in precipitation isotopic
477 composition could reflect the superimposed influence of the three major air masses that
478 supply moisture over the study area, namely (1) the Arctic air mass (from the north),
479 (2) the tropical vapor (from the south) and (3) the westerlies (Bryson 1966; Fritz et al.
480 1987). Given the relatively flat landscape of the study area, the effect of altitude on
481 precipitation $\delta^2\text{H}$ - $\delta^{18}\text{O}$ values is probably limited. Finally, evaporation processes might
482 also influence the isotopic composition of precipitation at the local scale, near large water
483 bodies such as the Témiscamingue, Simard and Abitibi lakes (Figure 1). Further data
484 would be required to better explain spatial variations in the isotopic composition of
485 precipitations at the scale of the study area.

486 Samples collected from the snowpack in March 2014 (prior to the onset of spring
487 snowmelt) revealed $\delta^2\text{H}$ and $\delta^{18}\text{O}$ values ranging from -184.4‰ to -151.6‰ and -24.5‰
488 to -20.6‰ respectively (Figure 3). Ice layers have been found in some places in the
489 snowpack during sample collection. It is recognized that isotopic fractionation occurs
490 during snowmelt (Taylor et al. 2001; Laudon et al. 2002). This suggests that the different
491 layers identified within the sampled snowpack (which reflect melting / refreezing events)
492 are most likely characterized by variable isotopic compositions. Nevertheless, here, the
493 bulk samples (i.e. composite samples of the entire thickness of snow) collected from the
494 snowpack prior to the onset of the snowmelt event plot uniformly on the left side of the
495 LMWL (Figure 3), with values similar to that measured at the precipitation monitoring
496 stations. This suggests that the bulk of the snowpack preserved an isotopic composition
497 that is similar to that of precipitation prior to the onset of the spring snowmelt event.

498 Overall, the isotopic compositions of samples retrieved from the snowpack do not define
499 a clear latitudinal gradient at the regional scale.

500

501 **Table 5. Amount weighted average isotopic composition of precipitation at the four**
502 **sampling stations (2013-2015).**

503

504 **Figure 5. Seasonal variations in precipitation amounts and $\delta^{18}\text{O}$ composition**

505

506 *The isotopic composition of surface waters: inflows, evaporative enrichment and mixing*

507 The regional dataset reveals a significant variability in surface waters $\delta^2\text{H}$ and $\delta^{18}\text{O}$, with
508 values ranging from -100.8‰ to -68.9‰ and -13.7‰ to -7.6‰, respectively (for regional
509 average values of -80.9‰ ($\delta^2\text{H}$) and -10.7‰ ($\delta^{18}\text{O}$)). The isotopic composition of surface
510 waters can present both spatial and temporal variations owing to the superimposed
511 influence of precipitation, groundwater inflow, evaporation and tributary mixing
512 processes (Yi et al. 2010). Temporal variations are commonly observed in regions
513 characterized by a marked seasonality where heavy isotope depletion following snowmelt
514 and enrichment caused by evaporation during the ice off period have been reported (e.g.
515 see Telmer and Veizer 2000; Yi et al. 2010). Here, most of the samples were collected in
516 summer (Table 2), and the data plot below the LMWL, consistent with the hypothesis
517 that surface waters have undergone heavy isotope enrichment owing to evaporation. A
518 surface Local Evaporation Line (LEL): $\delta^2\text{H} = 3.80 \delta^{18}\text{O} - 40.22 \text{‰}$ ($n = 99$, $r^2 = 0.65$)
519 (Figure 3a) is evaluated from the data. The evaporation over inflow ratio (E/I), as
520 described by Gibson and Edwards (2002), corresponds to the water balance of a surface
521 water body and relates to the combined effects of evaporative heavy isotope enrichment
522 and dilution by inflowing waters. The E/I ratio is used here for evaluating the
523 hydrological processes responsible for the distribution of surface water data plotting
524 below the LMWL in $\delta^2\text{H}$ vs $\delta^{18}\text{O}$ graphs (Figure 3). The *Hydrocalculator* application
525 from Skrzypek et al. (2015) was used to calculate evaporation over inflow (E/I) ratios

526 under steady state conditions (shown over the LEL in Figure 6a). This application solves
 527 a modified form of the [Craig and Gordon \(1965\)](#) model:

$$528 \quad \frac{E}{I} = \left[\frac{(\delta_L - \delta_P)}{(\delta^* - \delta_L) \times m} \right] \quad \text{Equation 2}$$

529 Where δ_L corresponds to the isotopic composition of the sampled surface water, δ_P is the
 530 initial isotopic composition of water and δ^* is the limiting isotopic enrichment factor (a
 531 function of the isotopic fractionation factor and atmospheric isotopic composition and
 532 relative humidity). Similar approaches are used to evaluate the E/I ratios of rivers
 533 (e.g.: [Telmer and Veizer 2000](#)) and lakes (e.g. [Gibson and Edwards 2002](#)). Here,
 534 theoretical calculations are proposed for points falling directly on the LEL evaluated from
 535 the dataset, the latter corresponding to the δ_L term of equation 2. The initial isotopic
 536 composition of water (δ_P) is assumed to correspond to the intersection between the
 537 LMWL and the LEL. The isotopic composition of atmospheric moisture (δ_A), which is
 538 needed to evaluate δ^* in equation 2, is evaluated according to three scenarios:

- 539 1- δ_A is in equilibrium with the flux weighted average isotopic composition of
 540 precipitation between the months of May and August ($\delta_A = -141.21 \text{ ‰}$; -19.45
 541 ‰);
- 542 2- δ_A is in equilibrium with the arithmetic average isotopic composition of
 543 precipitation between the months of May and August ($\delta_A = -137.42 \text{ ‰}$; -18.91
 544 ‰);
- 545 3- δ_A is in equilibrium with δ_P ($\delta_A = -156.03 \text{ ‰}$; -21.63 ‰).

546 A temperature of 22 °C and a relative humidity of 70% are assumed, consistent with the
 547 conditions encountered during the warm ice off period within the study region. The
 548 isotope fractionation factors are calculated from the [Horita and Wesolowski \(1994\)](#)
 549 equations.

550 Depending on the model parameters used, E/I ratios ranging between 0% and 36% are
 551 evaluated from the regional dataset (Figure 6a). Based on this theoretical framework,
 552 surface water samples plotting closer to the LMWL most likely present the lowest E/I

553 ratios. These surface waters might be affected by significant evaporation losses, but
554 inflows (mostly precipitation and groundwater, both plotting along the LMWL) are
555 sufficient to buffer the heavy isotope enrichment resulting from evaporation. On the
556 contrary, the samples that are most enriched in heavy isotopes (i.e. plotting further along
557 the right side of the LEL) most likely present the highest E/I ratios. Within the regional
558 hydrogeological framework, it is most likely that evaporation mainly occurs within lakes
559 and stagnant water bodies whereas dynamic in stream processes mainly reflect surface
560 water / groundwater interactions and tributary mixing.

561 One distinctive feature of the regional dataset is that most of the kettle lakes plot on the
562 right side of the LEL, suggesting overall high E/I ratios in comparison to other surface
563 waters (Figure 6b). Recent studies by Isokangas et al. (2015) and Arnoux et al. (2017)
564 provided novel methods to estimate the groundwater dependency of such lakes based on
565 E/I ratios calculated from stable isotope data, among other. Here, quantifying the
566 groundwater dependency of lakes is precluded by uncertainties related to volumes,
567 surface water inflows and potential heterogeneities caused by thermal stratification and
568 temporal variations. Nevertheless, the distinctive isotopic composition of kettle lakes
569 stresses the need for conducting further studies aimed at quantifying their water balance
570 to better anticipate their sensitivity to climate change and human pressure.

571

572 **Figure 6. Interpretation of the isotopic compositions of surface waters**

573

574 *The isotopic composition of groundwater and springs: recharge, discharge, mixing and regional*
575 *flowpaths*

576 The stable isotopes analyses ($\delta^2\text{H}$ - $\delta^{18}\text{O}$) conducted on 351 groundwater and springs
577 samples are shown on the graphs of Figures 3, 4a and 4b. The data associated with
578 groundwater samples are represented according to the four main types of regional
579 aquifers, as shown in Figure 2 (unconfined granular aquifers; unconfined fractured rock
580 aquifers; confined granular aquifers; confined fractured rock aquifers). The $\delta^2\text{H}$ and $\delta^{18}\text{O}$
581 values of groundwater samples ranged from -107.8‰ to -71.9‰ and -15.2‰ to -9.4‰,

582 respectively, for a regional average of -13.3‰ ($\delta^{18}\text{O}$); -93.4‰ ($\delta^2\text{H}$). The $\delta^2\text{H}$ and $\delta^{18}\text{O}$
583 values of spring samples ranged from -106.3‰ to -75.1‰ and -14.4‰ to -9.7‰ ,
584 respectively for a regional average of -13.8‰ ($\delta^{18}\text{O}$); -98.1‰ ($\delta^2\text{H}$). Previous work by
585 Castelli (2012) revealed that the isotopic composition of springs is fairly stable
586 (1) throughout the ice-off season and (2) from year to year. Similarly, it is assumed here
587 that the isotopic composition of groundwater is relatively constant in time. The focus is
588 therefore set on spatial rather than temporal variations in isotopic compositions. The data
589 associated with groundwater and springs are evenly distributed around the LMWL. This
590 suggests that (1) infiltration and recharge processes are not significantly affected by
591 evaporation losses and (2) aquifers are not significantly affected by inflows from
592 evaporated surface waters. This is consistent with the regional hydrogeological
593 framework because: (1) the main recharge areas consist of coarse grained glaciofluvial
594 formations where high infiltration rates prevent surface ponding (and associated
595 evaporation) and (2) physical measurements suggest that groundwater generally
596 discharges into surface waters at the regional scale (Cloutier et al. 2013; 2015). Although
597 water losses by plant transpiration prior to groundwater recharge might be significant, the
598 available dataset does not allow the quantification of this process. Results also support
599 the hypothesis that regional surface waters – groundwater interactions are essentially
600 restricted to groundwater discharging into surface waters, and not the inverse (otherwise,
601 some groundwater samples would most likely be plotting significantly below the LMWL
602 owing to mixing with evaporated surface waters). The data do not show systematic
603 latitudinal trends in the stable isotope composition of groundwater at the regional scale.
604 This is most likely caused by the complex mixing of recharge originating from the main
605 atmospheric air masses supplying precipitation over the study region.

606 The data associated with the four main types of aquifers present a significant dispersion
607 and overlap within Figures 3 and 4b. Nevertheless, two main observations arise from the
608 distribution of data within the $\delta^2\text{H}$ - $\delta^{18}\text{O}$ plots:

609 (1) The data associated with unconfined granular aquifers present a greater
610 dispersion in comparison to the other types of aquifers (Figure 4b);

611 (2) Confined granular aquifers and fractured rock aquifers (both confined and
612 unconfined) tend to plot further on the right side of LMWL in comparison to
613 springs and unconfined granular aquifers (Figure 3c).

614 These observations are interpreted in an effort to further constrain the understanding of
615 regional flowpaths within the hydrogeological environment conceptually represented in
616 Figure 2. The unconfined granular aquifers represent the main regional recharge zones
617 where rain and melted snow preferentially infiltrate at different periods of the year,
618 depending on local climatic and hydrological conditions (Cloutier et al. 2016). The
619 proposed interpretation of groundwater stable isotope data is that unconfined granular
620 aquifers present a greater $\delta^2\text{H}$ - $\delta^{18}\text{O}$ variability owing to a shorter residence time of water
621 within shallow and dynamic systems such as eskers and moraines. The results associated
622 with ^3H analyzes (Figure 4c) further support this interpretation. Overall, the tritium data
623 suggest that water samples from springs, unconfined granular aquifers and unconfined
624 fractured rock aquifers correspond to modern waters (with values ranging between
625 approximately 9 and 17 TU, consistent with the subdivisions proposed by Clark and Fritz
626 1997). Such results suggest a relatively short residence time of water in these shallow
627 components. In contrast, samples collected within confined aquifers (both granular and
628 fractured) present tritium contents below 0.8 TU (consistent with submodern values
629 according to the subdivisions proposed by Clark and Fritz 1997). Such results suggest
630 that within the region, confined aquifers generally contain older groundwater in
631 comparison to unconfined aquifers.

632 The sampled springs are all located on the flanks of unconfined granular aquifers
633 associated with glaciofluvial formations (eskera / moraines). Their heavy isotope
634 depletion (Figure 3c) with respect to most groundwater samples suggests that superficial
635 flow systems associated with springs are submitted to a greater influence of lighter
636 precipitation. As previously discussed, within the study area, precipitations depict a clear
637 seasonal pattern, with lighter values recorded during the cold period (Figure 5). Under
638 such conditions, one likely explanation is that a significant proportion of the groundwater
639 recharge associated with snowmelt contributes to springs discharge. This stands as a
640 plausible explanation because the springs located on the flanks of eskers and moraines

641 essentially consist in «overflows» allowing groundwater to discharge from unconfined
642 granular aquifers towards the surface water network. From a physical perspective, it
643 seems likely that «overflow» conditions would be favored following the thawing period,
644 as groundwater levels are generally raising within the eskers and moraines of the region
645 during this period (Cloutier et al. 2013; 2015). During the dryer summer period, recharge
646 rates are likely reduced, therefore limiting the supply of heavier water associated with
647 summer precipitation to shallow flow systems. In fall, at the end of the warmer period,
648 when the uptake of water by vegetation is reduced and precipitation are increased,
649 recharge rates most likely increase, further supplying water that is depleted in heavy
650 isotopes to the shallow flow systems associated with springs. The loss of lighter isotopes
651 resulting from groundwater discharge within springs could also partly explain why the
652 confined aquifers, located further downstream within the flow systems, are slightly
653 enriched in heavy isotopes with respect to springs and shallow unconfined aquifers
654 (Figures 3c; 4b). One key regional observation discussed by Cloutier et al. (2016) is that
655 extensive peatlands are developed on the flanks of several eskers and moraines of the
656 region. The contacts between esker/moraines and these extensive peatlands are
657 interpreted as diffuse groundwater seepage zones (Nadeau et al. 2015). The process
658 related to the loss of lighter isotopes within punctual springs might therefore be
659 generalized to most of the unconfined aquifers associated with extensive peatlands and
660 diffuse groundwater seepage zones. In addition, it seems most likely that the areas
661 associated with unconfined fractured rock aquifers are less prone to groundwater
662 recharge during snowmelt in comparison to the areas associated with glaciofluvial
663 formations (i.e. unconfined granular aquifers). This hypothesis is based on the contrast in
664 hydraulic conductivity (K) between the fractured bedrock (lower K) and the eskers and
665 moraines (higher K) (e.g. see Cloutier et al. 2015). The differences in the isotopic
666 composition of unconfined granular aquifers and springs compared to that of confined
667 granular aquifers and fractured rock aquifers (both confined and unconfined) could
668 therefore reflect the relative contribution of snowmelt induced recharge for the different
669 types of aquifers.

670 The interpretations discussed above are further illustrated in Figure 7a, where the SEC of
671 springs and groundwater is plotted against the corresponding $\delta^{18}\text{O}$ values. The data

672 presented therein reveals that confined fractured rock aquifers generally present the
673 highest SEC values whereas the unconfined granular aquifers present the lowest SEC
674 values. This is also depicted in Figure 4d where a gradual increase in SEC is observed
675 from unconfined towards confined aquifers. SEC can thus be used as a proxy for
676 groundwater residence time and geochemical evolution, higher SEC generally implying
677 more evolved groundwater. This pattern most likely reflects the combined effects of
678 (1) increased geochemical interactions between groundwater and the geological
679 environment along regional flowpaths and (2) recharge of precipitation having very low
680 dissolved solid concentration (thus quite low SEC) in unconfined granular aquifers areas.
681 The mixing model shown in Figure 7a further supports the interpretations proposed
682 above. Three end-members are proposed to explain the variability in both $\delta^{18}\text{O}$ and SEC
683 values:

- 684 1- End-member «A» presents low SEC and $\delta^{18}\text{O}$, it is identified as representative of
685 recent snowmelt induced recharge. The lowest non-outlier $\delta^{18}\text{O}$ value recorded in
686 unconfined granular aquifers (lower end of the boxplot shown in Figure 4b) is
687 used to characterize this end-member, whereas the SEC value is set to 0.05
688 mS/cm, corresponding to the lowest non-outlier SEC value recorded in
689 unconfined granular aquifers (lower end of the boxplot shown in Figure 4d);
- 690 2- End-member «B» presents low SEC and high $\delta^{18}\text{O}$ close to -12‰, it is identified
691 as representative of fall recharge affected by relatively high temperature rain. The
692 highest non-outlier $\delta^{18}\text{O}$ value recorded in unconfined granular aquifers (higher
693 end of the boxplot shown in Figure 4b) is used to characterize this end-member,
694 whereas the SEC value is set to 0.05 mS/cm, as previously defined for end-
695 member «A»;
- 696 3- End-member «C» presents high SEC and $\delta^{18}\text{O}$, it is identified as representative of
697 well mixed water (from various recharge zones or various recharge periods) that
698 is geochemically evolved owing to prolonged water-rock interactions along
699 flowpaths, thus leading to a higher SEC. The median $\delta^{18}\text{O}$ isotopic composition of
700 confined fractured rock aquifers is used to constrain this end-member, and the
701 SEC is set to 1.8 mS/cm (the highest recorded value in the dataset, Figure 4d).

702 The dashed lines in figure 7 a and b represent mixing lines between the three end-
703 members, at 25% intervals. In this mixing model, the SEC is considered linearly related
704 to the geochemical level of evolution of groundwater, which is most probably not linearly
705 related to the groundwater residence time.

706 Six main outliers are identified among the groundwater/springs dataset in Figures 3b.
707 These six points plot below the LMWL in Figure 3, suggesting an evaporative
708 enrichment that is not observed in the other groundwater samples. Five of these points are
709 associated with samples collected in the immediate vicinity of lakes whereas the fifth one
710 was collected within a mining site. It is proposed that the five samples collected near
711 lakes and presenting heavy isotope enrichment and low SEC values are directly affected
712 by the supply of evaporated and low SEC lake waters. The wells characterized by low
713 SEC and low $\delta^{18}\text{O}$ values are likely affected by nearby snowmelt induced groundwater
714 recharge, whereas those with low SEC and significantly higher $\delta^{18}\text{O}$ values are likely
715 affected by mixing with surface waters. In that sense, within the study region, stable
716 isotope data could be used as a screening tool to detect groundwater wells that are
717 affected by surface waters. Such wells would deserve a specific attention in groundwater
718 monitoring programs because they might be more vulnerable to microbial contamination
719 originating from surface waters. Finally, the sixth outlier point was collected in a
720 groundwater well located within a mining site; its isotopic composition and SEC value
721 most likely reflect anthropogenic effects.

722

723 **Figure 7. Specific electrical conductivity (SEC) vs $\delta^{18}\text{O}$ of springs, surface waters**
724 **and groundwater**

725

726 *Local and regional systems of groundwater flow: insights from water stable isotopes*

727 Tóth (1963) proposed a conceptual model of groundwater flow in 2D cross sections
728 involving local, intermediate and regional groundwater flow sub-systems. Such
729 conceptual models constitute a central unifying concept to describe groundwater

730 recharge, flow and discharge processes at various scales and provide a framework that is
731 of utmost importance to interpret regional hydrogeochemical datasets. Tóth (1999)
732 further developed his conceptual model by presenting groundwater as a geologic agent
733 interacting with the environment and inheriting a geochemical composition that preserves
734 an archive of hydrogeochemical processes and systematically evolves along flow paths.
735 Following Toth's pioneer work, the different flow systems of the study area are
736 interpreted with a focus on the isotopes of the water molecule itself, rather than other
737 dissolved species. The working hypothesis is that within the considered spatiotemporal
738 domain, the isotopic composition ($\delta^2\text{H}-\delta^{18}\text{O}$) of water is not significantly affected by
739 water-rock interactions but rather mainly reflects the superimposed effects of
740 precipitation isotopic composition, evaporation, recharge, mixing and discharge
741 processes that vary in time and space. The average $\delta^2\text{H}-\delta^{18}\text{O}$ compositions of
742 precipitation (including snow samples), surface waters, springs and groundwater are
743 shown in Figure 8, which consists in a more detailed representation of the regional
744 hydrogeological conditions that was initially conceptualized in Figure 2.

745 The conceptual model illustrates precipitation showing the greatest isotopic variability
746 and recharge occurring mainly within shallow unconfined aquifers, where the variability
747 in the isotopic composition of water is attenuated with respect to precipitation due to
748 mixing. Part of the water flowing within these unconfined aquifers discharges towards
749 surface waters through springs and diffuse seepage zones that are generally depleted in
750 heavy isotopes with respect to the average composition of groundwater. Surface waters
751 are subsequently submitted to evaporation (causing heavy isotope enrichments) and to
752 further dilution by precipitation and groundwater discharge. The remaining groundwater
753 flows from unconfined recharge areas towards confined aquifers, allowing two key
754 geochemical processes to occur along flow paths:

755 (1) Water rock interaction: groundwater becomes increasingly concentrated in
756 dissolved ions owing to prolonged water-rock interactions, as shown by the
757 gradual increase in SEC among the various hydrological components, from
758 precipitations towards confined fractured rock aquifers (Figure 4d). This
759 interpretation is further supported by the tritium analyses (Figure 4c), which

760 suggest that confined aquifers are characterized by longer water residence times
761 compared to springs and unconfined aquifers;

762 (2) Mixing: The isotopic composition of groundwater evolves towards a regional
763 average owing to the mixing of water originating from various recharge zones or
764 various recharge periods. This is illustrated in Figure 4b and Figure 7a, which
765 suggest that mixing processes tend to damp the variability recorded in $\delta^{18}\text{O}$ values
766 within confined aquifers. In other terms, the greater the proportion of unconfined
767 aquifers in the vicinity on a given site, the greater the potential variability in the
768 stable isotope composition of groundwater.

769 Clark and Fritz (1997) proposed a schematic representation of the attenuation of seasonal
770 isotope variations in recharge waters within the unsaturated zone. These authors reported
771 that temporal isotopic variations around an average value are gradually damped with
772 increasing depth as a function of flowpaths lengths and residence times. A comparable
773 interpretation is proposed here to explain the evolution of groundwater SEC and
774 $\delta^2\text{H}-\delta^{18}\text{O}$ along local, intermediate and regional flow systems (Figures 7 and 8). On one
775 hand, since recharge processes are primarily associated with sectors of unconfined
776 aquifers and partly controlled by hydrogeological conditions that are highly
777 heterogeneous at the regional scale, a significant variability is recorded in unconfined
778 aquifers isotopic compositions. Overall, the unconfined granular aquifers associated with
779 coarse grained glaciofluvial formations (eskers and moraines) are more prone to recharge
780 during snowmelt and therefore present low $\delta^2\text{H}-\delta^{18}\text{O}$ values in comparison to the other
781 types of aquifers. Similarly, the springs found on the flanks of eskers and moraines are
782 depleted in heavy isotopes. On the other hand, recharge areas associated with unconfined
783 bedrock aquifers are more prone to runoff generation during the intense snowmelt event,
784 and are therefore generally enriched in heavy isotopes in comparison to unconfined
785 granular aquifers. Finally, the confined aquifers located further along groundwater
786 flowpaths are fed by numerous interconnected unconfined aquifer areas that are
787 heterogeneously distributed over the region. The result of the mixing processes occurring
788 along flowpaths is a gradual evolution towards a regional average value within confined
789 aquifers. Based on this interpretation, groundwater samples from confined aquifers that

790 present an isotopic composition that significantly differs from the regional average most
791 likely represents sites where the preferential influence of a nearby recharge area are
792 recorded.

793

794 **Figure 8. Isotopic variations within local, intermediate and regional flow systems**

795

796 **Conclusions**

797 The isotopic composition of water ($\delta^2\text{H}$ - $\delta^{18}\text{O}$ and ^3H) in precipitation, surface waters, groundwater
798 and springs was used to improve the understanding of regional flow systems in the southern portion
799 of the Barlow-Ojibway Clay Belt. The data allowed the identification of the key hydrological
800 processes driving spatiotemporal isotopic variations and deciphering the interaction pathways
801 between the various components of the water cycle, from atmospheric moisture to groundwater
802 discharge:

803 (1) Regional precipitations depict a clear seasonal pattern with heavy isotope depletion
804 recorded in winter and heavy isotope enrichment in summer. The data did not reveal clear
805 spatial trends in precipitation $\delta^2\text{H}$ - $\delta^{18}\text{O}$, most likely in response to the complex mixing of
806 inputs from the main air masses supplying moisture to the study region. Overall,
807 precipitation data allowed the definition of the local meteoric water line (LMWL:
808 $\delta^2\text{H} = 7.84 \delta^{18}\text{O} + 9.20 \text{‰}$; $n = 164$; $r^2 = 0.99$);

809 (2) Surface waters plot below the LMWL in a $\delta^2\text{H}$ - $\delta^{18}\text{O}$ graph owing to evaporation losses.
810 Calculations conducted using a modified form of the Craig and Gordon (1965) model with
811 the *Hydrocalculator* application (Skrzypek et al. 2015) provided an estimation of
812 evaporation over inflow (E/I) ratios ranging between 0% and 36% at the regional scale;

813 (3) Groundwater samples plot on the LMWL and are evenly distributed around the average
814 isotopic composition of precipitation in a $\delta^2\text{H}$ - $\delta^{18}\text{O}$ graph, suggesting that evaporation
815 processes do not significantly affect regional recharge waters. The $\delta^2\text{H}$ - $\delta^{18}\text{O}$ of
816 groundwater in shallow unconfined aquifers presents a greater variability in comparison to
817 confined aquifers owing to mixing processes that tend to buffer the isotopic variability

818 along the regional groundwater flow paths. This buffering on the isotopic composition
819 variability is accompanied by increasing values of the SEC of groundwater from
820 unconfined to confined aquifers, most likely in response to increasing water-rock
821 interaction timescales, consistent with generally lower tritium contents in confined aquifers;

822 (4) Unconfined granular aquifers and associated springs are generally slightly depleted in
823 heavy isotopes in comparison to other groundwater samples, suggesting that these shallow
824 flow systems are preferentially recharged during spring snowmelt within the region.

825 Overall, the observations and interpretations of this study support an improvement of a conceptual
826 hydrogeological model of local and regional hydrological systems and flow paths that is consistent
827 with the geological framework. This conceptual model illustrates how water isotopes can be used to
828 constrain physically-based interpretations of local, intermediate and regional flow systems such as
829 those inspired by the pioneering work of Tóth (1963), even in a highly heterogeneous region.
830 The buffering of isotopic variability in groundwater from groundwater recharge areas
831 located in unconfined aquifers towards groundwater discharge areas and confined
832 aquifers located further along regional flow paths provides a framework for deciphering
833 the processes associated with the geochemical evolution of groundwater, as a
834 complement to understandings based on dissolved major ions and trace elements.

835

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850

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Domains	Hydrological components	Geological units	Hydrogeological units and aquifers
Atmosphere	(1) Precipitation (rain & snow)	-	-
Surface	(2) Snowpack (3) Surface water (4) Spring	-	-
Subsurface (groundwater)		<u>C1</u> Eskers & moraines	(5) Unconfined granular aquifer
		<u>E</u> Surface sediments near eskers	
		<u>D</u> Glaciolacustrine clay & silt	Regional aquitard
		<u>C2</u> Sediments underlying clay	(6) Confined granular aquifer
		<u>B</u> Till	-
		<u>A</u> Fractured rock	(7) Unconfined or (8) confined fractured rock aquifer

Type of samples	Number of samples (n)	Sampling timeline
Precipitation	164	Monthly samples at four stations 2009-2015
Snowpack	31	March 2014 (n=31)
Springs	76	Summer 2009 (n=30) Summer-Autumn 2010 (n=41) Summer-Autumn 2013 (n=5)
Kettle lakes	14	Summer 2009 (n=14)
Surface waters	85	Summer-Autumn 2013 (n=85)
Confined fractured rock aquifer	105	Summer-Autumn 2006 (n=14) Summer-Autumn 2010 (n=26) Summer 2013 (n=65)
Unconfined fractured rock aquifers	38	Autumn 2006 (n=2) Summer 2010 (n=7) Summer-Autumn 2013 (n=29)
Confined granular aquifers	53	Summer-Autumn 2006 (n=2) Summer-Autumn 2010 (n=15) Summer 2013 (n=36)
Unconfined granular aquifers	79	Summer-Autumn 2006 (n=2) Summer 2009 (n=14) Summer-Autumn 2010 (n=25) Autumn 2011 (n=15) Summer-Autumn 2013 (n=23)
Total	645	2006-2015

Surface waters	Names	Sampling techniques	Dates
Kettle lakes	-	1 m from the lake shore / Telescopic pole / 0.5-1.5 m below surface	Summer 2009
Fault Lakes	Opasatica		Summer 2013
	Rémigny		
	Témiscamingue	Boat / Telescopic pole / 0.5-1.5 m below surface	
Lakes on clay	Tee	Boat / Plastic tubing equipped with an inertial valve / 0.5-1.5 m below surface	Autumn 2013
	Macamic	Sampling at the outlet of the lake / Telescopic pole / 0.5-1.5 m below surface	Summer 2013
	Abitibi		
	Outaouais	Boat / Telescopic pole / central part of the main channel / 0.5-1.5 m below surface	
Duparquet			
Rivers	La Sarre		Summer 2013
	From lac Mance		
	Between Beauchastel and Montbeillard lakes	Sampling near the central part of the main channel / Telescopic pole or Bailer / 0.5-1.5 m below surface	
	Between Routier and Vallet lakes		
	Between Rondelet and Des Quinze lakes		
Small streams	Between two lakes (Mont Brun sector)		Summer/Autumn 2013
	-	Sampling near the central part of the main channel / Telescopic pole or Bailer / 0.5-1.5 m below surface	

In situ parameters	Stability criteria
Temperature (°C)	± 0.1
pH (pH unit)	± 0.1
Specific electrical conductivity (µS/cm)	± 2%
Dissolved oxygen (mg/L)	± 0.2
Redox potential (mV)	± 10
Turbidity (visual observation)	Constant color/opacity

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Stations	n	$\delta^2\text{H}$ (‰ VSMOW)	$\delta^{18}\text{O}$ (‰ VSMOW)
Station 1 (Béarn)	24	-91.9	-13.0
Station 2 (Montbeillard)	13	NA	NA
Station 3 (Sainte-Hélène-de-Mancebourg)	24	-100.9	-14.2
Station 4 (Amos)	24	-84.5	-12.0

Refer to Figure 1 for stations locations; n: number of samples.

NA: averages values are not provided owing to a lack of data from November to April.

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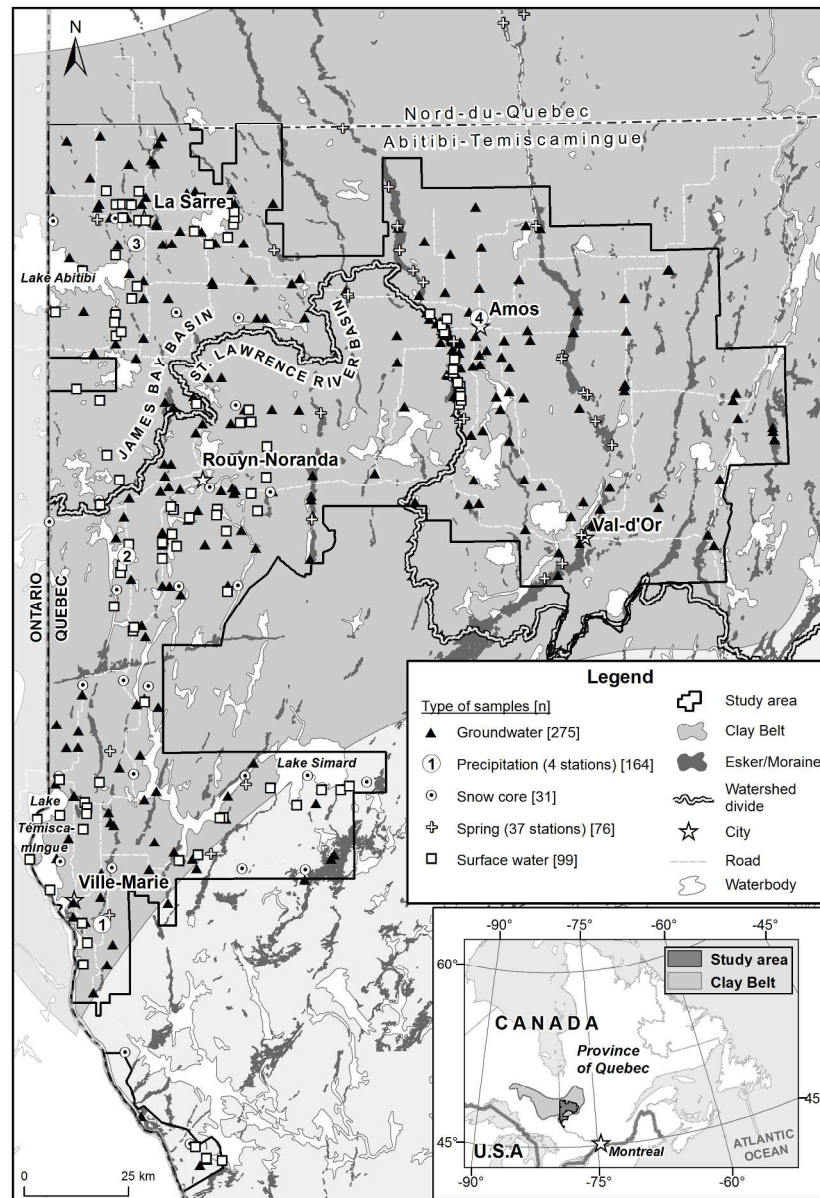


Figure 1. Study area with the location of sampling sites. The number (n) of available samples is shown brackets in the legend. The limits of the study area correspond to that of previous studies from Cloutier et al. (2013; 2015). The continental water divide between the St. Lawrence and the James Bay basins is also shown.

190x270mm (300 x 300 DPI)

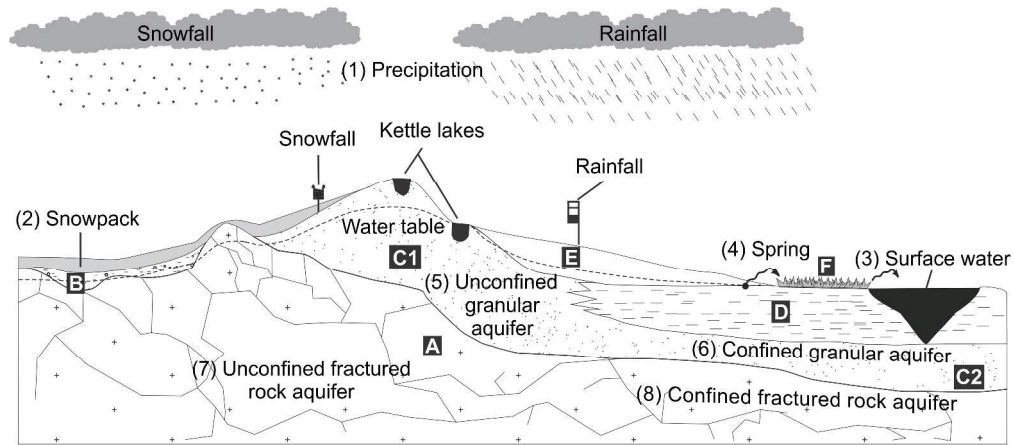


Figure 2. Regional hydrogeological conceptual model and water sampling scheme used to characterize the isotopic signature of all hydrological system components. (A) Fractured bedrock; (B) Till; (C1) Eskers and moraines; (C2) glaciofluvial sediments found between eskers and moraines, under the clay plain; (D) fine grained deep-water glaciolacustrine sediments; (E) Sublittoral sands and beach or eolian sediments; (F) Organic deposits. The numbers (1 to 8) correspond to the sampled hydrological components and hydrogeological units, as identified in table 1.

417x182mm (300 x 300 DPI)

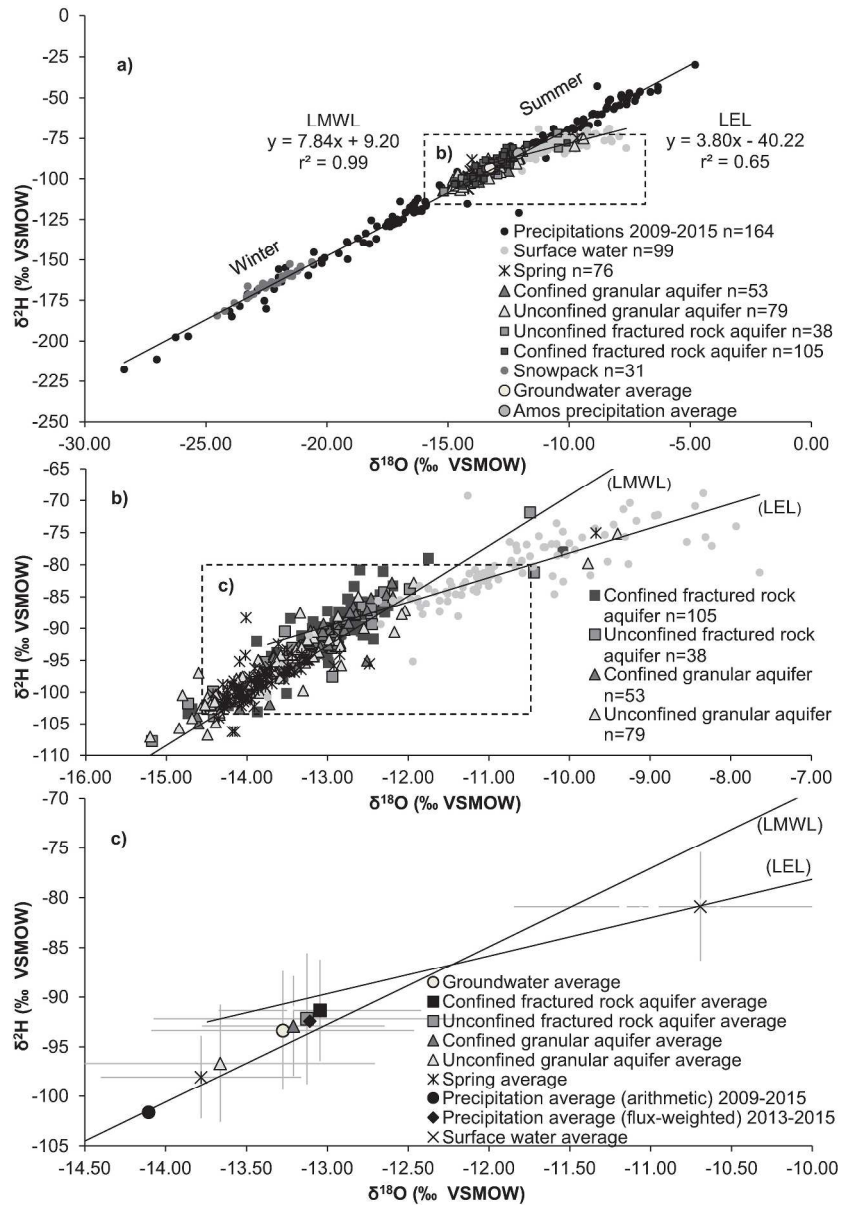


Figure 3. Isotopic composition of precipitation, surface waters, springs and groundwater. All data are shown in (a) whereas figure (b) provides a zoom on the surface and groundwater data. Average values are provided in figure (c). The flux-weighted average for precipitation is based on the three stations (Sainte-Hélène-de-Mancebourg, Amos, Bearn) where precipitation samples were collected between 2013-2015. The value correspond to the mean of the flux-weighted averages calculated for each station. The arithmetic average calculated for all precipitation samples is also shown. The error bars in (c) correspond to the standard deviation calculated for each sample series. The local meteoric water line (LMWL) and local evaporation line (LEL) derived from the data are also shown.

275x397mm (300 x 300 DPI)

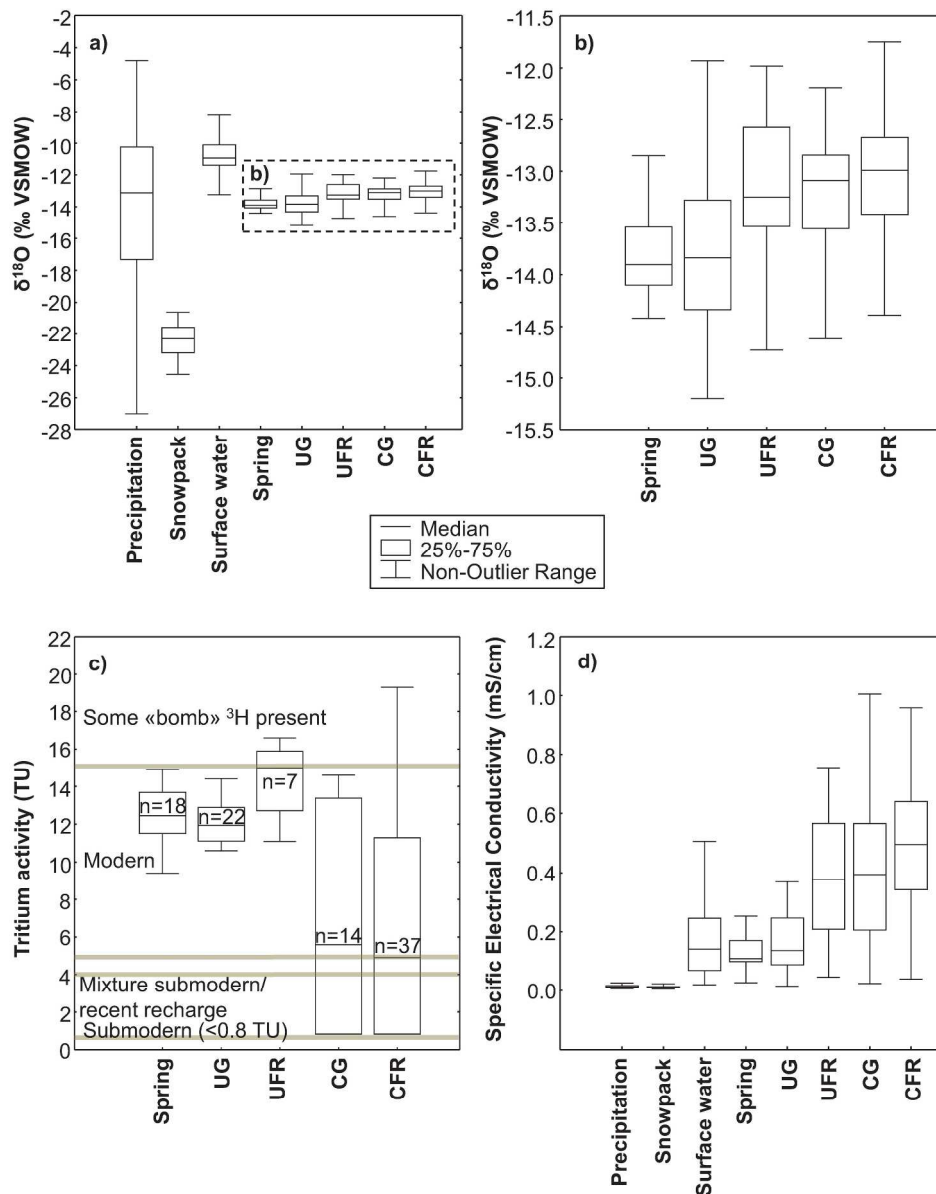


Figure 4. $\delta^{18}\text{O}$ (a; b), ^3H (c) and SEC (d) boxplots for precipitation, surface waters, springs and groundwater. Outlier data are excluded; the non-outlier range is defined as maximum and minimum values within 1.5 times the interquartile range. The available $\delta^{18}\text{O}$ data are shown in (a) whereas a clearer view of surface and groundwater data is provided in (b). The classification reported in (c) is adapted from Clark and Fritz (1997). UG: unconfined granular aquifers; UFR: unconfined fractured rock aquifers; CG: confined granular aquifers; CFR: confined fractured rock aquifers; SEC: specific electrical conductivity. In (a) and (d), the samples collected from the snowpack and at the precipitation monitoring stations are shown separately.

239x301mm (300 x 300 DPI)

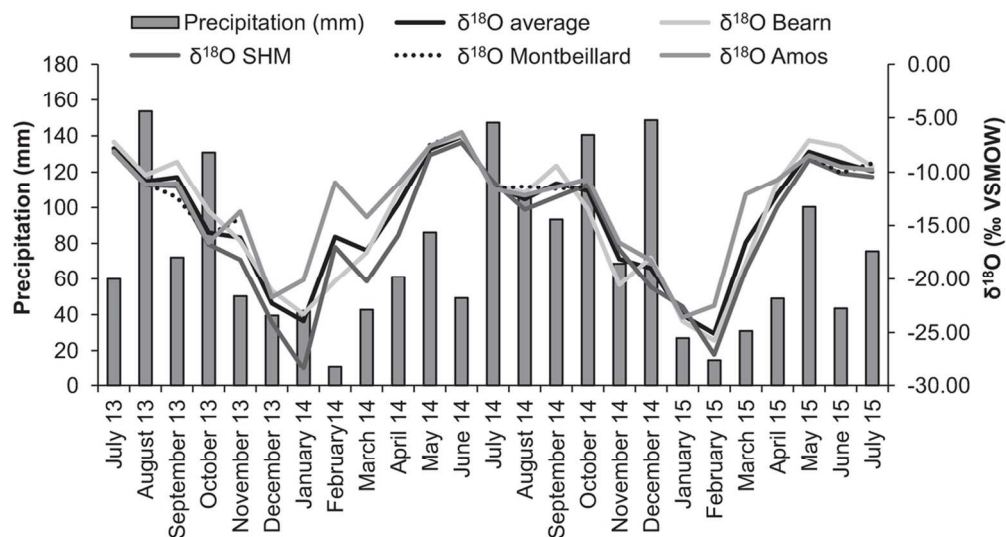


Figure 5. Seasonal variations in precipitation amounts and $\delta^{18}\text{O}$ composition. See figure 1 for the location of monitoring stations. The histogram showing precipitation amounts is based on data from July 2013 to July 2015 as recorded at the monitoring stations. SHM: Sainte-Hélène-de-Mancebourg.

104x57mm (300 x 300 DPI)

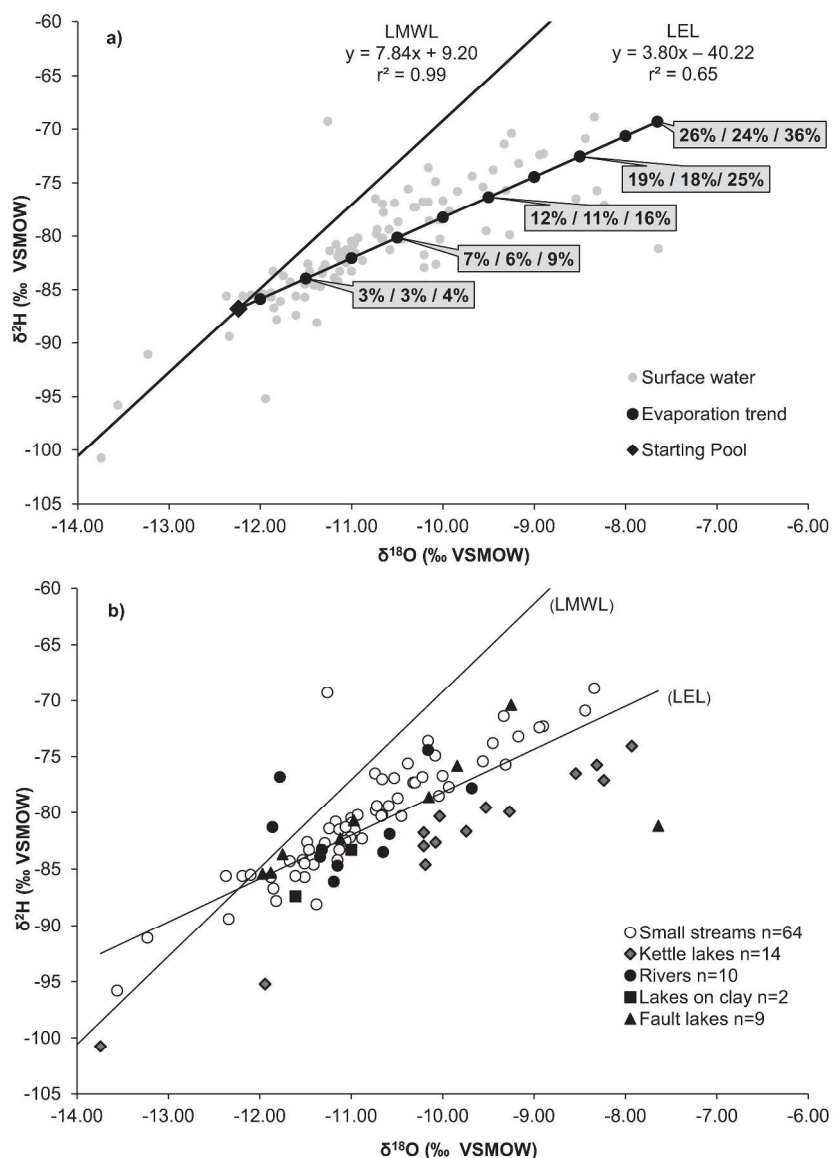


Figure 6. Interpretation of the isotopic compositions of surface waters. Data in (a) represent the local evaporation line evaluated using the 99 available surface water samples. The E/I ratios are shown for points plotting on the LEL at 0.5‰ intervals along the x-axis. These E/I ratios are shown for three scenarios (see text for details). Further details with respect to the types of surface water samples are provided in (b) and in table 3.

275x397mm (300 x 300 DPI)

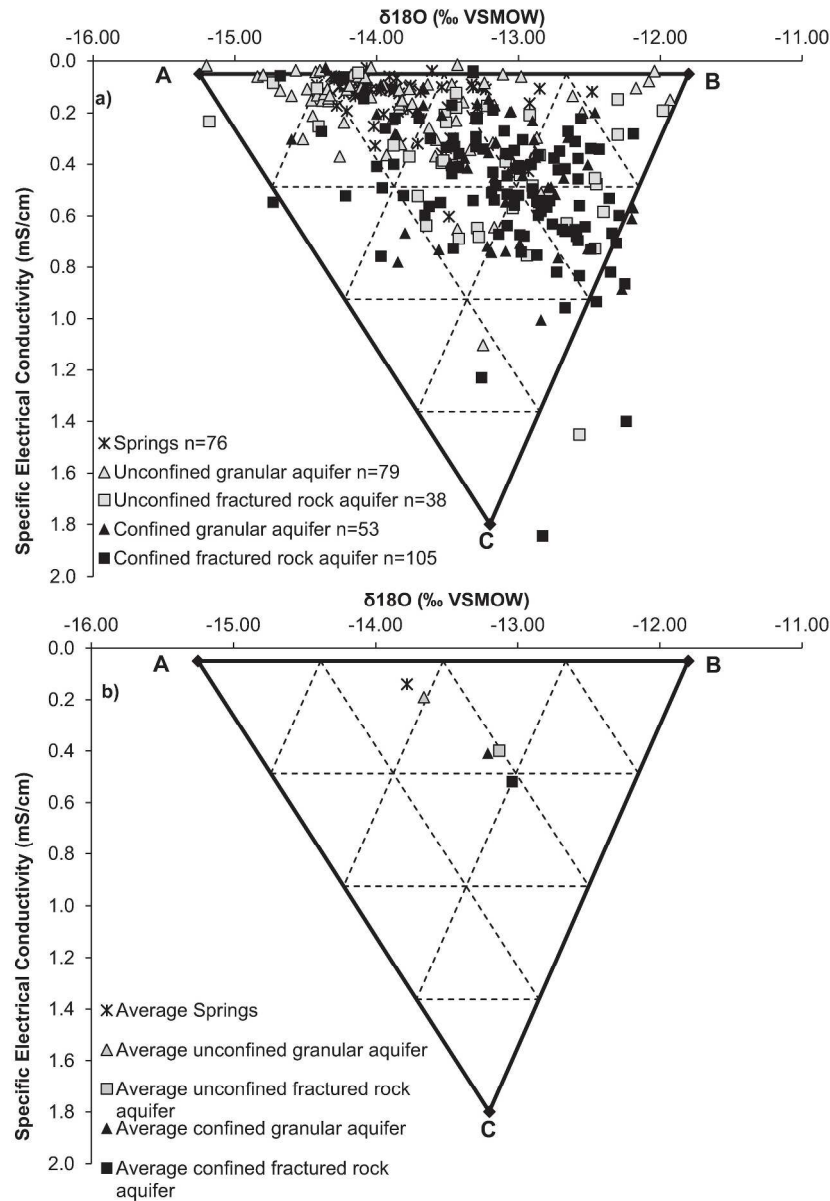


Figure 7. Specific electrical conductivity (SEC) vs $\delta^{18}\text{O}$ of springs, surface waters and groundwater. Three end-members are shown: «A» corresponds to groundwater recharged during snowmelt, «B» corresponds to groundwater that is less impacted by snowmelt induced recharge and «C» corresponds to geochemically evolved groundwater found further along regional flowpaths. See text for further details. The six main outliers (as described in the text) are not shown in this figure. Figure (a) shows the distribution of data whereas the corresponding averages are shown in (b).

263x365mm (300 x 300 DPI)

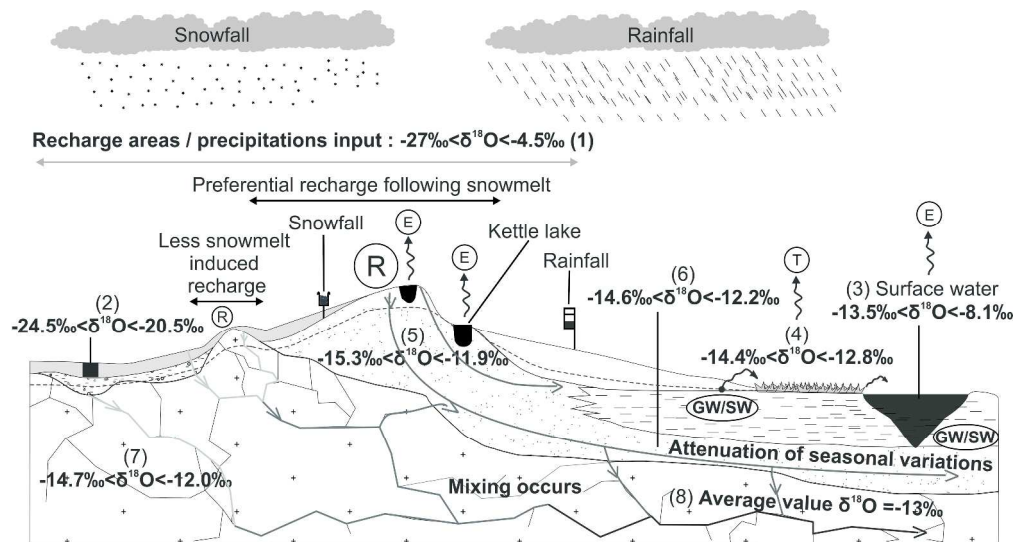


Figure 8. Isotopic variations within local, intermediate and regional flow systems. The range of measured isotopic compositions is shown for the sampled hydrological components and hydrogeological units, as identified in tables 1 and 2. Areas of preferential snowmelt induced recharge are identified. Flowpaths at various scales are shown. (R) Recharge area; (E) Evaporation; (T) Transpiration; (GW/SW) Groundwater / surface water interactions (herein mainly groundwater exfiltration). The numbers (1 to 8) correspond to the sampled hydrological components and hydrogeological units, as identified in table 1.

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